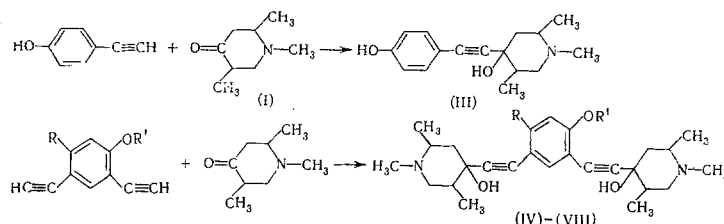


# PIPERIDOL DERIVATIVES OF ETHYNYLPHENOLS

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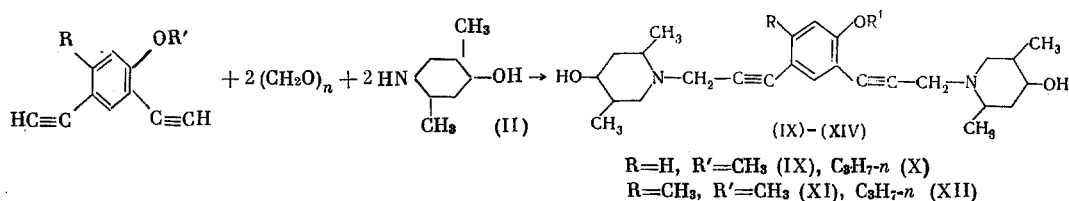
UDC 542.91+547.566

In order to vary the structure of compounds, tested for the presence of cholinergic and germistatic properties in ethynylphenols and their derivatives [1, 2], some ethynylphenols were condensed with 1,2,5-trimethyl-4-piperidone (I) by the Favorskii reaction, and with 2,5-dimethyl-4-piperidol (II) by the Mannich reaction. The Favorskii reaction proceeds initially at  $-5^{\circ}$ , and then at room temperature. An increase in the reaction temperature causes tarring, which is especially noticeable for p-ethynylphenol



R=H, R'=CH<sub>3</sub> (IV), C<sub>6</sub>H<sub>7-n</sub> (V)  
R=CH<sub>3</sub>, R'=CH<sub>3</sub> (VI), C<sub>6</sub>H<sub>7-n</sub> (VII)  
R=OC<sub>2</sub>H<sub>5</sub>, R'=C<sub>6</sub>H<sub>7-n</sub> (VIII)

The Mannich reaction is run in the presence of catalytic amounts of Cu<sub>2</sub>Cl<sub>2</sub> at  $50^{\circ}$ , in which connection the ethers of the ethynylphenols give higher yields of the end products than do the ethynylphenols



R=H, R'=CH<sub>3</sub> (IX), C<sub>6</sub>H<sub>7-n</sub> (X)  
R=CH<sub>3</sub>, R'=CH<sub>3</sub> (XI), C<sub>6</sub>H<sub>7-n</sub> (XII)

TABLE 1

Compound No.	Yield, %	Mp, °C	Found N, %	Empirical formula	Calculated N, %	Hydrochlorides			
						mp, °C	found Cl, %	empirical formula	Calculated Cl, %
III	36,4	196—197	5,53	C <sub>16</sub> H <sub>21</sub> NO <sub>2</sub>	5,40	230	12,17	C <sub>16</sub> H <sub>22</sub> NO <sub>2</sub> Cl	11,93
IV	57	Liquid	6,65	C <sub>27</sub> H <sub>39</sub> N <sub>2</sub> O <sub>3</sub>	6,39	195—197	13,79	C <sub>27</sub> H <sub>40</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	13,86
V	65,2	The same	5,90	C <sub>29</sub> H <sub>42</sub> N <sub>2</sub> O <sub>3</sub>	6,00	174—175	13,66	C <sub>29</sub> H <sub>44</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	13,14
VI	62,5	" "	6,24	C <sub>28</sub> H <sub>40</sub> N <sub>2</sub> O <sub>3</sub>	6,19	200	12,95	C <sub>28</sub> H <sub>42</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	13,49
VII	83	" "	5,80	C <sub>30</sub> H <sub>44</sub> N <sub>2</sub> O <sub>3</sub>	5,83	190—191	13,34	C <sub>30</sub> H <sub>46</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	12,81
VIII	52	211—212	5,57	C <sub>32</sub> H <sub>48</sub> N <sub>2</sub> O <sub>4</sub>	5,34	220—221	12,89	C <sub>32</sub> H <sub>50</sub> N <sub>2</sub> O <sub>4</sub> Cl <sub>2</sub>	11,87
			5,58				13,02		
							11,82		
							11,96		

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TABLE 2

Compound No.	Yield, %	Mp, °C	Found N, %	Empirical formula	Calculated N, %	Hydrochlorides			
						mp, °C	found Cl, %	empirical formula	calculated Cl, %
IX	76,7	Liquid	6,37	C <sub>27</sub> H <sub>33</sub> N <sub>2</sub> O <sub>3</sub>	6,39	179—180	13,82 13,74	C <sub>27</sub> H <sub>40</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	13,87
X	73,7	The same	6,13	C <sub>29</sub> H <sub>42</sub> N <sub>2</sub> O <sub>3</sub>	6,00	170	13,11 13,08	C <sub>29</sub> H <sub>44</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	13,14
XI	71	179—180	6,18	C <sub>28</sub> H <sub>40</sub> N <sub>2</sub> O <sub>3</sub>	6,19	215	13,28 13,29	C <sub>28</sub> H <sub>42</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	13,49
XII	86	55—57	6,12	C <sub>30</sub> H <sub>44</sub> N <sub>2</sub> O <sub>3</sub>	5,83	210	12,71 13,09	C <sub>30</sub> H <sub>46</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	12,81
XIII	72,5	Liquid	5,13	C <sub>35</sub> H <sub>50</sub> N <sub>2</sub> O <sub>3</sub>	4,84	140—142	10,8 11,13	C <sub>35</sub> H <sub>52</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>2</sub>	10,88

The obtained products are listed in Tables 1 and 2. Their IR spectra correspond to the structure. As a rule, the obtained products represent a mixing of spatial isomers, and for this reason have a broad melting point range. The starting piperidone and piperidol were obtained as described in [3-6]. The preliminary testing disclosed the presence of cholinergic properties for some of the described acetylenic aminophenols.

#### EXPERIMENTAL METHOD

Propyl Ether of 5-Methyl-2,4-di-(2',4',5'-trimethyl-1'-piperidolyl)ethynylphenol (VII). To a cooled to -5° solution of 1.98 g of the propyl ether of 5-methyl-2,4-diethynylphenol and 3 g of 1,2,5-trimethyl-4-piperidone (I) in anhydrous ether was added 3 g of freshly fused KOH powder. A stream of dry N<sub>2</sub> was steadily passed through the reaction mixture. The reaction was run at -5° for 3 h and at 20° for 10 h. The reaction course was followed by means of TLC on Al<sub>2</sub>O<sub>3</sub> (II activity), and elution was with the system: alcohol-ether (1:1). The ether solution was filtered from the KOH and the ether was removed. We isolated 4 g (83%) of (VII). In a similar manner were obtained (III)-(VI) and (VIII), which are listed in Table 1.

Methyl Ether of 2,4-Di-[3'-N-(2'',5''-dimethyl-4''-piperidol)-1'-propynyl]phenol (IX). To a mixture of 2.5 g of 2,5-dimethyl-4-piperidol (II) and 0.6 g of paraform in dioxane-methanol solution were added 1.2 g of the methyl ether of 2,4-diethynylphenol and 0.1 g of Cu<sub>2</sub>Cl<sub>2</sub>. The reaction was run in a stream of dry pure N<sub>2</sub> for 4 h at 50°. The mixture was decomposed with 30 ml of water, and then extracted with ether. The ether extract was passed through a bed of Al<sub>2</sub>O<sub>3</sub>, the ether was removed, and the residue was washed with low-boiling petroleum ether. We isolated 1.2 g of (IX).

In a similar manner were obtained (X)-(XII), which are listed in Table 2. The propionate (XIII) was obtained by refluxing a mixture of 5 g of (X), 10 ml of CH<sub>3</sub>CH<sub>2</sub>COCl and 0.1 g of Mg in 30 ml of anhydrous benzene for 5 h. After neutralization, the organic layer was passed through Al<sub>2</sub>O<sub>3</sub>, followed by elution with ether. After removal of the solvents we isolated 4.5 g of (XIII) as a viscous liquid. The hydrochlorides of all of the synthesized amines (III)-(XIII) were obtained by the passage of dry HCl through a solution of the amine in anhydrous ether.

#### CONCLUSION

A number of acetylenic aminophenols and their ethers, representing derivatives of 2,5-dimethyl-4-piperidol and 1,2,5-trimethyl-4-piperidone, was synthesized.

#### LITERATURE CITED

1. M. I. Bardamova, R. N. Myasnikova, and I. L. Kotlyarevskii, *Izv. AN SSSR, Ser. Khim.*, 443 (1967).
2. M. I. Bardamova, I. L. Kotlyarevskii, T. G. Shishmakova, and Z. P. Trotsenko, *Izv. AN SSSR, Ser. Khim.*, 658 (1967).
3. I. N. Nazarov, V. Ya. Raigorodskaya, and V. A. Rudenko, *Izv. AN SSSR, Otd. Khim. Nauk*, 504 (1949).
4. I. N. Nazarov, N. S. Prostakov, and N. N. Mikheeva, *Zh. Obshch. Khim.*, 28, 2431 (1958).
5. I. N. Nazarov, D. V. Sokolov, and V. N. Rakcheeva, *Izv. AN SSSR, Otd. Khim. Nauk*, 80 (1954).
6. I. N. Nazarov and E. T. Golovin, *Zh. Obshch. Khim.*, 26, 832 (1956).