

Synthesis of Some 5-Alkenyl-4-ethoxycarbonyl-1-phenylpyrazoles and Ethyl 1-Anilino-4-oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylates

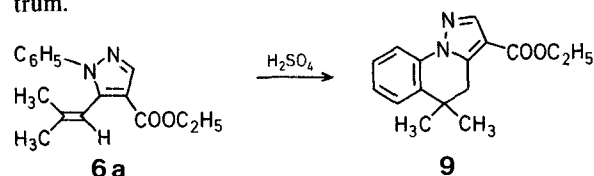
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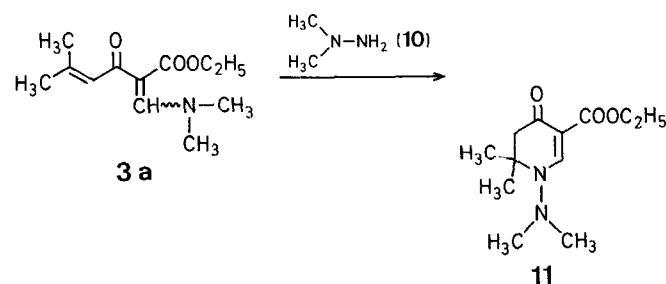
N,N-Dimethylamino derivatives of ketones and β -keto esters are useful intermediates in the synthesis of heterocyclic compounds¹⁻⁴. We report here the reaction of *N,N*-dimethylaminomethylene derivatives (**3**) of β -keto esters **1** bearing an alkenyl substituent with phenylhydrazine (**4**) to give alkenylpyrazoles **6** or 4-oxo-1,4,5,6-tetrahydropyridines **7** according to the reaction conditions employed.

Condensation of the γ,δ -unsaturated β -keto ester **1** with dimethylformamide dimethylacetal (**2**) affords, in almost quantitative yield, the crude enamine derivative **3**, which in turn, is submitted to the reaction of phenylhydrazine (**4**). The formation of three products could be expected by ring closure of the initially formed intermediate **5** followed by the attack of the hydrazino moiety at the carbonyl function or at the double bond. The former reaction will give the alkenylpyrazoles **6** as the product while the latter can yield either the 4-oxo-1,4,5,6-tetrahydropyridines **7** or the 5-oxo-2,5,6,7-tetrahydro-1H-1,2-diazepines **8**.

We have found that the 5-alkenyl-4-ethoxycarbonyl-1-phenylpyrazoles **6** are formed when the reaction is carried out in acetic acid as solvent. The structure of the alkenylpyrazoles **6** is in good agreement with the microanalytical and spectral data (Table 1). The position of the 1-phenyl group in compounds **6b-e** is deduced from the observed singlet resonance in the ¹H-N.M.R. spectra, characteristic of an α -substituent⁵. Confirmation of the position of the phenyl group in compound **6a** is supported by its electrophilic intramolecular cyclization to 5,5-dimethyl-3-ethoxycarbonyl-4,5-dihydropyrazolo[1,5-*a*]quinoline (**9**) as evidenced by the ¹H-N.M.R. spectrum.



When the reaction is carried out in anhydrous 1,2-dimethoxyethane as solvent, ethyl 1-anilino-4-oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylates (**7**) are formed. This ring system is very



stable; no rearrangement is observed by refluxing in acetic acid for 12 h or on standing at room temperature in trifluoroacetic acid. This lack of reactivity is in marked contrast to that observed for a 5-oxo-2,5,6,7-tetrahydro-1*H*-1,2-diazepine structure⁶. In order to support the tetrahydropyridine structure for compounds **7**, we synthesized ethyl 6,6-dimethyl-1-dimethylamino-4-oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylate (**11**) by reacting *N,N*-dimethylhydrazine (**10**) with the enamine derivative **3a** ($R^1=R^2=CH_3$, $R^3=H$).

The microanalytical, ¹H-N.M.R., I.R., and U.V. data for compounds **7** (see Table 2) and compound **11** are entirely consistent with the assignment of a tetrahydropyridine structure. This is also well confirmed by comparison of the ¹³C-N.M.R. chemical shifts for compounds **7a** and **11** (see Table 2). Moreover those chemical shifts for compound **7a** are different from those reported for a 7,7-dimethyl-5-oxo-2-phenyl-2,5,6,7-tetrahydro-1*H*-1,2-diazepine structure [$\delta=194.0$ (C-4); 57.5 (C-5); 68.5 ppm (C-6)]⁶.

Although a few syntheses of alkenylpyrazoles have been reported in the literature⁷, we found no examples of 1-anilino- or 1-amino-4-oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylic acid derivatives.

γ,δ -Unsaturated β -Keto Esters **1**:

These are prepared as described in the literature; ethyl 5-methyl-3-oxo-4-hexenoate (**1**; $R^1=R^2=CH_3$, $R^3=H$)⁸; ethyl (*E*)-4-methyl-3-oxo-4-hexenoate (**1**; $R^1=R^3=CH_3$, $R^2=H$)⁹; ethyl (*E*)-3-oxo-4-hexenoate (**1**; $R^1=CH_3$, $R^2=R^3=H$)¹⁰; ethyl 4-methyl-3-oxo-4-pentenolate (**1**, $R^1=R^2=H$, $R^3=CH_3$)¹⁰; ethyl (*E,E*)-3-oxo-4,6-octadienolate (**1**, $R^1=CH_3-CH=CH-$; $R^2=R^3=H$)¹¹.

5-Alkenyl-4-ethoxycarbonyl-1-phenylpyrazoles (**6a-e**); General Procedure:

A solution of dimethylformamide dimethylacetal (**2**; 1.91 g, 16 mmol) in benzene (10 ml) is added in one portion to a solution of the γ,δ -unsaturated β -keto ester **1** (10 mmol) in benzene (10 ml). The mixture is refluxed for 1 h. Evaporation of the solvent under reduced pressure affords the crude enamine derivatives **3**.

¹H-N.M.R. (CDCl₃/TMS_{int}): $\delta=2.91-3.01$ ppm [s, 6 H, N(CH₃)].

The enamine **3** obtained is dissolved in acetic acid (50 ml), phenylhydrazine (**4**; 1.08 g, 10 mmol) is added and the mixture is refluxed overnight. The reaction mixture is poured into water (250 ml) and extracted with chloroform (3 \times 50 ml). The combined organic phase is washed with 5% sodium hydrogen carbonate (2 \times 50 ml), water (50 ml), and then dried with sodium sulfate. The chloroform is evaporated and the residue is chromatographed through a column (22 mm \times 35 cm) of silica gel (50 g) using hexane/ethyl acetate (1 : 1) as eluent. The range of fraction collected for each product is given below.

Product	Fraction Collected	Yield [g]
6a	90 to 180 ml	2.11
6b	90 to 180 ml	2.11
6c	85 to 175 ml	1.90
6d	85 to 165 ml	1.88
6e	85 to 175 ml	2.09

Ethyl 4-Oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylates (**7a-d**); General Procedure:

The crude enamine **3** obtained as described above for the preparation of pyrazoles **6** is dissolved in anhydrous 1,2-dimethoxyethane (50 ml) and refluxed with phenylhydrazine (**4**; 1.08 g, 10 mmol) overnight. For the isolation of **7a-c**, the solvent is evaporated and the residue is

Table 1. 5-Alkenyl-4-ethoxycarbonyl-1-phenylpyrazoles **6a-e**

Product No.	R ¹	R ²	R ³	Yield [%]	m.p. [°C]	Molecular formula ^a	I.R. (CHCl ₃) ν [cm ⁻¹]	U.V. (C ₂ H ₅ OH) λ [nm] (ϵ)	¹ H-N.M.R. (CDCl ₃ /TMS _{int}) δ [ppm]
6a	CH ₃	CH ₃	H	78	oil	C ₁₆ H ₁₈ N ₂ O ₂ (270.3)	1710, 1605, 965	251 (14200)	1.36 (t, 3 H, $J=7$ Hz); 1.42 (s, 3 H); 1.88 (s, 3 H); 4.31 (q, 2 H, $J=7$ Hz); 6.05–6.25 (m, 1 H); 7.2–7.6 (m, 5 H); 8.10 (s, 1 H)
6b	CH ₃	H	CH ₃	78	75 ^{ob}	C ₁₆ H ₁₈ N ₂ O ₂ (270.3)	1715, 1605, 965	246 (14000)	1.36 (t, 3 H, $J=7$ Hz); 1.70 (apparent doublet, 3 H, $J=7$ Hz); 1.8–1.9 (m, 3 H); 4.33 (q, 2 H, $J=7$ Hz); 5.4–5.8 (m, 1 H); 7.46 (s, 5 H); 8.10 (s, 1 H)
6c	CH ₃	H	H	74	oil	C ₁₅ H ₁₆ N ₂ O ₂ (256.3)	1710, 1605, 965	257 (14000)	1.39 (t, 3 H, $J=7$ Hz); 1.83 (d, 3 H, $J=5.5$ Hz); 4.35 (q, 2 H, $J=7$ Hz); 6.1–6.8 (m, 2 H); 7.50 (s, 5 H); 8.10 (s, 1 H)
6d	H	H	CH ₃	73	78 ^{ob}	C ₁₅ H ₁₆ N ₂ O ₂ (256.3)	1710, 1600, 965, 910	244 (12400)	1.36 (t, 3 H, $J=7$ Hz); 2.0–2.1 (m, 3 H); 4.35 (q, 2 H, $J=7$ Hz); 5.10 (s, 1 H); 5.4–5.5 (m, 1 H); 7.50 (s, 5 H); 8.11 (s, 1 H)
6e^c	H ₃ C—CH=CH—	H	H	74	oil	C ₁₇ H ₁₈ N ₂ O ₂ (282.3)	1710, 1600, 990, 955	296 (17900)	1.34 (t, 3 H, $J=7$ Hz); 1.79 (d, 3 H, $J=5.5$ Hz); 4.38 (q, 2 H, $J=7$ Hz); 5.3–7.2 (m, 4 H); 7.51 (s, 5 H); 8.14 (s, 1 H)

^a The microanalyses were in satisfactory agreement with the calculated values: C ± 0.17 , H ± 0.14 , N ± 0.30 .

^b Recrystallized from hexane.

^c (*E,E*)-Isomer.

Table 2. Ethyl 4-Oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylates **7a-d**

Product No.	R ¹	R ²	R ³	Yield [%]	m.p. [°C]	Molecular formula ^a	I.R. (CHCl ₃) ν [cm ⁻¹]	U.V. (C ₂ H ₅ OH) λ [nm] (ε)	¹ H-N.M.R. (CDCl ₃ /TMS _{int}) δ [ppm]
7a	CH ₃	CH ₃	H	72	203°	C ₁₆ H ₂₀ N ₂ O ₃ (288.3)	3290, 1730, 1675, 1605, 1580	237 (16500), 306 (15900)	1.28 (t, 3 H, <i>J</i> = 7 Hz); 1.36 (s, 6 H); 2.69 (s, 2 H); 4.24 (q, 2 H, <i>J</i> = 7 Hz); 6.8–7.5 (m, 6 H, 1 H exchangeable with D ₂ O); 8.26 (s, 1 H) ^c
7b^b	CH ₃	H	CH ₃	52	183°	C ₁₆ H ₂₀ N ₂ O ₃ (288.3)	3280, 1730, 1670, 1610, 1585	238 (17300), 308 (15200)	1.1–1.5 (m, 9 H); 2.2–3.2 (m, 1 H); 3.3–4.4 (m, 3 H, with a quadruplet at 4.21); 6.8–7.5 (m, 6 H, 1 H exchangeable); 8.29, 8.33 (2s, 1 H)
7c	CH ₃	H	H	56	170°	C ₁₅ H ₁₈ N ₂ O ₃ (274.3)	3280, 1730, 1670, 1605, 1585	238 (16700), 307 (15100)	1.28 (t, 3 H, <i>J</i> = 7 Hz); 1.33 (d, 3 H, <i>J</i> = 7 Hz); 2.44 (dd, 1 H, <i>J</i> = 15 and 6 Hz); 2.65 (dd, 1 H, <i>J</i> = 15 and 6 Hz); 3.9–4.4 (m, 3 H with a quadruplet at 4.19); 6.8–7.5 (m, 6 H, 1 H exchangeable); 8.35 (s, 1 H)
7d	H	H	CH ₃	63	138°	C ₁₅ H ₁₈ N ₂ O ₃ (274.3)	3270, 1730, 1675, 1615, 1590	237 (15800), 306 (15600)	1.18 (d, 3 H, <i>J</i> = 7 Hz); 1.29 (t, 3 H, <i>J</i> = 7 Hz); 2.5–3.0 (m, 1 H); 3.3–3.9 (m, 2 H); 4.23 (q, 2 H, <i>J</i> = 7 Hz); 6.8–7.5 (m, 6 H, 1 H exchangeable); 8.44 (s, 1 H)

^a The microanalyses were in satisfactory agreement with the calculated values: C ± 0.16, H ± 0.14, N ± 0.08.^b Mixture of *cis/trans* isomers.^c ¹³C-N.M.R. (CDCl₃/TMS_{int}): δ = 51.5 (C-5); 63.1 (C-6); 187.8 ppm (C-4).

treated with ethyl acetate (10 ml). The crystals formed are collected by filtration. Compound **7d** is obtained by chromatography of the residue on a column (22 mm × 35 cm) of silica gel (50 g) using ethyl acetate as eluent from the fraction 150–350 ml; yield: 1.75 g (63%). Analytical samples of **7** are prepared by recrystallization from acetonitrile.

5,5-Dimethyl-3-ethoxycarbonyl-4,5-dihydropyrazolo[1,5-*a*]quinoline (9): A mixture of the alkenylpyrazole **6a** (1.35 g, 5 mmol) and concentrated sulfuric acid (13.5 ml) is stirred overnight at room temperature. The resulting mixture is diluted with crushed ice and extracted with chloroform (3 × 50 ml). The extracts are washed with water (2 × 50 ml) and dried with sodium sulfate. The chloroform is evaporated and the residue is chromatographed through a column (22 mm × 35 cm) of silica gel (50 g) using hexane/ethyl acetate (1:1) as eluent. The compound **9** is collected in the fraction 80–140 ml, oil; yield: 1.00 g (74%).

C₁₆H₁₈N₂O₂ calc. C 71.09 H 6.71 N 10.36
(270.3) found 71.14 6.73 10.31

I.R. (CHCl₃): ν = 1710 cm⁻¹ (C=O).

¹H-N.M.R. (CDCl₃/TMS_{int}): δ = 1.33 (s, 6 H); 1.39 (t, 3 H, *J* = 7 Hz); 3.24 (s, 2 H); 4.36 (q, 2 H, *J* = 7 Hz); 7.1–7.6 (m, 3 H); 7.9–8.2 ppm (m, 2 H, with a singlet at 8.09).

Ethyl 6,6-Dimethyl-1-dimethylamino-4-oxo-1,4,5,6-tetrahydro-3-pyridinecarboxylate (11):

A solution of dimethylformamide dimethylacetal (**2**; 1.91 g, 16 mmol) in benzene (10 ml) is added to a solution of **1a** (1.7 g, 10 mmol) in benzene. The mixture is refluxed for 1 h and the solvent is evaporated under reduced pressure to afford the crude enamine **3**. The enamine **3** is dissolved in anhydrous dimethoxymethane (50 ml) and refluxed with *N,N*-dimethylhydrazine (**10**; 0.6 g, 10 mmol) overnight. After evaporation of the solvent, the residue is triturated with hexane (20 ml) and the crystalline solid obtained is filtered under suction; yield: 2.09 g (87%). An analytical sample is obtained by recrystallization from hexane/ethyl acetate (4:1); m.p. 106 °C.

C₁₂H₂₀N₂O₃ calc. C 59.98 H 8.39 N 11.66
(240.3) found 59.72 8.58 11.56

I.R. (CHCl₃): ν = 1725, 1665, 1575 cm⁻¹.U.V. (C₂H₅OH): λ_{max} = 240 (11000); 309 nm (14800).

¹H-N.M.R. (CDCl₃/TMS_{int}): δ = 1.34 (t, 3 H, *J* = 7 Hz); 1.38 (s, 6 H); 2.51 (s, 2 H); 2.73 (s, 6 H); 4.28 (q, 2 H, *J* = 7 Hz); 8.39 ppm (s, 1 H).

¹³C-N.M.R. (CDCl₃/TMS_{int}): δ = 52.0 (C-5); 63.1 (C-6); 186.6 (C-4) ppm.

Received: February 11, 1983

- E. J. Breaux, K. E. Zwickelmaier, *J. Heterocyclic Chem.* **18**, 183 (1981).
- C. Kashima, S. Shirai, Y. Yamamoto, *Heterocycles* **12**, 657 (1979).
- Y. Lin, S. A. Lang Jr., *J. Org. Chem.* **45**, 4857 (1980).
- E. E. Garcia, L. E. Benjamin, R. Ian Fryer, *J. Heterocyclic Chem.* **11**, 275 (1974).
- I. L. Finar, D. M. Rackham, *J. Chem. Soc. [B]* **1968**, 211.
- S. N. Ege, M. L. C. Carter, R. L. Spencer, C. E. Nordman, H. Z. Friedman, *J. Chem. Soc. Perkin Trans. 1* **1976**, 868.
- C. Deshayes, M. Chabannet, S. Gelin, *J. Heterocyclic Chem.* **18**, 1057 (1981) and references cited therein.
- S. Gelin, R. Gelin, *Bull. Soc. Chim. Fr.* **1969**, 4091.
- H. A. Abramson, H. C. Wormser, *J. Heterocyclic Chem.* **18**, 363 (1981).
- L. Pichat, J. P. Beaucourt, *Synthesis* **1973**, 537.
- V. J. Lee, A. R. Branfman, T. R. Herrin, K. L. Rinehart Jr., *J. Am. Chem. Soc.* **100**, 4225 (1978).