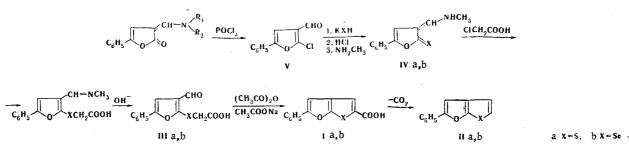
I. Ya. Kvitko, N. B. Sokolova, and L. S. Éfros UDC 547.724'733'739.3.07

We have synthesized new heterocyclic systems -2-phenylthieno[2,3-b]- and 2-phenylselenopheno[2,3-b]furan (IIa,b) - by decarboxylation of 2-phenylthieno[2,3-b]- and 2-phenylselenopheno[2,3-b]furan-5-carboxylic acids (Ia,b). Compounds Ia,b are formed in the cyclization of the corresponding 5-furylthio (or seleno)acetic acids (IIIa,b) in acetic anhydride in the presence of sodium acetate, together with IIa,b, in a ratio of ~ 2:1. In alkaline media, because of the high electrophilicity of the carbon in the 5 position, the -XCH₂COOH residue is replaced by a hydroxyl group to give 2-phenyl-4-formyl-5-hydroxyfuran [3], and cyclization of IIIa,b is not observed (see [1,2]). Compounds IIIa,b were synthesized from the enamines of the 4-formyl derivatives of thio- and selenofurans (IVa,b). 2-Phenyl-4-formyl-5-chlorofuran (V) was obtained by the method in [4] from disubstituted enamines.

EXPERIMENTAL

<u>2-Phenyl-4-formyl-5-furylthio (seleno)acetic Acids (IIIa,b).</u> A mixture of 0.01 mole of IV and 0.011 mole of perchloric acid was refluxed in benzene for 2 h, after which the precipitate was removed by filtration, washed with chloroform, and extracted with 10% aqueous sodium carbonate solution. The extract was acidified, and the resulting precipitate was recrystallized from benzene to give IIIa [83% yield, mp 129°, λ_{max} (ethanol) 296 nm (log ε 4.53). Found: S 12.5%. C₁₃H₁₀O₄S. Calculated: S12.2%] and IIIb [87% yield, mp 128°, λ_{max} (ethanol) 296 nm (log ε 4.09). Found: C 50.2; H 3.5%. C₁₃H₁₀O₄Se. Calculated: C 50.5; H 3.2%].

2-Phenylthieno[2,3-b] - and 2-Phenylselenopheno[2,3-b]furans (IIa,b). A 0.01-mole sample of III was refluxed in a mixture of 20 ml of acetic anhydride and 2 g of sodium acetate for 1 h. The precipitated substance was extracted several times with boiling petroleum ether, the solvent was removed by distillation, and the residue was recrystallized from aqueous ethanol (IIa,b). The residue remaining after extraction was dissolved in water, and the solution was acidified. The precipitate was removed by filtration and crystallized from ethanol (Ia,b). Compound Ia: 46% yield, mp 277°, λ_{max} (ethanol) 315 nm (log ϵ 4.27). Found: C 64.3; H 3.8; S 13.3%. C₁₃H₈O₃S. Calculated: C 63.8; H 3.8; S 13.1%; compound Ib was not isolated from the reaction mass. Compound IIa: 21% yield, mp 81°, λ_{max} (ethanol) 297 nm (log ϵ 4.37). Found: S 16.5%. C₁₂H₈OS. Calculated: S 16.0%. Compound IIb: 17% yield, mp 102°, λ_{max} (ethanol) 299 nm (log ϵ 4.50). Found: C 58.5; H 3.6%. C₁₂H₈OSe. Calculated: C 58.3; H 3.2%.



 $\frac{2-\text{Phenyl-4-methylaminomethylene-5-thio} \text{ (seleno) furanones (IVa,b). Compound IVa: 91\% yield, mp}{160^{\circ} \text{ (from benzene)}, \lambda_{\max} \text{ (ethanol) } 432 \text{ nm} \text{ (log } \epsilon \text{ 4.10)}. \text{ Found N 6.6; S } 14.7\%. C_{12}H_{11}\text{NOS}. \text{ Calculated: N 6.5; S } 14.9\%. \text{ Compound IVb: } 86\% \text{ yield, mp } 138^{\circ} \text{ (from benzene)}, \lambda_{\max} \text{ (ethanol)}, 452 \text{ nm} \text{ (log } \epsilon \text{ 4.01)}. \text{ Found: N 5.3\%. } C_{12}H_{11}\text{NOSe}. \text{ Calculated: N 5.3\%.}$

Lensovet Leningrad Technological Institute. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 5, pp. 715-716, May, 1973. Original article submitted October 23, 1972.

© 1975 Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.

2-Phenyl-4-formyl-5-chlorofuran (V). This compound was obtained in 84% yield and had mp 74° (from cyclohexane) and λ_{max} 278 nm (log ϵ 4.27). Found: Cl 17.1%. C₁₁H₇ClO₂. Calculated: Cl 17.2%.

LITERATURE CITED

- 1. I. Ya. Kvitko, Khim. Geterotsikl. Soedin., 760 (1969).
- 2. I. Ya. Kvitko and E. A. Panfilova, Khim. Geterotsikl. Soedin., 507 (1973).
- 3. R. S. Rao and R. Filler, Chem. Ind., 280 (1964).
- 4. I. Ya. Kvitko and B. A. Porai-Koshits, Zh. Organ. Khim., 2, 169 (1966).