

(5 α H)-6-Epidihydrocornin, the First Known Iridoid Glycoside With a Trans-fused Ring System

Tommaso A. Foderaro and Frank R. Stermitz*

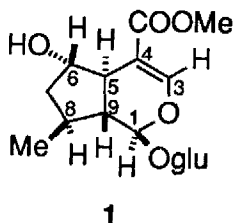
Department of Chemistry, Colorado State University, Fort Collins, Colorado 80523 USA

Håkon Hope

Department of Chemistry, University of California, Davis, California 95616 USA

Abstract: An iridoid glycoside isolated from *Penstemon secundiflorus* contains a trans-fused cyclopentanopyran ring system, unlike all other known iridoid glycosides, which contain a cis-fused cyclopentanopyran.

Iridoid glycosides are common constituents of a select group of higher plant families and they currently encompass over 500 individual structures.¹⁻⁴ All of these contain a cis-fused bicyclic cyclopentanopyran ring system. We have now isolated, from fresh leaves of *Penstemon secundiflorus* Benth., the first trans-fused iridoid glycoside. It was assigned structure **1**, (5 α H)-6-epidihydrocornin, from NMR analysis and a single crystal X-ray study of a tri-*p*-bromobenzoate derivative. *P. secundiflorus* from southern Colorado yielded an array of iridoid glycosides, including penstemoside, cornin, hastatoside, 8-epihastatoside, β -dihydrohastatoside, and adoxoside. The major iridoid isolate (0.3% of the fresh leaf weight) was **1**.



The ¹H and ¹³C NMR spectra (including DEPT and HETCOR) for **1** were typical for iridoid glycosides related to **1** except for three pieces of data which were outside the range of those reported for all other iridoid glycosides: 1) the resonance of C-1 was at 103 ppm ("normal" 96 to 100 ppm), 2) the resonance for H-9 was at 1.5 ppm ("normal" 2.0 to 2.7 ppm), and 3) the allylic coupling of the H-3/H-5 resonances was 2.1 Hz ("normal" 0.5-0.9 Hz).

Strong nOe enhancements for noncontiguous protons were seen between H-6 and H-9, H-8 and H-5, and H-8 and H-1, which showed that H-9 and H-6 were on the same side of the ring system, while H-8, H-5 and H-1 were together on the opposite side, as shown in **1**. Dreiding models indicated that the trans ring fusion caused significant conformational changes in the pyran ring which could account for the anomalies in the NMR spectra. The Oglucose substituent at C-1 takes up a pseudo-equatorial position, while it is pseudo-axial in cis-fused iridoids; the C-3/C-4 π -bond and the H-9 proton are tilted toward each other, which places the H-9 proton in a shielding region; and the H-5 C-H bond is parallel (rather than perpendicular) to the p-orbitals of the C-3/C-4 π -bond, which results in a maximum (rather than minimum) coupling constant.⁵

Because of the novelty of this suggested structure, the isolate was treated with *p*-bromobenzoyl chloride and the

derivative mixture was purified by preparative TLC to yield crystals (mp 182° dec. from hexane/EtOAc) of the 2', 3', 6'-tri-*p*-bromobenzoate derivative. This was shown by a single crystal X-ray study⁶ to have the structure given in Figure 1, which confirmed in all respects the assigned stereochemistry for **1**. A trans-fused structure was once postulated for an isomer of deoxyloganic acid from *Nepata cataria*, but was subsequently reassigned⁷ a cis-fused structure, with absolute stereochemistry at C-5 and C-9 opposite to that of other iridoid glycosides.

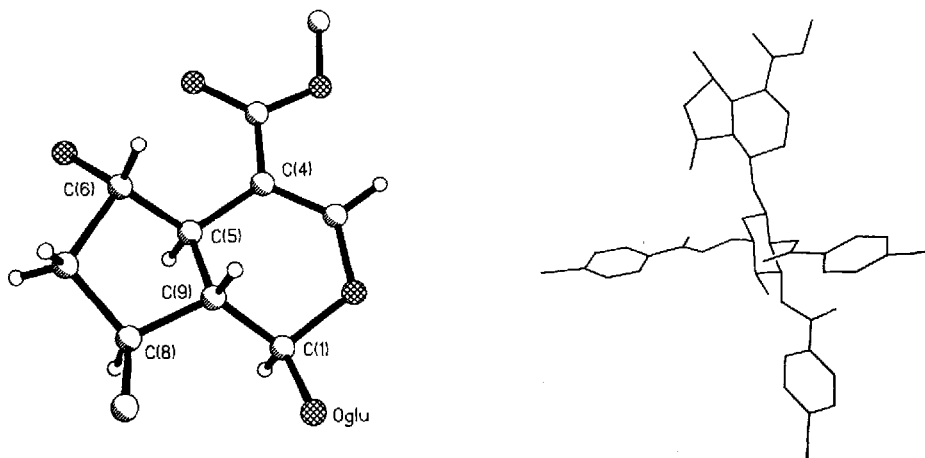


Fig. 1. Iridoid portion (left) and complete molecule (right, line-bond structure) of the 2',3',6'-tri-*p*-bromobenzoate derivative of **1**, computer drawn from the single crystal X-ray data.

There are no known instances of epimerization at C-5 among the hundreds of iridoid isolations and it seems likely that the trans ring fusion for **1** is set at the time of cyclization from an acyclic monoterpene precursor.⁸

Acknowledgments. This work was supported by NSF grant CHE-9023608 to FRS. The diffraction and computing equipment was purchased under NSF grant CHE-8802721 to the University of California, Davis. *Penstemon secundiflorus* was identified by Prof. William A. Weber, University of Colorado, Boulder.

- (1) El-Naggar, L. J.; Beal, J. L. *J. Nat. Prod.* **1980**, *43*, 649.
- (2) Boros, C. A.; Stermitz, F. R. *J. Nat. Prod.* **1990**, *53*, 1055.
- (3) Boros, C. A.; Stermitz, F. R. *J. Nat. Prod.* **1991**, *54*, 1173.
- (4) Jensen, S. R., personal communication.
- (5) Günther, H. *NMR Spectroscopy*; John Wiley & Sons, New York, 1980.
- (6) X-ray data (made available to reviewers) will be published in *Acta Crystallographica*.
- (7) Murai, F.; Tagawa, M.; Damtoft, S.; Jensen, S. R.; Nielsen, B. J. *Chem. Pharm. Bull.* **1984**, *32*, 2809.
- (8) Inouye, H.; Uesato, S. *Progr. Chem. Org. Nat. Prod.* **1986**, *50*, 180.