Photochemical Cyclodimerisation and Rearrangement of 5*H*-Dibenz[*b*,*f*]azepine Derivatives

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Summary Unsensitised and benzophenone-sensitised irradiation of N-acyl, -aroyl, and -ethoxycarbonyl derivatives of 5H-dibenz[b,f]azepine (1) (iminostilbene) gives good yields of cyclobutane dimers (2); the parent compound and its N-alkyl derivatives are photochemically inactive, whereas the N-tosyl compound undergoes photo-Fries rearrangement.

The 5H-dibenz[b,f]azepine ring system (1) (iminostilbene) and its 10,11-dihydro analogue (iminobibenzyl) have received considerable attention in recent years, since derivatives bearing a γ -dialkylaminopropyl substituent at the 5-position have pharmacological activity. However much of the fundamental chemistry of these ring systems remains unexplored. Descriptions of the photochemistry of 5H-

dibenz[b, f]azepines are limited to the parent compound (1a) and to 5-nitrosodibenz[b, f]azepine (1i). Both materials were found to be photostable in the presence of argon but irradiation of the latter material in the presence of oxygen gave 2-nitrodibenz[b, f]azepine (3b).²

Benzophenone-sensitised irradiation (Hanovia Reading Reactor; Pyrex) of a degassed acetone or benzene solution of N-acetyliminostilbene (1b)† in vacuo gave the dimer (2a) (77% after 5 min; quantum yield Φ 0·27), m.p. 342—345° (M^+ at m/e 470). The unsensitised dimerisation was much less efficient (36% after 1 h). In both cases the presence of dissolved air (O₂) had a retarding effect on the photodimerisation suggesting triplet-state intermediates. Michler's ketone ($E_{\rm T}$ 61·0 kcal mol⁻¹), but not fluorenone ($E_{\rm T}$ = 53·3 kcal mol⁻¹), was found to sensitise the dimerisa-

† All new compounds had satisfactory elemental analyses and spectra.

tion in acetone or benzene indicating that $E_{\mathbf{T}}$ for (1b) lies in the region 53·3-61·0 kcal mol-1.

Similarly, sensitised irradiation of N-propionyl- (1c), N-chloroacetyl- (1d), N-benzoyl- (1e), and N-ethoxycarbonyl-iminostilbene (1g) afforded the dimers (2b), m.p. 323-325°, (2c), m.p. 317—320°, (2d), m.p. 300—304°, and (2e), m.p. 325-327°, respectively. Only one type of dimer could be isolated from each substrate and this, taken with

the presumed intermediacy of triplet state reactants, leads to an assumption of a thermodynamically stable transconfiguration for the cyclodimers.3

Sensitised and unsensitised photolyses of the parent (1a) and N-alkyl substituted [e.g. (1f)] iminostilbenes gave only unchanged starting materials. It is significant that in these compounds, the lowest energy transition (λ_{max}). 365 nm), which gives rise to yellow colouration,4 is (presumably) $n-\pi^*$ in character. The derivatives which photodimerised (1b-e,g) carry substituents which lower the energy of the non-bonding electron pair on nitrogen, as evidenced by u.v. spectra and absence of colour, and show photoreactivity similar to that of cis-stilbene analogues.

It should be noted however, that sensitised and unsensitised irradiation of 5-tosyliminostilbene (1h) afforded, in low conversion, 2-tosyliminostilbene (3a), m.p. 230—231°, ν(NH) 3350 cm⁻¹. This product was isolated by column chromatography (neutral Al₂O₃; benzene-methanol 9:1 v/v) and the position occupied by the tosyl group is assumed by analogy with the related photo-Fries rearrangement of sulpho-anilides, shown to yield exclusively paratosylanilines.5

The intermolecular [2 + 2] photodimerisation of 5-acyliminostilbenes is in contrast to derivatives of the 1Hazepine ring system which undergo internal [2 + 2] photodimerisation to produce the isomeric 2-azabicyclo[3,2,0]hepta-3,6-diene ring system.6

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