```
BIS(1-TRIFLUOROMETHYL-2,2,2-TRIFLUOROETHOXY)TRIPHENYLPHOSPHORANE.
A NEW ROUTE TO TRIFLUOROMETHYLATED HETEROCYCLES
```

Toshio KUBOTA<sup>\*</sup>, Kazuhiro YAMAMOTO, and Tatsuo TANAKA Department of Industrial Chemistry, Ibaraki Uiversity, Hitachi-shi, Ibaraki 316

Bis(l-trifluoromethyl-2,2,2-trifluoroethoxy)triphenylphosphorane(<u>1</u>) readily reacted with anilines to form N-hexafluoroisopropylated products(<u>2</u>). By the dehydrofluorination of <u>2</u>, 2-arylamino-1,1,3,3,3-pentafluoropropenes(<u>3</u>) were obtained. Furthermore, when anilines with  $\rho$ -functional group were used as nucleophile, benzo-1,4-dihetero-six membered rings that have CF<sub>3</sub> group at N- $\alpha$ -position were obtained.

Phosphoranes with fluoroalkoxy group as ligand are known to be remarkable compounds as to the stereochemistry for phosphorus atom.<sup>1)</sup> However, little other utilities of these compounds in the synthesis of organofluorine compounds have been studied.<sup>2)</sup>

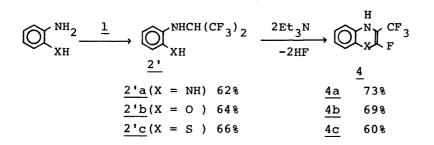
On the other hand, heterocyclic compounds with trifluoromethyl group are attracting attention owing to their biological unique properties in these days. $^{3)}$ 

In our continuing study of phosphoranes,  $^{4-6)}$  we have found a new route to introduce the trifluoromethyl group into a desired position of heterocycles.

Bis(1-trifluoromethy1-2,2,2-trifluoroethoxy)triphenylphosphorane(<u>1</u>) was readily reacted with anilines to produce the corresponding N-hexafluoroisopropylated derivertives(<u>2</u>). This fact shows the remarkable contrast to analogous phosphorane, i.e., bis(2,2,2-trifluoroethoxy)triphenylphosphorane, which was inert to N-nucleophiles.<sup>6</sup> Resulting N-hexafluoroisopropylanilines were dehydrofluorinated to form very interesting 2-arylamino-1,1,3,3,3-pentafluoropropenes(<u>3</u>) by base, e.g., triethylamine, potassium <u>tert</u>-buthoxide, etc. However, it is impossible to obtain these compounds(<u>3</u>) directly from the reaction of hexafluoropropene(HFP) with anilines, because HFP is attacked on the terminal CF<sub>2</sub> group by the nucleophiles.<sup>7</sup>

We also found that wider variety of anilines with o-functional group underwent the addition-elimination reaction to afford the trifluoromethylated heterocycles(4).

In conclusion, we believe that bis(1-trifluoromethy1-2,2,2-trifluoroethoxy)triphenylphosphorane is useful intermediate to introduce the trifluoromethyl group into a desired position of heterocycles, and that this procedure provides a practical and convenient route to heterocycles.



In a typical procedure,  $\rho$ -phenylenediamine(1.80g, 10mmol) was added into a solution of <u>1</u> in dichloromethane(40ml)-diethyl ether(10ml) which was prepared from triphenylphosphine dibromide(4.22g, 10mmol) and sodium 1-trifluoromethyl-2,2,2-tr-ifluoroethoxide(3.80g, 20mmol) in situ.<sup>6</sup>) After 1h of stirring, precipitates(NaBr) were centrifuged from the reaction mixture, and the solvent was removed. then distillation gave N-hexafluoroisopropylated product(<u>2'a</u>, 1.51g) in a yield of 62%, b.p.84.0-85.0°C/7mmHg. <sup>1</sup>H nmr(CDCl<sub>3</sub>) :  $\delta$ 4.40(NH), 4.50(NH<sub>2</sub>), 5.40(sep, J<sub>H-F</sub>=6.1Hz, 1H), 6.20-6.75(4H). <sup>19</sup>F nmr(CDCl<sub>3</sub>) :  $\delta$ +1.22(d, CF<sub>3</sub>) from CF<sub>3</sub>COOH as an ext. standard. MS : m/e 258(M<sup>+</sup>).

The mixture of <u>2'a</u>(1.22g, 5mmol), triethylamine(1.50g, 15mmol) and tetrahydrofuran(30ml) was refluxed for lh, and then worked-up as usual. 2-Fluoro-3-trifluoromethyl-1,4-dihydroquinoxaline(<u>4a</u>, 0.80g) was obtained by column chromatography on silica gel using chloroform as eluent in 73% yield, m.p.89.5-91.0°C. <sup>1</sup>H nmr(CDCl<sub>3</sub>) :  $\oint 4.40(2H)$ , 6.10-6.90(4H). <sup>19</sup>F nmr(CDCl<sub>3</sub>) :  $\oint -15.9(d, J_{CF_3}-F^{=16.2Hz}, CF_3)$ , -19.1(q, CF) from CF<sub>3</sub>COOH as an ext. standard. MS : m/e 238(M<sup>+</sup>).

References

- 1. a) E.L.Muetterties and R.A.Shunn, Quart. Rev. Chem., Chem. Soc., <u>20</u>, 245(1966).
  b) E.L.Muetterties, Acc. Chem. Res., 3, 226(1970).
- 2. D.Daktenieks, G.V.Roshenthaler and R.Schmutsler, J.Fluorine Chem., 11, 378(1978).
- 3. Reviews; B.Filler, "Organofluorine Chemicals and their Industrial Applications", ed., by R.E.Banks, Soc. Chem. Ind., London(1979), p.123; W.G.M.Jones, ibid., p.154; G.T.Newbolt, ibid., p.169.
- 4. T.Kubota, T.Kitazume and N.Ishikawa, Chem. Lett., 1978, 889.
- 5. T.Kubota, S.Miyashita, T.Kitazume and N.Ishikawa, Chem. Lett., 1979, 845.
- 6. T.Kubota, S.Miyashita, T.Kitazume and N.Ishikawa, J.Org. Chem., 45, 5052(1980).
- 7. R.D.Chambers, F.G.Drakesmith and W.K.R.Musgrave, J. Chem. Soc., 1965, 5045.

(Received November 17, 1982)