

[Chem. Pharm. Bull.]  
36(8)3043-3048(1988)

**Metabolism of Paeoniflorin and Related Compounds by Human Intestinal Bacteria. IV. Formation and Structures of Adducts of a Metabolic Intermediate with Sulfhydryl Compounds by *Lactobacillus brevis***

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(Received January 25, 1988)

Paeoniflorin from peony roots was converted to new compounds in the presence of various sulfhydryl compounds by *Lactobacillus brevis*. The compounds formed in the presence of 3-mercaptopropionic acid, 2-mercaptoethanol and thiobenzoic acid were determined to be 7*R* and 7*S* mixture of 8-(2-carboxyethylthio)paeonimetabolin I, 8-(2-hydroxyethylthio)paeonimetabolin I and 8-benzoylthiopaeonimetabolin I, respectively, by various spectroscopic methods. These compounds are adducts of a metabolic intermediate of paeoniflorin with the sulfhydryl compounds.

**Keywords**—8-(2-hydroxyethylthio)paeonimetabolin I; 8-(2-carboxyethylthio)paeonimetabolin I; 8-benzoylthiopaeonimetabolin I; human intestinal bacteria; metabolism; paeoniflorin; *Lactobacillus brevis*

During the course of our studies on the metabolism of paeoniflorin (**1**) and related compounds from *Paeoniae Radix* by human intestinal bacteria, we found that **1**, as well as oxypaeoniflorin and benzoylpaeoniflorin, was converted to various metabolites by human intestinal flora.<sup>1)</sup> The structure of the major metabolite was concluded to be 7*S*-paeonimetabolin I (**2**),<sup>2)</sup> and possible metabolic processes from **1** to **2** were proposed. Many defined strains of human intestinal bacteria such as *Lactobacillus brevis* and *Bacteroides fragilis* were able to metabolize **1** to a mixture of 7*S*- and 7*R*-paeonimetabolins I (**2** and **3**, respectively),<sup>3,4)</sup> and the structure of the latter was confirmed by X-ray analysis.<sup>3)</sup>

In the present paper, we report the formation of adducts of a metabolic intermediate and sulfhydryl compounds during incubation of **1** with *L. brevis* in the presence of sulfhydryl compounds, and the characterization of these adducts by various spectroscopic methods.

#### Materials and Methods

**Instruments**—Infrared (IR) spectra were measured with a Hitachi 260-10 infrared spectrophotometer. Proton and carbon-13 nuclear magnetic resonance (<sup>1</sup>H- and <sup>13</sup>C-NMR) spectra were measured with JEOL FX-270 (<sup>1</sup>H, 270 MHz) and JEOL-FX 90Q (<sup>13</sup>C, 22.5 MHz) NMR spectrometers. In <sup>13</sup>C-NMR spectra, the multiplicities were determined on the basis of off-resonance decoupling (OFR) and insensitive nuclei enhanced by polarization transfer (INEPT) techniques. Tetramethylsilane was used as an internal standard in all the measurements. Mass spectra (MS) were measured with a JEOL D-200 mass spectrometer at an ionization voltage of 70 eV. High-performance liquid chromatography (HPLC) was carried out on Tri-Rotar SR-1 equipped with a UVIDEC-100-IV detector (JASCO).

**Materials**—Paeoniflorin was isolated from peony roots according to the method of Kaneda *et al.*<sup>5)</sup> GAM broth was a product of Nissui Seiyaku Co., Ltd. (Tokyo). All chemicals used were of analytical reagent grade.

**Chromatography of Metabolites**—Wakogel C-200 was used for column chromatography. Thin layer chromatography (TLC) was performed on Merck Kieselgel 60 F or Merck PSC-60 F (for preparative purposes) plates with a

solvent system of  $\text{CHCl}_3$ -MeOH-benzene (5:2:1). Spots on the plates were visualized by exposure to iodine vapor or by spraying with an anisaldehyde- $\text{H}_2\text{SO}_4$  reagent, followed by heating.

**Metabolism of Paeoniflorin by *Lactobacillus brevis* in the Presence of Sulfhydryl Compounds**—A bacterial suspension of *L. brevis* precultured overnight in an anaerobic box was added to 9 vol. of GAM broth and cultivated for 5 h at 37 °C in the anaerobic box. The cells obtained by centrifugation at 7000 rpm for 10 min were washed once with a saline solution and suspended in 1/40 of the culture volume of 50 mM potassium phosphate buffer (pH 7.2). The reaction mixture contained 0.2 ml of the bacterial suspension, 40  $\mu\text{l}$  of 17 mM paeoniflorin solution and 0.2 ml of 50 mM potassium phosphate buffer (pH 7.2). A sulfhydryl compound was added to the reaction mixture at a final concentration of 5 mM. After incubation for 1 h at 37 °C, the reaction was stopped by adding 0.8 ml of butanol and then extracted. An aliquot of the butanol layer was chromatographed on a silica gel TLC plate with the solvent system described above.

**Isolation of 7R- and 7S-8-(2-Carboxyethylthio)paeonimetalolins I**—A precultured bacterial suspension (600 ml) of *L. brevis* was added to GAM broth (6 l) and cultivated for 12 h at 37 °C. The culture was centrifuged at 7000 rpm for 10 min. The precipitates were washed with a saline solution, centrifuged and suspended in 50 mM phosphate buffer (900 ml). The suspension was transferred into six tubes. 3-Mercaptopropionic acid (500 mg) and then paeoniflorin (1, 1.2 g) were added to each tube. The mixture was anaerobically incubated for 4 h at 37 °C and adjusted to pH ca. 4 with dilute HCl, then extracted three times with ethyl acetate (AcOEt, 200 ml each). The organic layer was washed with a saturated NaCl solution, and concentrated *in vacuo* to give an oily residue. The combined residues (0.6 g) were applied to a column of silica gel (60 g, 24  $\times$  240 mm). The column was thoroughly washed with  $\text{CHCl}_3$  and eluted with  $\text{CHCl}_3$ -MeOH (100:2). Fractions (30 ml each) were collected and monitored by TLC and  $^1\text{H}$ -NMR. Fractions 1–3 afforded a colorless oil (4, 20 mg) and fractions 14–20 an oil (5, 15 mg). Fractions 4–13 gave a mixture of 4 and 5 (48 mg).

**7S-8-(2-Carboxyethylthio)paeonimetalolin I (4)**—Colorless oil. High-resolution MS: Found, 302.0859, Calcd for  $\text{M}^+$ ,  $\text{C}_{13}\text{H}_{18}\text{O}_6\text{S}$ , 302.0824. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400 (br, OH), 1720 (br, C=O). MS  $m/z$  (rel. int): 302 ( $\text{M}^+$ , 2), 284 (2), 266 (6), 248 (2), 197 (20), 178 (56), 161 (50), 151 (76), 121 (30), 106 (78), 88 (63), 69 (100).

**7R-8-(2-Carboxyethylthio)paeonimetalolin I (5)**—Colorless oil. High-resolution MS: Found, 302.0838, Calcd for  $\text{M}^+$ ,  $\text{C}_{13}\text{H}_{18}\text{O}_6\text{S}$ , 302.0824. MS  $m/z$  (rel. int): 302 ( $\text{M}^+$ , 2), 284 (4), 266 (10), 197 (15), 178 (100), 161 (74), 150 (83), 121 (54), 106 (83), 88 (64), 77 (64), 69 (88).

**Isolation of 7R- and 7S-8-(2-Hydroxyethylthio)paeonimetalolins I**—Paeoniflorin (1, 1.2 g) was incubated with a suspension of *L. brevis* (900 ml) in the presence of 2-mercaptoethanol (370 mg) under conditions similar to those described above. The products were extracted with AcOEt without acidification. The organic layer was evaporated *in vacuo* to give an oily residue (0.5 g). The residue was chromatographed on a silica gel column (60 g, 24  $\times$  240 mm). The column was washed with  $\text{CHCl}_3$  and eluted with  $\text{CHCl}_3$ -MeOH (100:0.7). Fractions (40 ml/flask) were collected and monitored by TLC and  $^1\text{H}$ -NMR. Fractions 24–27, 28–39 and 40–44 afforded 6 (colorless oil, 19 mg), a mixture of 6 and 7 (oil, 38 mg) and 7 (oil, 13 mg), respectively.

**7S-8-(2-Hydroxyethylthio)paeonimetalolin I (6)**—Colorless oil. High-resolution MS: Found, 274.0859; Calcd for  $\text{M}^+$ ,  $\text{C}_{12}\text{H}_{18}\text{O}_5\text{S}$ , 274.0874. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3420 (OH), 1720 (C=O). MS  $m/z$  (rel. int): 274 ( $\text{M}^+$ , 3), 256 (12), 197 (15), 151 (83), 123 (10), 69 (100).

**7R-8-(2-Hydroxyethylthio)paeonimetalolin I (7)**—Colorless oil. High-resolution MS: Found, 274.0857, Calcd for  $\text{M}^+$ ,  $\text{C}_{12}\text{H}_{18}\text{O}_5\text{S}$ , 274.0874. MS  $m/z$  (rel. int): 274 ( $\text{M}^+$ , 4), 256 (15), 197 (28), 151 (89), 123 (9), 69 (100).

**Isolation of 7R- and 7S-8-Benzoylthiopaeonimetalolins I**—Paeoniflorin (1, 2.05 g) was incubated with a suspension of *L. brevis* (900 ml) in the presence of thiobenzoic acid (1.7 g) under conditions similar to those described above. The AcOEt extract (1.35 g) was subjected to column chromatography on silica gel (70 g, 24  $\times$  260 mm). Elution with benzene- $\text{CHCl}_3$  (3:1) afforded a mixture (140 mg) of compounds 8 and 9 in a ratio of 6:4. A portion of the mixture (50 mg) was separated into 8 (14 mg) and 9 (6 mg) by preparative HPLC under the following conditions: column, Chemcosorb 5-S1, (10  $\mu$ , 10  $\times$  500 mm, Chemco Ltd., Osaka); mobile phase,  $\text{CH}_2\text{Cl}_2$ - $\text{CH}_3\text{CN}$  (100:4); flow rate, 3.5 ml/min; pressure, 40 kg/cm<sup>2</sup>; detection, ultraviolet (UV) at 235 nm.

**7S-8-Benzoylthiopaeonimetalolin I (8)**—Colorless oil. High-resolution MS: Found, 334.0843, Calcd for  $\text{M}^+$ ,  $\text{C}_{17}\text{H}_{18}\text{O}_5\text{S}$ , 334.0874. UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 238 (4.28), 266 (4.17). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3420 (br, OH), 1720 (ketonic C=O), 1660 (thioester C=O). MS  $m/z$  (rel. int): 334 ( $\text{M}^+$ , 2), 316 (2), 298 (4), 197 (15), 196 (25), 151 (44), 105 (90), 77 (100), 69 (51).

**7R-8-Benzoylthiopaeonimetalolin I (9)**—Colorless oil. High-resolution MS: Found, 334.0845, Calcd for  $\text{M}^+$ ,  $\text{C}_{17}\text{H}_{18}\text{O}_5\text{S}$ , 334.0874. MS  $m/z$  (rel. int): 334 ( $\text{M}^+$ , 1), 316 (1), 298 (5), 197 (16), 196 (27), 151 (46), 105 (100), 77 (95), 69 (49).

## Results

### New Products Formed from Paeoniflorin and Dithiothreitol by *Lactobacillus brevis*

*L. brevis* suspended in phosphate buffer (pH 7.2) transformed 1 to paeonimetalolin I (2

and 3) under either aerobic or anaerobic conditions. In an attempt to isolate the metabolic intermediates, various reagents which inhibit enzyme activity were added to the bacterial suspension. The transforming activity was not affected by pretreatment with sulfhydryl reagents such as 5,5'-dithiobis-(2-nitrobenzoic acid), *p*-chloromercuriphenylsulfonic acid and *N*-ethylmaleimide. However, in the presence of dithiothreitol, *L. brevis* transformed 1 not to paeonimetabolin I (2 and 3) but to new products, which showed one spot on TLC (Fig. 1). These products were found to be an isomeric mixture by  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR. They were not produced without 1 or *L. brevis* even in the presence of dithiothreitol. In addition, no such products were obtained when dithiothreitol was added after transformation of 1 to paeonimetabolin I.

### Effects of Various Sulfhydryl Compounds on Paeoniflorin Metabolism

The effects of other sulfhydryl compounds on the metabolism of 1 by *L. brevis* were examined. Addition of 1-thioglycerol, 2-mercaptoethanol, 3-mercaptopropionic acid, 2-mercaptoethylamine, thiobenzoic acid and L-cysteine also resulted in the formation of new products with a corresponding decrease of 1, little or no paeonimetabolin I (2 and 3) being produced (Fig. 2). The *R<sub>f</sub>* values of the new products on TLC varied depending on the sulfhydryl compounds used. Moreover, the products formed in the presence of 3-mercaptopropionic acid and 2-mercaptoethylamine had higher *R<sub>f</sub>* values with acidic and alkaline solvent systems, respectively, on TLC, but lower values in the opposite systems, showing acidic and basic properties, respectively.

Addition of glutathione (reduced form) also decreased the amounts of paeonimetabolin I (2 and 3) formed, but the new products were not extracted with butanol and remained in the aqueous layer. Similarly, the products formed in the presence of 3-mercaptopropionic acid, 2-mercaptoethylamine and L-cysteine were water-soluble.

On the other hand, the addition of glutathione (oxidized form), L-cysteic acid or L-

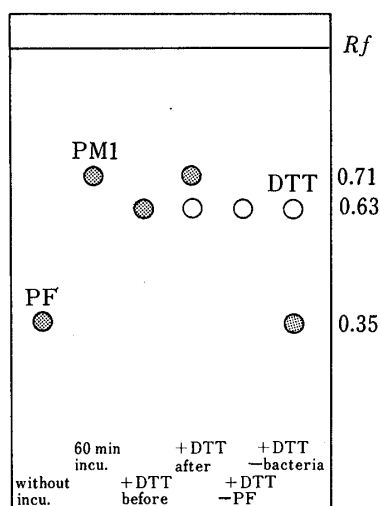


Fig. 1. Adduct Formation with Dithiothreitol

*L. brevis* was incubated with paeoniflorin (PF) for 1 h at 37°C as described in Materials and Methods. Dithiothreitol (DTT) at a final concentration of 5 mM was added to the reaction mixture before (+DTT, before) or after incubation (+DTT, after). In the latter case, the mixture was incubated for a further 30 min. Control experiments were carried out in the absence of PF (+DTT, -PF) and bacteria (+DTT, -bacteria). Open and shaded circles indicate yellowish spots of DTT and pink-colored spots, respectively. PM1, paeonimetabolin I.

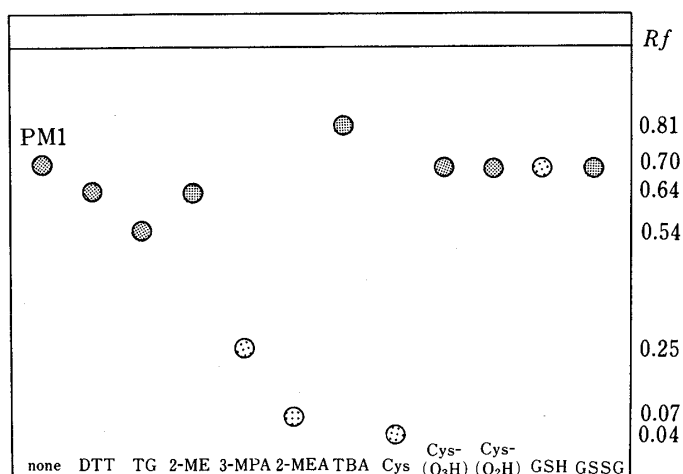


Fig. 2. Effect of Various Sulfhydryl Compounds on the Metabolism of Paeoniflorin

*L. brevis* was incubated with paeoniflorin in the presence of 5 mM DTT, 1-thioglycerol (TG), 2-mercaptoethanol (2-ME), 3-mercaptopropionic acid (3-MPA), 2-mercaptoethylamine (2-MEA), thiobenzoic acid (TBA), L-cysteine (Cys), L-cysteic acid (Cys-(O<sub>3</sub>H)), L-cysteinesulfinic acid (Cys-(O<sub>2</sub>H)) or glutathione, reduced form (GSH) or oxidized form (GSSG) for 1 h at 37°C. Shaded and dotted circles indicate intensely and feebly pink-colored spots, respectively.

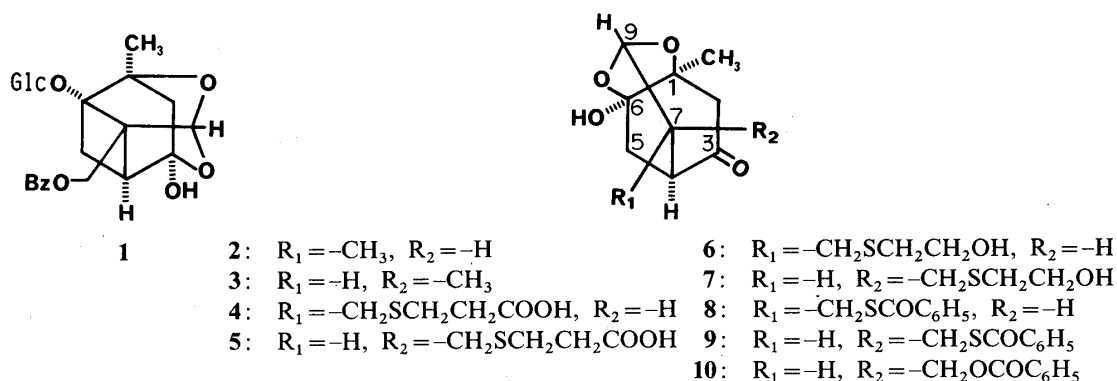


Chart 1. Structures of Paeoniflorin and Paeonimetabolin I Derivatives Listed in the Paper

cysteinesulfinic acid did not affect the amounts of paeonimetabolin I (**2** and **3**) formed, and no new products were formed. Moreover, when new products obtained in the presence of dithiothreitol (having two sulfhydryl groups) were treated with *p*-chloromercuribenzoic acid, the products were converted to other compounds. However, the products obtained in the presence of 1-thioglycerol (having one sulfhydryl group) were not converted by treatment with *p*-chloromercuribenzoic acid.

These results suggest that an intermediate in the metabolism of **1** forms adducts with various sulfhydryl compounds, and the sulfhydryl group takes part in the adduct formation.

### Structures of Adducts

Following incubation of **1** with a suspension of *L. brevis* in the presence of 3-mercaptopropionic acid, 2-mercaptoethanol and thiobenzoic acid, major products were extracted and separated into two isomers, **4** and **5**, **6** and **7**, and **8** and **9**, respectively, by repeated column chromatography. The structures of these isomers were characterized as follows.

**(HOOC-CH<sub>2</sub>-CH<sub>2</sub>-SH)·Adducts (**4** and **5**)**—Products **4** and **5** were obtained as colorless oils. The molecular formulae were determined to be C<sub>13</sub>H<sub>18</sub>O<sub>6</sub>S by high-resolution MS. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra (Tables I and II) of **4** and **5** were quite similar to those of **2** and **3**, respectively, except for the signals due to the presence of a HOOC-CH<sub>2</sub>-CH<sub>2</sub>-S-CH<sub>2</sub>- group instead of a methyl signal (8-H) in the latter two, thus indicating that **4** and **5** are an isomeric mixture of 8-(2-carboxyethylthio)paeonimetabolin I. The configuration around at C-7 was deduced on the basis of the presence or absence of  $\gamma$ -gauche steric effect<sup>6)</sup> at the C-5 and C-8 signals in the <sup>13</sup>C-NMR spectra, as observed in **2**. The signal of C-5 in **4** appeared at  $\delta$  31.3, 4.2 ppm units higher than that in **5** ( $\delta$  35.5), indicating the  $\gamma$ -gauche effect in **4**. These findings led us to conclude the structures of **4** and **5** to be (7*S*)- and (7*R*)-8-(2-carboxyethylthio)paeonimetabolin I (Chart 1), respectively.

**(HO-CH<sub>2</sub>-CH<sub>2</sub>-SH)·Adducts (**6** and **7**)**—Both **6** and **7** were obtained as colorless oils with the same molecular formula, C<sub>12</sub>H<sub>18</sub>O<sub>5</sub>S. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra (Tables I and II) were indicative of paeonimetabolin I derivatives in which a HO-CH<sub>2</sub>-CH<sub>2</sub>-S-group is attached to the C-8 position. On the basis of the presence or absence of  $\gamma$ -gauche steric effect between C-5 ( $\delta$  31.5 and 35.6 in **6** and **7**, respectively) and C-8 the structures of **6** and **7** were concluded to be (7*S*)- and (7*R*)-8-(2-hydroxyethylthio)paeonimetabolin I (Chart 1), respectively.

**(C<sub>6</sub>H<sub>5</sub>-CO-SH)·Adducts (**8** and **9**)**—Similarly, oily products with the molecular formula, C<sub>17</sub>H<sub>18</sub>O<sub>5</sub>S, were concluded to be (7*S*)- and (7*R*)-8-benzoylthiopaeonimetabolin I (Chart 1), respectively, by <sup>1</sup>H- and <sup>13</sup>C-NMR analyses (Tables I and II). The (7*R*)-isomer showed spectroscopic properties quite similar to those of naturally occurring paeoniflorin.

TABLE I.  $^1\text{H}$ -NMR (270 MHz) Spectral Data of Adducts (ppm in Pyridine- $d_5$ )<sup>a)</sup>

	4 7S	5 7R	6 7S	7 7R	8 7S <sup>b)</sup>	9 7R <sup>b)</sup>
2-H <sub>2</sub>	2.97, 3.03, ABq (16.8)	3.00, 3.07, ABq (16.8)	2.90, 2.96, ABq (17.1)	2.89, 2.96, ABq (16.9)	2.67, 2.74, ABq (17.2)	2.68, s
4-H	2.97, m	3.00, m	2.92, m	2.92, m	2.75, m	2.97, m
5-H <sub>2</sub>	2.32, dd; 2.58, dd (13.7, 1.8; 13.7, 3.0)	2.42, dd; 2.53, dd (13.7, 2.4; 13.7, 3.0)	2.27, dd; 2.54, dd (13.7, 2.2; 13.7, 3.4)	2.34, dd; 2.48, dd (13.4, 2.4; 13.4, 3.2)	2.11, dd; 2.56, dd (13.9, 2.3; 13.9, 3.2)	2.20, dd; 2.55, dd (13.6, 2.7; 13.6, 3.4)
7-H	2.08, t (7.8)	2.24, m	2.02, t (7.8)	2.20, m	1.96, t (7.7)	2.22, td (7.3, 1.0)
8-H <sub>2</sub>	2.71, dd; 2.93, dd (14.0, 7.8; 14.0, 7.8)	2.60—2.90, m	2.67, dd; 2.92, dd (13.4, 7.8; 13.4, 7.8)	2.68, dd; 2.91, dd (13.3, 8.3; 13.3, 7.6)	3.22, d (7.7)	2.95, d (7.3)
9-H	5.69, br s	5.74, br s	5.62, br s	5.69, br s	5.39, br s	5.41, br s
10-H <sub>3</sub>	1.48, s —CH <sub>2</sub> COOH, 2.77, t (6.6) —CH <sub>2</sub> S—, 2.92, t (6.6)	1.51, s —CH <sub>2</sub> COOH, 2.81, t (6.6) —CH <sub>2</sub> S—, 2.95, t (6.6)	1.45, s —CH <sub>2</sub> S—, 2.78, td (6.6, 1.9) —CH <sub>2</sub> OH, 3.90, t (6.6)	1.45, s —CH <sub>2</sub> S—, 2.78, td (6.6, 1.9) —CH <sub>2</sub> OH, 3.90, t (6.6)	3',5'-H, 7.45, br t (7.6) 4'-H, 7.59, br t (7.6) 2',6'-H, 7.96, br d (7.6)	3',5'-H, 7.45, br t (7.6) 4'-H, 7.59, br t (7.6) 2',6'-H, 7.96, br d (7.6)

a) Coupling constants (Hz) are shown in parentheses. b) Measured in CDCl<sub>3</sub>.TABLE II.  $^{13}\text{C}$ -NMR (22.5 MHz) Spectral Data of Adducts and Paeonimetabolin I (ppm in Pyridine- $d_5$ )

Carbon No.	4 7S	5 7R	6 7S	7 7R	8 7S <sup>a)</sup>	9 7R <sup>a)</sup>	2 7S <sup>a)</sup>	3 7R <sup>a)</sup>
C-1	78.3 (s)	79.1 (s)	78.5 (s)	79.0 (s)	78.0 (s)	78.6 (s)	77.4 (s)	78.3 (s)
C-2	47.9 (t)	47.6 (t)	48.0 (t)	47.4 (t)	47.4 (t)	46.8 (t)	47.5 (t)	46.8 (t)
C-3	210.9 (s)	210.4 (s)	211.1 (s)	210.4 (s)	210.5 (s)	210.2 (s)	210.9 (s)	210.1 (s)
C-4	47.8 (d)	48.8 (d)	48.1 (d)	48.7 (d)	47.4 (d)	47.8 (d)	50.2 (d)	49.6 (d)
C-5	31.3 (t)	35.5 (t)	31.7 (t)	35.6 (t)	30.8 (t)	34.2 (t)	30.4 (t)	34.3 (t)
C-6	102.2 (s)	102.2 (s)	102.3 (s)	102.0 (s)	101.7 (s)	101.5 (s)	101.6 (s)	101.3 (s)
C-7	43.1 (d)	44.3 (d)	43.6 (d)	44.4 (d)	43.3 (d)	44.1 (d)	37.9 (d)	38.4 (d)
C-8	31.3 (t)	31.3 (t)	31.7 (t)	31.3 (t)	27.7 (t)	27.2 (t)	14.6 (q)	13.3 (q)
C-9	101.2 (d)	101.0 (d)	101.4 (d)	100.0 (d)	101.1 (d)	100.8 (d)	103.1 (d)	103.0 (d)
C-10	21.7 (q)	22.0 (q)	21.7 (q)	21.7 (q)	20.9 (q)	20.9 (q)	20.9 (q)	21.1 (q)
—CH <sub>2</sub> S—	35.1 (t)	35.5 (t)	35.2 (t)	35.4 (t)				
—CH <sub>2</sub> —	27.3 (t)	28.1 (t)	61.6 (t)	61.6 (t)				
—COOH	173.9 (s)	174.2 (s)						
C <sub>6</sub> H <sub>5</sub> COS—					127.1 (d)	127.1 (d)		
					128.5 (d)	128.1 (d)		
					133.5 (d)	133.5 (d)		
					136.3 (s)	136.4 (s)		
					190.9 (s)	191.3 (s)		

Abbreviations given in parentheses indicate the signal patterns based on the OFR and INEPT methods. a) Measured in CDCl<sub>3</sub>.genone (10),<sup>7)</sup> which could be regarded as a benzoic acid/paeonimetabolin-I adduct.

### Discussion

Paeoniflorin (1) was transformed to new products by *L. brevis* in the presence of various

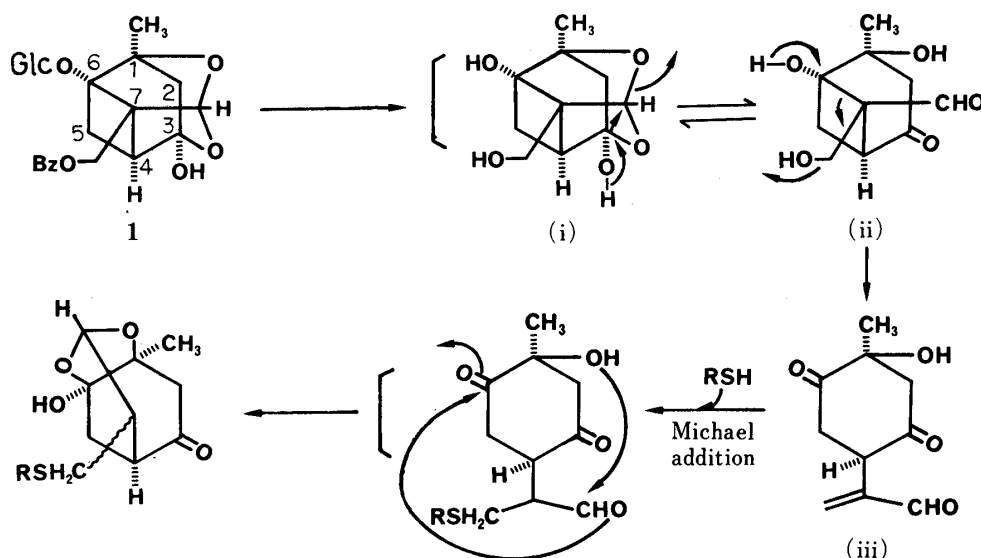


Chart 2. Processes for Formation of Adducts in the Presence of Sulfhydryl Compounds

sulfhydryl compounds. These products possessed similar structures in which the sulfhydryl compounds are covalently bound at the C-8 position of paeonimetabolin I (2 and 3) through a thioether linkage. They could not be formed by incubation of paeonimetabolin I (2 and 3) with sulfhydryl compounds, nor in the absence of *L. brevis*. The mechanism of formation of the new products seems to be that an  $\alpha$ ,  $\beta$ -unsaturated aldehyde intermediate (iii in Chart 2) in the metabolic processes of 1 to paeonimetabolin I (2 and 3), as we proposed in the preceding papers,<sup>1,3)</sup> reacts in preference with a mercapto group of the compounds by Michael addition, followed by the formation of a hemiketal-acetal system. Thus, the adducts obtained contained the *S*- and *R*-isomers, which were formed in almost equal amounts in the presence of 3-mercaptopropionic acid, 2-mercaptoethanol and thiobenzoic acid. Among nucleophilic functional groups of the compounds, only mercapto and mercaptocarbonyl groups took part in the reaction; hydroxy, carboxy and amino groups were less reactive.

On the other hand, compound 9 could be regarded as an analogue of paeoniflorigenone (10)<sup>7)</sup> in which the benzoyl moiety is replaced by a thiobenzoyl group. The formation of this adduct from the metabolic intermediate (iii) in the presence of thiobenzoic acid suggests a similar pathway for biosynthesis of paeoniflorigenone (10) in peonies.

**Acknowledgement** This study was funded in part by Tsumura Juntendo Co., Ltd. (Tokyo).

#### References and Notes

- 1) M. Hattori, Y. Z. Shu, M. Shimizu, T. Hayashi, N. Morita, K. Kobashi, G. J. Xu and T. Namba, *Chem. Pharm. Bull.*, **33**, 3838 (1985).
- 2) A tentative name "paeonimetaboline I" used in the previous papers<sup>1,3,4)</sup> was revised to "paeonimetabolin I" to avoid confusion, as it does not contain a nitrogen atom in the molecule.
- 3) Y. Z. Shu, M. Hattori, T. Akao, K. Kobashi, K. Kagei, K. Fukuyama, T. Tsukihara and T. Namba, *Chem. Pharm. Bull.*, **35**, 3726 (1987).
- 4) Y. Z. Shu, M. Hattori, T. Akao, K. Kobashi and T. Namba, *J. Med. Pharm. Wakan-Yaku*, **4**, 82 (1987).
- 5) M. Kaneda, Y. Iitaka and S. Shibata, *Tetrahedron*, **28**, 4309 (1972).
- 6) E. Breitmaier and W. Voelter, "<sup>13</sup>C-NMR Spectroscopy," 2nd Edition, Verlag Chemie, Weinheim-New York, 1978, pp. 74–75.
- 7) M. Shimizu, T. Hayashi, N. Morita, F. Kiuchi, H. Noguchi, Y. Iitaka and U. Sankawa, *Chem. Pharm. Bull.*, **31**, 577 (1983).