## COMMUNICATIONS

## Sulfur-free Penicillin Derivatives. VI. Synthesis of the 1-Oxacephem Ring System

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The title compound has been synthesized by ring closure of an alkyl 2-(2'-chloro-3'-S-phthalimido-4'-oxo)azetidinyl-3-methyl-4-hydroxy-*trans*-2-butenoate using  $S_N1$  experimental conditions. The required *E*-allylic alcohols were obtained in two ways starting with an alkyl 2-(2'-chloro-3'-S-phthalimido-4'-oxo)azetidinyl-3-methyl-2-butenoate. The first involves allylic bromination with 1 molar equiv. of *N*-bromosuccinimide, conversion of the mixed *Z*- and *E*-allylic bromides into the mixed *Z*- and *E*-allylic formates, and deformylation to a mixture of *E*-alcohol and lactone. The alcohol was separated from the lactone via its tetrahydropyranyl derivative, and then regenerated. In the second route, allylic bromination was performed with 2 molar equiv. of *N*-bromosuccinimide and the resulting dibromo compound was converted into the  $\beta,\gamma$ -unsaturated olefin with zinc in acetic acid. Epoxidation with peroxytrifluoroacetic acid followed by exposure to triethylamine again afforded a mixture of *E*-alcohol and lactone. The u.v. spectrum of a 1-oxacephem shows a maximum at 266 nm.

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Le composé désiré a été préparé par cyclisation d'un (chloro-2' S-phtalimido-3' oxo-4') azétidinyl-2 méthyl-3 hydroxy-4 butène-2 oate d'alkyle *trans* utilisant des conditions expérimentales  $S_NI$ . Les alcools allyliques-*E* nécessaires ont été obtenus de deux façons en partant d'un (chloro-2' S-phtalimido-3' oxo-4') azétidinyl-2 méthyl-3 butène-2 oate d'alkyle. La première façon implique une bromuration allylique avec un équivalent molaire de N-bromosuccinimide, la transformation du mélange des bromures allyliques-*E* et -*Z* en un mélange de formiates allyliques -*E* et -*Z* et la déformylation pour donner un mélange d'alcool -*E* et de lactone. L'alcool est séparé de la lactone par l'intermédiaire de son dérivé tétrahydropyrannyle et est ensuite regénéré. Dans la deuxième route, la bromuration allylique est effectué avec deux équivalents de *N*-bromosuccinimide et le composé dibromé qui en résulte est l'aide du zinc dans l'acide acétique. L'époxydation de cet oléfine par l'acide peroxytrifluoroacétique suivie par un contact avec de la triéthylamine conduit à nouveau à un mélange d'alcool-*E* et de lactone. Le spectre u.v. de l'oxa-1 céphem montre un maximum à 266 nm.

We wish to report the transformation of the thiazolidine ring of the penicillin nucleus into a dihydrooxazine ring, to form the oxygen analog (1*a*) of a  $\Delta^3$ -cephem (1*b*).

The starting materials were the (2R)-*cis*- and (2S)-*trans*-chloroazetidinones 2a-c (Ft = phthalimido), which are obtained (1) from anhydro-6phthalimidopenicillin (2) by chlorinolysis, hydrolysis of the derived acid chloride, and esterification with diazomethane or diphenyldiazomethane. Each of these azetidinones was monobrominated (NBS) (3, 4) to give approximately 1:1 mixtures of 3a + b, 3c + d, and 3e + f. The functionality present in the *E*-isomers 3b, 3*d*, and 3*f* allows two approaches to be considered for the generation of the six-membered ring, which differ in the timing of attachment of the C—O bond to the  $\beta$ -lactam. In the route described here, this bond has been formed in the final step of the synthesis, by cyclization of the *E*-allylic alcohols 4a-c.

The conversion of the allylic bromides into the required allylic alcohols was achieved in two ways. In the first, the mixtures 3a + b, 3c + d, and 3e + f were treated, in chloroform or methylene chloride at room temperature, with excess tetramethylguanidinium formate to form mixtures of the formate esters 5a + b (64%).

5c + d(57%), and 5e + f(52%). The esters were obtained in each case as approximately 2:1 Z:Emixtures; 5a-d have been described previously (3). The n.m.r. spectrum of 5e + f showed peaks at 8.22 (0.33H, s), 8.10 (0.67H, s), 7.83 (4H, d), 7.37 (10H), 7.03 (1H, s), 6.13 (1H, d, 4.0), 5.68 (1H, d, 4.0), 5.37 (2H, br s), 2.37 (2.0H, s), 2.32 (1.0H, s). Removal of the formyl groups (HCl, MeOH-CH<sub>2</sub>Cl<sub>2</sub>,  $0^{\circ}$ , 1.5 h) gave the Ealcohols 4a-c, admixed with the lactones 6a or 6b. The tetrahydropyranyl ethers 7a-c were separated from the lactones by chromatography on silica gel to yield, from 5a + b: 6a (59%) and

**2**a

ĊO<sub>2</sub>Me

**3**a

<sup>|</sup>CO<sub>2</sub>Me

30

B

Br

Br

ĊO<sub>2</sub>CHPh<sub>2</sub>

3e

7a (36%), m.p. 115–116.5°; from 5c + d: 6b(62%) and 7b (31%); from 5e + f: 6a (63%) and 7c (29%). Removal of the tetrahydropyranyl protecting groups (HCl, MeOH– $CH_2Cl_2$ , 0°, 4 h) afforded. Compound 4a (93%), m.p. 134.5-135.5°; anal. found: C, 54.09; H, 4.20; N, 7.44; n.m.r.: 7.79 (4H, d), 6.17 (1H, d, 4.0), 5.71 (1H, d, 4.0), 4.75 (1H, d, 12.5), 4.25 (1H, d, 12.5), 3.82 (3H, s), 2.78 (1H, s, exchanges with D<sub>2</sub>O), 2.38 (3H, s); **4***b* (98%), n.m.r.: 7.75 (4H, d), 6.18 (1H, d, 1.5), 5.55 (1H, d, 1.5), 4.30 (1H, d, 13), 4.27 (1H, d, 13), 3.85 (3H, s), 3.17 (1H, br s), 2.35 (3H); 4c (95%), n.m.r.: 7.83 (4H, d), 7.25

CO2CHPh2

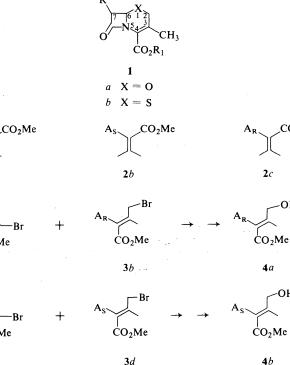
-OH

-OH

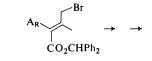
-OH

ĊO<sub>2</sub>CHPh<sub>2</sub>

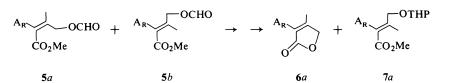
**4**c







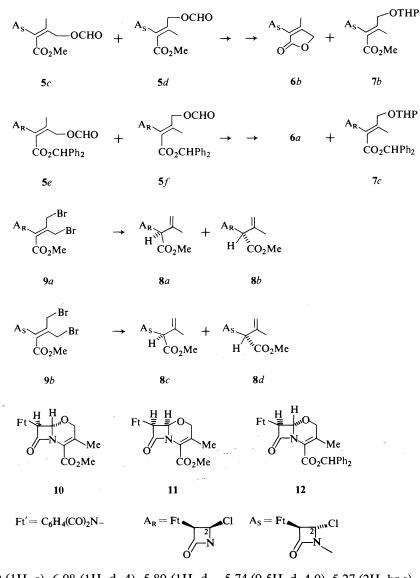




3f

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(10H), 7.00 (1H, s), 6.08 (1H, d, 4), 5.80 (1H, d, 4), 4.32 (1H, d, 14), 4.08 (1H, d, 14), 2.80 (1H, br s), 2.38 (3H, s).

A second synthesis of the alcohols 4a and 4bproceeded via the  $\beta$ ,  $\gamma$ -unsaturated olefins 8*a*-*d*. These were prepared by zinc dust debromination (2 equiv. zinc/mol, HOAc,  $0^{\circ}$ , 4 min) of the monobromo compounds 3a + b and 3c + d or, better, by debromination of the dibromo compounds 9a and 9b (4.4 equiv. zinc/mol, HOAc,  $0^{\circ}$ , 4 min). The  $\beta$ ,  $\gamma$ -isomers were obtained as 1:1 mixtures of epimers in each case; n.m.r. of 8a + b (78% from 9a): 7.92 (4H, d), 6.43 (0.5H, d, 4.0), 6.05 (0.5H, d, 4.0), 5.76 (0.5H, d, 4.0), These were separated as described above. The

5.74 (0.5H, d, 4.0), 5.27 (2H, br s), 5.14 (0.5H, s), 4.76 (0.5H, s), 3.85 (3H, s), 2.05 (3H, br s); n.m.r. of 8c + d (90% from 9b): 7.77 (4H), 5.95 (0.5H, d, 2), 5.88 (0.5H, d, 2), 5.57 (0.5H, d, 2), 5.47 (0.5H, d, 2), 5.25 (1H, br s), 5.15 (1H, br s), 4.83 (0.5H, s), 4.22 (0.5H, s), 3.83 (1.5H, s), 3.80 (1.5H, s), 2.00 (3H, br s). Each of the  $\beta$ ,  $\gamma$ isomers rearranged rapidly to the  $\alpha,\beta$ -isomer upon exposure to triethylamine (5). Oxidation of the mixture 8a + b (CF<sub>3</sub>CO<sub>3</sub>H, CH<sub>2</sub>Cl<sub>2</sub>, 20°, 1.5 h), afforded a mixture of four epoxides which, without purification, was rearranged (triethylamine-methanol) to a 3:1 mixture of 6a and 4a.

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mixture 8c + d similarly afforded a 2:1 mixture of the lactone 6b and the *E*-allylic alcohol 4b.

Depending on the experimental conditions (4, 6), substitution at C-2 of these azetidinones may proceed by  $S_N 1$  or  $S_N 2$  pathways. When  $S_N 2$ conditions are employed, the rate of the reaction depends upon the nature of the C-3 substituent in the order  $H_2N > acylamino > phthalimido$ (1). The alcohols 4a-c contain a phthalimido substituent at C-3 and did not cyclize under a variety of  $S_N 2$  conditions (Et<sub>3</sub>N, NaH, LiH). Unimolecular conditions were therefore necessary, with the result that the oxacephems 1awere obtained as epimeric mixtures at C-6, upon treatment of 4a-c with stannous chloride in dimethoxyethane or tetrahydrofuran or, in the case of the crystalline alcohol 4a, upon heating to the melting point.

Cyclization proceeded smoothly to give oxacephems in yields of 80-92% and, under identical conditions, the same mixture of epimers was formed from 4a and b, indicative of a common intermediate (4, 6). Some variation in the proportions of these epimers could be achieved by systematic variation of the experimental conditions. Thus, treatment of 4a (229 mg) at room temperature with 1.38 molar equiv. of SnCl<sub>2</sub> in dimethoxyethane (15 ml) yielded 176 mg (86% after recrystallization from chloroform-hexane) of 10, m.p. 170-171°,  $[\alpha]_D - 16.9$  (c 0.3, CHCl<sub>3</sub>).

Anal. Calcd. for  $C_{17}H_{14}N_2O_6$ : C, 59.65; H, 4.12; N, 8.18. Found: C, 59.78; H, 3.94; N, 8.42.

The i.r. (KBr): 5.61, 5.65, 5.79µ; n.m.r.: 7.80 (4H, d), 5.33 (1H, d, 1.8), 5.27 (1H, d, 1.8), 4.37

(2H, s), 3.90 (3H, s), 2.03 (3H, s). The u.v. spectrum of 10 shows  $\lambda_{max}$  (EtOH) 266 (12 500), 226 (40 000). The presence of a maximum at 266 nm supports conclusions (7, 8) that the  $\Delta^3$ -cephem chromophore near 260 nm is not associated with the presence of sulfur. Repetition of the above experiment with 1.67 molar equiv. of SnCl<sub>2</sub> in dimethoxyethane (30 ml) afforded, in 90% yield, a 1:1 mixture of 10 and 11, which possesses the natural configuration at C-6;  $[\alpha]_{D}$ 52.5 (c 0.3, CHCl<sub>3</sub>); M<sup>+</sup>, 342; n.m.r.: 7.80 (4H, d), 5.56 (1H, d, 3.8), 5.10 (1H, d, 3.8), 4.30 (2H), 3.90 (3H, s), 2.03 (3H, s). The n.m.r. spectrum of the oxacephem 12 showed peaks at 7.68 (4H, d), 7.23 (10H), 6.83 (1H), 5.20 (1H, d, 1.5), 5.13 (1H, d, 1.5), 4.26 (2H), 1.97 (3H).

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