#### 2139

# Reactions of Vinylphosphonates. 3.1) One-Pot Synthesis of Dienes and Their Analogues from Vinylphosphonates, Aldehydes and Diethyl Phosphonate

Toru Minami,\* Shigehumi Tokumasu, and Ichiro Hirao Department of Industrial Chemistry, Kyushu Institute of Technology, Sensuicho, Tobata, Kitakyushu 804 (Received January 24, 1985)

Synonsis. Functionalized dienes and tetraenes were synthesized in 26—63% yields by the one-pot reaction of the phosphoryl-stabilized carbanions, prepared from the Michael addition of diethyl lithiophosphonate to vinylphosphonates, with aromatic aldehydes. A similar reaction using phthalaldehyde gave 2-methylsulfonylnaphthalene (7) in 31% yield via an intramolecular double Horner-Wittig reaction.

© 1985 The Chemical Society of Japan

We have previously reported a new synthesis of vinylphosphonates bearing electronegative substituents such as ethoxycarbonyl and cyano groups, and their use in syntheses of functionalized heterocyclic compounds, and olefins, dienes and their analogues.<sup>1,2)</sup> In connection with our continuing interest in the synthetic application of the vinylphosphonates, we have now developed an one-pot synthesis of unsaturated systems such as dienes and tetraenes, etc. having functionality.

#### **Results and Discussion**

The reaction of a phosphoryl-stabilized carbanion 2a, generated in-situ from a Michael addition of diethyl lithiophosphonate to diethyl [1-(methylsulfonyl)vinyl]phosphonate (la), with one equiv of benzaldehyde (3a) gave a mixture of 2-methylsulfonyl-1,4-diphenyl-1,3-butadiene (4a)(15%), diethyl (2-methylsulfonyl-3-phenyl-2-propenyl)phosphonate (5a)(35%), and diethyl [2-diethoxyphosphinyl-1-(methylsulfonyl)ethyl]phosphonate (6a)(14%). This result suggests the possibility of an one-pot synthesis of 1,3dienes and their analogues via the double Horner-Wittig reaction.

Accordingly, the reaction of 2a with an equiv of 3a, followed by a sequential treatment with an equiv

each of lithium diisopropylamide (LDA) and 3a, afforded the expected 4a in 63% yield. reactions of 2a with aromatic aldehydes such as pchlorobenz- (3b), cinnam- (3c), and 2-thiophenecarbaldehydes (3d) gave the corresponding dienes and tetraenes 4b-d in 42-53% yields. A similar treatment of [1-(ethoxycarbonyl)vinyl]phosphonate 1b with 3a,d produced dienes 4e,f, albeit in low vields.

An application of similar procedures using the phosphonate la to phthalaldehyde successfully gave 2-methylsulfonylnaphthalene (7) in 31% yield via an intramolecular double Horner-Wittig reaction (Eq. 1).

1a 
$$\xrightarrow{\text{LiP}(OEt)_2}$$
  $\xrightarrow{\text{CHO}}$   $\xrightarrow{\text{CH$ 

In contrast to aromatic aldehydes, the one-pot reaction of la with butyraldehyde (3e) under similar conditions afforded an unexpected product, diethyl 1,3-hexadienylphosphonate (8) in 49% yield, whose formation can not be reasonably accounted for at present. On the other hand, a similar treatment of 1b with 3e produced only an 1:1 mixture of diethyl (E)-(9a) and (Z)-(2-ethoxycarbonyl-2-hexenyl)phosphonates (9b) in 40% yield.

In conclusion, vinylphosphonates serve as versatile reagents for the one-pot synthesis of dienes, tetraenes,

$$\begin{array}{c} O \\ (EtO)_{2}\overset{P}{P} & O \\ C^{-}CH_{2} + Li\overset{P}{P}(OEt)_{2} & \longrightarrow \\ & X & \\ \hline & 1a, \ b & 2 \\ \hline & & & \\ \hline & &$$

and condensed aromatic compounds, which could be useful as synthetic precursors.

## **Experimental**

General. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra were taken in a CDCl<sub>3</sub> solution on a JEOL JNM-FX-60 operating at 60 and 15.04 MHz with Me<sub>4</sub>Si as the internal standard. IR spectra were recorded with a Shimadzu IR-408 instrument. Mass spectra were taken with a JEOL DX-300 spectrometer. Melting points were measured in open capillary tubes and are uncorrected.

Reaction of Diethyl [1-(Methylsulfonyl)vinyl]phosphonate (1a) with Benzaldehyde (3a) in the Presence of One Equiv. of Diethyl Lithiophosphonate. To a cooled solution of diethyl lithiophosphonate, prepared from diethyl phosphonate (0.33 g, 2.39 mmol) and butyllithium (2.39 mmol), in THF (20 ml) at -75 °C was added 1a (0.52 g, 2.15 mmol). The solution was stirred for 0.5 h at this temperature. Then 3a (0.228 g, 2.15 mmol) was added to the solution and the mixture was stirred at room temperature for 4 h. After a conventional work-up, the residue was chromatographed on a preparative TLC with ether as the eluent to give 4a (90 mg, 0.317 mmol, 15%), 5a (0.25 g, 0.753 mmol, 35%), and 6a (0.11 g, 0.289 mmol, 14%), respectively.

2-Methylsulfonyl-1,4-diphenyl-1,3-butadiene (4a): Mp 91—93 °C; IR (KBr): 1300, 1140 cm<sup>-1</sup>;  $^1$ H NMR δ=3.04 (3H, s, Me), 6.70—7.90 (13H, m, olefinic and phenyl H); MS Found: m/z 284.0868. Calcd for  $C_{17}H_{16}O_2S$ : 284.0871 (M<sup>+</sup>).

Found: C, 71.74; H, 5.80. Calcd for  $C_{17}H_{16}O_2S$ : C, 71.80; H. 5.67%.

Diethyl (2-Methylsulfonyl-3-phenyl-2-propenyl)phosphonate (5a): Oil; IR (neat): 1300, 1250, 1130, 1020, 960 cm<sup>-1</sup>; <sup>1</sup>H NMR δ=1.30 (6H, t, Me), 3.25 (3H, s,  $SO_2Me$ ), 3.40, 3.41 (2H, 2×d, 22.27, 22.12 Hz, P-CH<sub>2</sub>), 3.80—4.50 (4H, m, OCH<sub>2</sub>CH<sub>3</sub>), 7.10—8.00 (6H, m, olefinic and phenyl H); MS Found: m/z 333.0931. Calcd for C<sub>14</sub>H<sub>22</sub>O<sub>5</sub>SP: 333.0926 (M++1).

Diethyl [2-Diethoxyphosphinyl-1-(methylsulfonyl)ethyl]phosphonate (6a): Mp 43—45 °C; IR (KBr): 1310, 1240,1135, 1010, 960 cm<sup>-1</sup>; <sup>1</sup>H NMR δ=1.00—1.70 (12H, m, Me), 2.00—3.00 (3H, m, -CH<sub>2</sub>-CH), 3.29 (3H, s, SO<sub>2</sub>Me), 3.70—4.60 (8H, m, OCH<sub>2</sub>CH<sub>3</sub>); MS m/z 380 (M<sup>+</sup>).

General Procedure for the One-Pot Synthesis of 1,3-Dienes 4a—f. To a stirred solution of a phosphonate carbanion 2 (3 mmol) in THF (20 ml) at -75 °C was added an aldehyde (3 mmol). After stirring at room temperature for 4 h, the solution was cooled to -75 °C and LDA (3 mmol) in THF (5 ml) was added to the solution. After the solution was stirred at this temperature for 0.5 h, an aldehyde (3 mmol) was added to the solution. The reaction mixture was then stirred at room temperature for 10 h. After a similar work-up, a pure sample of each was obtained by preparative TLC.

2-Methylsulfonyl-1,4-diphenyl-1,3-butadiene (4a): Yield 0.54 g (1.90 mmol, 63%).

2-Methylsulfonyl-1,4-bis(p-chlorophenyl)-1,3-butadiene (4b): Yield 0.442 g (1.25 mmol, 42%); Mp 149—151 °C; IR (KBr): 1300, 1140 cm<sup>-1</sup>; <sup>1</sup>H NMR  $\delta$ =3.04 (3H, s, Me), 6.60—7.80 (11H, m, olefinic and phenyl H); MS m/z 352, 354, 356 (M<sup>+</sup>).

Found: C, 57.83; H, 4.00. Calcd for C<sub>17</sub>H<sub>14</sub>O<sub>2</sub>SCl<sub>2</sub>: C, 57.80; H, 3.99%.

4-Methylsulfonyl-1,8-diphenyl-1,3,5,7-octatetraene (4c): Yield 0.51 g (1.52 mmol, 51%); Mp 143—145 °C; IR (KBr): 1300, 1120 cm<sup>-1</sup>;  $^{1}$ H NMR  $\delta$ =3.00 (3H, s, Me), 6.40—7.80 (17H, m, olefinic and phenyl H); MS m/z 336 (M<sup>+</sup>).

Found: C, 74.89; H, 6.22. Calcd for C<sub>21</sub>H<sub>20</sub>O<sub>2</sub>S: C, 74.97; H, 5.99%.

2-Methylsulfonyl-1,4-di(2-thienyl)-1,3-butadiene (4d): Yield 0.47 g (1.59 mmol, 53%); Mp 123—125 °C; IR (KBr): 1290, 1130 cm<sup>-1</sup>;  $^{1}$ H NMR  $\delta$ =3.03 (3H, s, Me), 6.70—8.00 (9H, m, olefinic and aromatic H); MS m/z 296 (M<sup>+</sup>).

Found: C, 52.68; H, 4.02. Calcd for C<sub>13</sub>H<sub>12</sub>O<sub>2</sub>S<sub>3</sub>: C, 52.68; H, 4.08%.

2-Ethoxycarbonyl-1,4-diphenyl-1,3-butadiene (4e): Yield 0.218 g (0.786 mmol, 26%); Oil; IR(neat) 1710 cm<sup>-1</sup>; <sup>1</sup>H NMR δ=1.38 (3H, t, Me), 4.34 (2H, q,  $\frac{\text{CH}_2\text{CH}_3}{\text{CH}_3}$ ), 6.50—7.80 (13H, m, olefinic and phenyl H); MS Found: m/z 278.1302. Calcd for C<sub>19</sub>H<sub>18</sub>O<sub>2</sub>: 278.1302 (M<sup>+</sup>).

2-Ethoxycarbonyl-1,4-di(2-thienyl)-1,3-butadiene (4f): Yield 0.248 g (0.855 mmol, 29%); Oil; IR (neat) 1700 cm<sup>-1</sup>;  $^{1}$ H NMR  $\delta$ =1.37 (3H, t, Me), 4.32 (2H, q,  $^{1}$ CH<sub>2</sub>CH<sub>3</sub>), 6.50—7.80 (9H, m, olefinic and phenyl H); MS m/z 290 (M+). Found: C, 62.05; H, 4.88. Calcd for  $^{1}$ C<sub>15</sub>H<sub>14</sub>O<sub>2</sub>S<sub>2</sub>: C, 62.04;

H, 4.86%. Synthesis of 2-Methylsulfonylnaphthalene (7). The similar reaction of **1a** (2 mmol) with phthalaldehyde (0.268 g, 2 mmol) gave 0.126 g (31%) yield of **7** as white solid: Mp 138—140 °C; IR (KBr) 1290, 1120 cm<sup>-1</sup>; <sup>1</sup>H NMR  $\delta$ =3.12 (3H, s, Me), 7.20—8.60 (7H, m, aromatic H); MS

Found: C, 63.64; H, 4.92. Calcd for  $C_{11}H_{10}O_2S$ : C, 64.06; H, 4.89%.

One-Pot Reaction of 1a,b with Butyraldehyd '?e). The Reaction Using 1a: The reaction of 1a (5 mmol) with 3e (0.433 g, 6 mmol) gave 0.323 g (49%) yield of 8: Oil: IR (neat) 1240, 1020, 960 cm<sup>-1</sup>;  $^{1}$ H NMR  $\delta$ =1.04 (3H, t, J=7.03 Hz, CH<sub>2</sub>CH<sub>3</sub>), 1.32 (6H, t, J=7.03 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 1.80—2.50 (2H, m, CH<sub>2</sub>CH<sub>3</sub>), 3.70—4.20 (4H, m, OCH<sub>2</sub>CH<sub>3</sub>), 5.10—7.60 (4H, m, olefinic H);  $^{13}$ C NMR  $\delta$ =12.57, 16.12 (d,  $^{3}J_{P}$ -=6.88 Hz), 25.49, 61.27 (d,  $^{2}J_{P}$ -=5.16 Hz), 114.34 (d,  $^{1}J_{P}$ -=191.70Hz), 128.20 (d,  $^{3}J_{P}$ -=27.51 Hz), 144.77, 148.92 (d,  $^{2}J_{P}$ -=6.02 Hz). MS Found: m/z 218.1075. Calcd for C<sub>10</sub>H<sub>19</sub>O<sub>3</sub>P: 218.1072.

The Reaction Using 1b: The reaction of 1b (0.70 mmol) with 3e (0.106 g, 1.47 mmol) produced 0.082 g (0.27 mmol, 40%) of an 1:1 mixture of 9a and 9b: Oil; IR (neat) 1710, 1250, 1030, 970 cm<sup>-1</sup>; <sup>1</sup>H NMR  $\delta$ =1.29 (9H, t, J=7.03 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 0.60—1.90 (5H, m, =CHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.90—3.30 (4H, m, PCH<sub>2</sub>- and =CHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 3.70—4.50 (6H, m, OCH<sub>2</sub>CH<sub>3</sub>), 5.85—6.35 (0.5H, m, vinylic H), 6.70—7.15 (0.5H, m, vinylic H); MS Found: m/z 292.1413. Calcd for C<sub>13</sub>H<sub>25</sub>O<sub>5</sub>P: 292.1438.

### References

m/z 206 (M<sup>+</sup>).

- 1) For the preceding paper in this series, see: T. Minami, K. Nishimura, I. Hirao, H. Suganuma, and T. Agawa, J. Org. Chem., 47, 2360 (1982).
- 2) T. Minami, H. Suganuma, and T. Agawa, *Chem. Lett.*, **1978**, 285.