



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

BF³ Etherate Mediated 1,4-Addition Reaction of (1-Alkenyl)dialkoxyboranes to α -Acyl- α,β -unsaturated Esters

Masataka Aoyama^a, Shoji Hara^a & Akira Suzuki^a

^a Department of Applied Chemistry, Faculty of Engineering, Hokkaido University, Sapporo, 060, Japan

Published online: 23 Sep 2006.

To cite this article: Masataka Aoyama, Shoji Hara & Akira Suzuki (1992) BF³ Etherate Mediated 1,4-Addition Reaction of (1-Alkenyl)dialkoxyboranes to α -Acyl- α,β -unsaturated Esters, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 22:17, 2563-2569, DOI: [10.1080/00397919208021652](https://doi.org/10.1080/00397919208021652)

To link to this article: <http://dx.doi.org/10.1080/00397919208021652>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and

are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

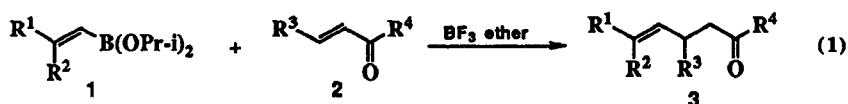
**BF₃ ETHERATE MEDIATED 1,4-ADDITION REACTION OF
(1-ALKENYL)DIALKOXYBORANES TO α -ACYL-
 α,β -UNSATURATED ESTERS**

Masataka Aoyama, Shoji Hara, and Akira Suzuki*

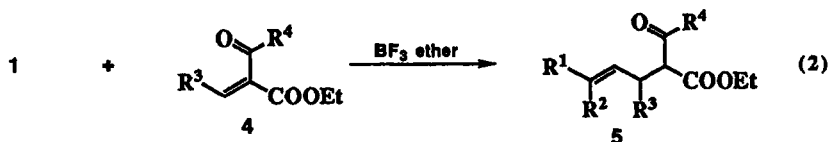
Department of Applied Chemistry, Faculty of Engineering,
Hokkaido University, Sapporo 060, Japan

Abstract: In the presence of BF₃ etherate, (1-alkenyl)dialkoxyboranes react with α -acyl- α,β -unsaturated esters to give α -acyl- γ,δ -unsaturated esters in good yields.

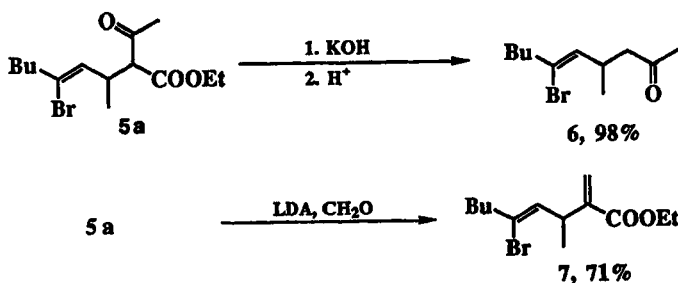
For the carbon-carbon bond formation, 1,4-addition of organometallic compounds to α,β -unsaturated ketones is one of the most efficient methods, among which the use of 1-alkenylborane derivatives for the synthesis of γ,δ -unsaturated ketones is convenient and practical because they can be prepared regio- and stereoselectively by the hydroboration reaction or haloboration reaction of alkynes, and add to α,β -unsaturated ketones without any formation of undesired 1,2-addition products.¹ Recently we reported² that (1-alkenyl)-dialkoxyboranes (**1**) react with α,β -unsaturated ketones in the presence of BF₃ etherate to give γ,δ -unsaturated ketones (**2**) stereoselectively (eq. 1).



However, other α,β -unsaturated carbonyl compounds such as unsaturated esters never react with **1** under the same or modified conditions. During the course of study on the 1,4-addition reaction of **1** with unsaturated carbonyl compounds, we found that α,β -unsaturated esters (**4**) having acyl group at the α -position react with **1** to give α -acyl- γ,δ -unsaturated esters (**5**) in good yields (eq. 2).



The reaction was carried out in the presence of BF_3 etherate³ at 40 °C for 1-3 days. The resulting β -ketoesters (**5**) have been used for the synthesis of many valuable compounds such as ketones,⁴ esters,⁵ and heterocyclic compounds.⁶ In order to indicate the usefulness of this reaction, γ,δ -unsaturated ketone (**6**) and α -methylene- γ,δ -unsaturated ester (**7**) were prepared from **5a** (Scheme 1).



Scheme 1

Table 1. Synthesis of α -Acyl- γ,δ -unsaturated Esters (5)

Borane (1)	Ketone (4) ^a	React. time (h)	Product (5) ^b	Yield, % ^c
		2 4		7 3
		2 4		7 0
		4 8		9 3
		7 2		7 4
		4 8		7 5
		2 4		8 0
		7 2		6 0

a. A mixture of (E)- and (Z)-isomers was used.

b. A mixture of syn- and anti-isomers.

c. Isolated yield based on the organoborane used.

Table 2. NMR, IR, and MS Data of α -Acyl- γ,δ -unsaturated Esters (5) and their Derivatives

Compounds	^1H NMR (CDCl_3 / TMS), (ppm)	IR, $(\text{C}=\text{O})(\text{cm}^{-1})$	M. S. (M^+)
5a	5.68-5.38 (m, 1H), 4.32-4.02 (m, 2H), 3.49-3.25 (m, 2H), 2.38 (t, 6.9 Hz, 2H), 2.22 (s, 1.5H), 2.19 (s, 1.5H), 1.71-0.71 (m, 13H)	1735, 1712	320.0805
5b	5.68-5.35 (m, 1H), 4.32-4.02 (m, 2H), 3.56-3.17 (m, 2H), 2.60-2.34 (m, 4H), 1.75-1.11 (m, 11H), 0.81 (t, $J = 6.2$ Hz, 9H)	1740, 1718	360.1303
5c	7.25 (s, 5H), 5.95-5.73 (m, 1H), 4.72-4.47 (m, 1H), 4.25-3.72 (m, 3H), 2.52-2.22 (m, 2H), 2.25 (s, 1.5H), 2.05 (s, 1.5H), 1.61-0.78 (m, 10 H)	1740	382.0987
5d	7.61-7.01 (m, 10H), 6.48-6.22 (m, 1H), 4.89-4.60 (m, 1H), 4.35-3.77 (m, 3H), 2.39-2.03 (m, 3H), 1.39-0.82 (m, 3H)	1720	402.0656
5e	7.22 (s, 5H), 5.97-5.41 (m, 2H), 5.10-4.80 (m, 2H), 4.73-4.39 (m, 1H), 4.28-3.74 (m, 3H), 2.58-2.26 (m, 4H), 2.25 (s, 1.5H), 2.05 (s, 1.5H), 1.24 (t, $J = 6.8$ Hz, 1.5 H), 1.00 (t, $J = 6.8$ Hz, 1.5 H)	1715	380.0827
5f	4.86 (t, $J = 6.8$ Hz, 1H), 4.29-3.91 (m, 2H), 3.31-3.11 (m, 2H), 2.20 (s, 1.5H), 2.13 (s, 1.5 H), 2.10-1.80 (m, 2H), 1.63 (s, 3H), 1.50-1.10 (m, 7H), 1.01-0.88 (m, 6H)	1718	254.1908
5g	5.67-5.00 (m, 2H), 4.35-3.94 (m, 2H), 3.28 (d, $J = 8.75$ Hz, 1H), 3.08-2.72 (m, 1H), 2.19 (s, 1.5H), 2.14 (s, 1.5H), 2.05-1.66 (m, 2H), 1.42-1.08 (m, 11H), 1.08-0.71 (m, 6H)	1745, 1718	268.2051
6	5.44 (d, $J = 8.75$ Hz, 1H), 3.27-2.88 (m, 1H), 2.47-2.33 (m, 4H), 2.14 (s, 3H), 1.67-1.17 (m, 4H), 1.03 (d, $J = 7.5$ Hz, 3H), 0.92 (t, $J = 7.5$ Hz, 3H)	1720	248.0596
7	6.11 (s, 1H), 5.71-5.47 (m, 2H), 4.19 (q, $J = 6.2$ Hz, 2H), 3.71 (m, 1H), 2.41 (t, $J = 7.5$ Hz, 2H), 1.70-1.11 (m, 10H), 0.81 (t, $J = 6.2$ Hz, 3H)	1720	288.0704

The reaction was found to proceed without any isomerization of the double bond and the stereochemistry of **1** was demonstrated to be kept in **5**. However, the syn- and anti- isomers were always formed in almost 1:1 ratio and the ratio is not dependent on the stereochemistry of **4** used. Even when a pure (E)- or (Z)-isomer of **4** was used, both isomers of **5** were formed in a same ratio. The representative results are shown in Table 1.

Experimental

Materials: (1-Octenyl)diisopropoxyborane,⁷ (2-bromo-1-alkenyl)diisopropoxyboranes,² and (E)-(2-methyl-1-hexenyl)diisopropoxyborane² were prepared according to the literatures. α -Acyl- α,β -unsaturated esters were prepared from the corresponding β -ketoesters and aldehydes according to the literature.⁸

General procedure for the preparation of α -acyl- γ,δ -unsaturated esters

A mixture of (Z)-(2-bromo-1-hexenyl)diisopropoxyborane (0.291 g, 1 mmol), ethyl 2-acetyl-2-butenate (0.156 g, 1 mmol), and BF₃ etherate (0.142 g, 1 mmol) in 5 mL of CH₂Cl₂ was stirred under reflux for 2 days. After extraction with ether, ethyl (Z)-2-acetyl-5-bromo-3-methyl-4-nonenate (**5a**) was isolated by preparative tlc (silicagel / benzene : hexane = 9:1) in 75 % yield (0.240 g).

Preparation of **6**

In an ethanolic solution of KOH (15 mL of 0.5 M solution), **5a** (0.319 g, 1 mmol) was stirred at room temperature and the reaction was monitored by tlc.

After 3 days, the consumption of **5a** was confirmed and the reaction mixture was acidified by addition of 3M HCl. After extraction with ether, **6** was isolated by preparative tlc (silica gel / benzene : hexane = 9:1) in 98 % yield (0.242 g).

Preparation of **7^{5a}**

A THF solution of LDA was prepared by addition of a hexane solution of BuLi (0.75 mL of 1.6 M solution, 1.2 mmol) to a THF solution (5 mL) of diisopropylamine (0.121 g, 1.2 mmol) at -78 °C. After 30 min, a THF solution (1 mL) of **5a** (0.319 g, 1 mmol) was added to give a clear yellow solution. The mixture was stirred at -78 °C for 1h and then paraldehyde (0.07 g, 2.3 mmol) was added. Then the cooling bath was removed and the mixture was stirred under reflux for 2h. After extraction with ether, **7** was isolated by preparative tlc (silica gel / benzene : hexane = 9:1) in 71 % yield (0.2 g).

References and notes:

1. Jacob, P. and Brown, H. C., *J. Am. Chem. Soc.*, **1976**, *98*, 7832;
Sato, Y., Serizawa, H., Hara, S., and Suzuki, A., *J. Am. Chem. Soc.*, **1985**, *107*, 5225.
2. Hara, S., Hyuga, S., Aoyama, M., Sato, M., and Suzuki, A., *Tetrahedron Lett.*, **1990**, *31*, 247.
3. BF₃ etherate is used to activate **1** by converting it to the more reactive fluoroborane derivative.²

4. Marvell, C. S. and Hager, F. D., *Organic Syntheses* Coll. Vol. 1, 1941, pp. 248-250; Johnson, J. R. and Hager, F. D., *ibid.*, Coll. Vol. 1, 1941 pp. 351-353.
5. (a). Ueno, Y., Setoi, H., and Okawara, M., *Tetrahedron Lett.*, 1978, 39, 3753. (b). Ritter, J. J. and Kaniecki, T. J., *J. Org. Chem.*, 1962, 27, 622.
6. Sethna, S. and Phadke, R., *Organic Reactions* Vol. VII, Ed., Adams, R. John Wiley & Sons, New York 1953, pp. 1-58.
7. Brown, H. C., Bhat, N. G., and Somayaji, V., *Organometallics*, 1983, 2, 1311.
8. Lehnert, W., *Tetrahedron*, 1972, 28, 663.

(Received in USA 29 April, 1992)