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188 Jílek and Protiva: Some Amidines and Dihydroglyoxalines of

42. Some Amidines and Dihydroglyoxalines of the 1:1-Diphenylethane and 1:1-Diphenylpropane Series.

By J. O. JÍLEK and M. PROTIVA.

In view of the well-known similarity of the pharmacological properties between ethyldialkylamines and 2-methyldihydroglyoxaline derivatives 2-(2:2-diphenylethyl)-4:5-dihydroglyoxaline(VI) has been prepared as a glyoxaline analogue of "Aspasan" (V). Three similar amidines (VIII) and two lower homologues (IX and X) have also been prepared, one of them (VIII; R = H) exerting a relatively strong antihistaminic action.

DJERASSI and SCHOLZ (J. Amer. Chem. Soc., 1947, 69, 1688) noticed the analogy between the known antihistaminic agents "Antergan" (I) and "Antistin" (II). (II) differs from (I) only in that the 2-methyldihydroglyoxaline group replaces the dimethylaminoethyl moiety.

$$\begin{array}{c} CH_2Ph\cdot NPh\cdot CH_2\cdot CH_2\cdot NMe_2 \\ (I.) \\ \end{array} \qquad \qquad CH_2Ph\cdot NPh\cdot CH_2\cdot C \swarrow NH\cdot CH_2 \\ (II.) \\ \end{array}$$

Making use of this analogy, these authors prepared a series of aryloxyacetamidines and 2-aryloxymethyldihydroglyoxalines and found them to show antihistaminic activity similar to that of the corresponding aryloxyethyldialkylamines. On the basis of the same assumptions the dihydroglyoxaline analogue (IV) of "Benadryl" (III) was recently prepared almost simultaneously in different laboratories (Protiva and Urban, Coll. Trav. chim. Tchécosl., 1948, 13, 340; Cavallini and Mazzucchi, Farmaco, 1947, 2, 273; Dahlbom and Sjögren, Acta Chem. Scand., 1947, 1, 777; Djerassi and Scholz, J. Org. Chem., 1948, 13, 830) and was found to be quite similar to (III) in its pharmacological properties.

$$\begin{array}{c} CHPh_2 \cdot O \cdot CH_2 \cdot CH_2 \cdot NMe_2 \\ (III.) \\ \end{array} \qquad CHPh_2 \cdot O \cdot CH_2 \cdot C \ll \stackrel{N \longrightarrow CH_2}{NH \cdot CH_2} \\ (IV.) \\ \end{array}$$

Considering these results, it seemed desirable to prepare amidine and dihydroglyoxaline analogues of **3**: **3**-diphenylpropylamine derivatives, another pharmacologically interesting

group characterised by "Aspasan" (V), showing antispasmodic and antihistaminic activity (Bockmühl and Ehrhart, Annalen, 1948, 561, 52). Attention was paid first to the preparation

$$\begin{array}{c} CHPh_2 \cdot CH_2 \cdot CH_2 \cdot N < [CH_2]_5 \\ (V.) \\ (V.) \\ (VI.) \\ \end{array} \begin{array}{c} CHPh_2 \cdot CH_2 \cdot C < NH \cdot CH_2 \\ NH \cdot CH_2 \\ (VI.) \\ \end{array} \begin{array}{c} Ph_2 CH \cdot CH_2 \cdot CO \cdot NH \cdot CH_2 \cdot CH_2 \cdot NH_2 \\ (VII.) \\ (VII.) \\ \end{array}$$

of 2-(2:2-diphenylethyl)-4:5-dihydroglyoxaline (VI), the analogue of "Aspasan" (V). This compound was obtained from $\beta\beta$ -diphenylpropionitrile either through *ethyl* $\beta\beta$ -diphenylpropionimidate hydrochloride by the method of Klarer and Urech (Helv. Chim. Acta, 1944, 27, 1762), or by fusion with 2-aminoethylammonium toluene-p-sulphonate (Oxley and Short, J., 1947, 497). The first method yielded an impure hydrochloride, which by the action of ammonia gave the base (VI), m. p. 101°. From the base a crystalline hydrochloride (m. p. 88-90°) and *picrate* were prepared. By the second method, a crystalline toluene-p-sulphonate was obtained, which could be converted into the base and hydrochloride, identical with the products obtained by other route. The base, toluene-p-sulphonate, and picrate gave good analytical results; the hydrochloride in contrast, even when thoroughly dried in vacuo, gave results which indicated that it was a monohydrate. Since, however, it might have been $\beta\beta$ -diphenylpropiono-2-aminoethylamide hydrochloride (VII), which could be formed by opening the glyoxaline ring of (VI), the base (VII) was prepared from ethyl ßß-diphenylpropionate by the action of excess of ethylenediamine at 210-220° (Hill and Aspinall, J. Amer. Chem. Soc., 1939, 61, 822). The hydrochloride of (VII) melts at 214-215° and differs from that having m. p. $88-90^{\circ}$. The base (VII) is a solid of m. p. 118° .

Three new $\beta\beta$ -diphenylpropionamidines (VIII; $NR_2 = NH_2$, NMe_2 and $N < [CH_2]_5$) of a similar type were prepared by the action of an excess of ammonia or of the appropriate amine on ethyl ßß-diphenylpropionimidate hydrochloride. For comparison diphenylacetamidine (IX) and 2-benzhydryl-4: 5-dihydroglyoxaline (X) were also prepared by standard methods.

$$\begin{array}{c} \text{CHPh}_{2} \cdot \text{CH}_{2} \cdot \text{C} \swarrow_{\text{NR}_{2}}^{\text{NH}} & \text{CHPh}_{2} \cdot \text{C} \swarrow_{\text{NH}_{2}}^{\text{NH}} & \text{CHPh}_{2} \cdot \text{C} \swarrow_{\text{NH}}^{\text{N}-\text{CH}_{2}} \\ \text{(VIII.)} & \text{(IX.)} & \text{(X.)} \end{array}$$

The soluble derivatives of compounds (VI), (VIII), (IX), and (X) were tested for antihistamine activity. Only the hydrochloride of (VIII; R = H) was found to be approximately equal in activity to "Aspasan" (0.1 unit of "Benadryl" activity). Further testing for antispasmodic, analgesic, and pressor activity is in progress.

EXPERIMENTAL.

M.p.s are corrected. Analytical samples were dried at 0.2 mm. (P_2O_5) for 10 hours at suitable temperatures. Semimicro-analyses were by Miss Bruchalová, Ing. Mňouček, Dr. Králíčková, and Ing. Rýpar.

and fing. Rypar. $\beta\beta$ -Diphenylpropionitrile.—This was prepared from a-cyano- $\beta\beta$ -diphenylpropionic acid by the method of Kohler and Reimer (Amer. Chem. J., 1905, **33**, 338) and purified by distillation in vacuo (Freeman, Ringk, and Spoerri, J. Amer. Chem. Soc., 1947, **69**, 858) and by crystallisation from ethyl alcohol-light petroleum. Yield 72%. B. p. 175—178°/2 mm. M. p. 88-5°. Ethyl $\beta\beta$ -Diphenylpropionimidate Hydrochloride.—A solution of the above nitrile (11-7 g.) in a mixture of chloroform (60 c.c.) and absolute ethyl alcohol (7.5 c.c.) was saturated with dry hydrogen chloride at 0° and set aside for 10 days in a closed flask at 20°. After removal of the solvents at **30°** in remove o corrections of the product of m p. 128° in quantitative

in vacuo a crystalline residue remained, representing the crude product of m. p. 128°, in quantitative yield.

2-(2:2-Diphenylethyl)-4:5-dihydroglyoxaline (VI).-(a) The above imidate hydrochloride (8.8 g.), absolute ethylenediamine (1.9 g.), and absolute ethyl alcohol (50 c.c.) were refluxed for 7 hours in a slow stream of dry air in a bath at 100—115°. After cooling, the solution was filtered from ethylenediamine dihydrochloride [0.1 g.; m. p. 335—338° (decomp.)] which separated. The filtrate was concentrated dinydrochloride [0·1 g.; m. p. 335-338° (decomp.)] which separated. The filtrate was concentrated under reduced pressure, the oily residue dissolved in water (70 c.c.), and the solution filtered and treated with concentrated aqueous ammonia (7 c.c.). The base (4·4 g.) was filtered off and recrystallised from 50% ethyl alcohol. It had m. p. 84-86°, or, after drying (P₂O₅ in vacuo), m. p. 101° (Found : C, 81·2; H, 7·5. C₁₇H₁₈N₂ requires C, 81·6; H, 7·2%). In another similar experiment the crude oily hydro-chloride was dissolved in acetone (10 c.c.) and the solution treated with absolute ether (10 c.c.). The oil which separated crystallised (7·5 g.). Crystallisation from acetone raised the m. p. of the hydrate to 88-90° (Found : N, 9·2; Cl, 11·8. C₁₇H₁₉N₂Cl,H₂O requires N, 9·2; Cl, 11·6%). The picrate crystallised from ethyl alcohol and melted at 168-170° (Found : C, 57·8; H, 4·9. C₂₃H₂₁O₇N₅ requires C, 57·6; H, 4·4%). (b) A mixture of 88-diphenvlpropionitrile (35·0 g) and 2-aminoethylammonium toluene-to-sulphonete

(b) A mixture of $\beta\beta$ -diphenylpropionitrile (35.0 g.) and 2-aminoethylammonium toluene-*p*-sulphonate (39.2 g.) (m. p. 122—124°; Oxley and Short, *loc. cit.*) was heated for 3 hours at 200°. After cooling, the solidified mass was recrystallised from aqueous ethyl alcohol. The yield of *toluene-p-sulphonate*, m. p. 170—174°, was 92.5% (Found : C, 68.5; H, 6.5; N, 6.7. C₂₄H₂₆O₃N₂S requires C, 68.2; H, 6.2; N, 6.6%). A suspension of this salt (15.0 g.) in hot water was treated with 20% sodium hydroxide

(30 c.c.) and extracted with chloroform. After evaporation of the solvent the residue crystallised (7.8 g.). After recrystallisation from acetone, the m. p. was 102° (Found : N, 11.3%). This base (2.0 g.) in acetone (2 c.c.) was treated with 23% alcoholic hydrogen chloride solution (1.2 c.c.).

(2.0 g.) in acetone (2 c.c.) was treated with 23% alcoholic hydrogen chloride solution (1.2 c.c.). Recrystallisation of the crude product from acetone gave the hydrated hydrochloride, m. p. 88-90° (Found: C, 66.9; H, 7.0; N, 9.1%), whence the base, m. p. 100.5°, could be regenerated. *Ethyl* $\beta\beta$ -*Diphenylpropionate.*—This was prepared from the nitrile by application of Spiegel's method (*Ber.*, 1918, **51**, 296). Diphenylpropionitrile (8.0 g.), absolute ethyl alcohol (7 c.c.), and concentrated sulphuric acid (2.0 c.c.) were refluxed for 3 hours at 120—130° (bath). Ethyl alcohol was distilled off *in vacuo* and the residue mixed with water (100 c.c.). The oil which separated was extracted with ether, and the solution was dried and evaporated. The residue distilled from a Hickman flask at 138—141°(0.25 mm. (yield 5.5 g.). Wislicenus and Eble (*Ber.*, 1917, **50**, 253) gave b. p. 138-141°/0.25 mm. (yield, 5.5 g.). Wislicenus and Eble (Ber., 1917, 50, 253) gave b. p. 190-193°/12 mm.

 $\beta\beta$ -Diphenylpropiono-2-aminoethylamide (VII).—The above ester (5.5 g.) and absolute ethylenedi-amine (6.0 c.c.) were heated in a sealed tube for 18 hours at 210—230°. After cooling, the excess of diamine was distilled off and the residual oil transformed (in ethyl alcohol solution) into the hydromainine was distinct on and the residual of transformed (in early alcohol solution) into the *Myaro-*chloride. Crystallisation from ethyl alcohol gave 2-0 g. of the product, m. p. 214—215° (Found : N, 9·5; Cl, 11·7. $C_{17}H_{21}ON_2Cl$ requires N, 9·2; Cl, 11·6%). From aqueous solution of this the base was obtained by treatment with 40% aqueous sodium hydroxide. Recrystallisation from benzene gave white prisms, m. p. 118° (Found : N, 10·6. $C_{17}H_{20}ON_2$ requires N, 10·4%). $\beta\beta$ -Diphenylpropionamidine (VIII; R = H).—To a suspension of the crude imidate hydrochloride (prepared from 4·5 g. of nitrile) in absolute ethyl alcohol (10 c.c.), 8% absolute ethyl-alcoholic ammonia (100 c.c.) was added and the mixture was shaken in a closed flask for 3 hours at 20°. After 12 hours, babing was repeated for a further 2 hours.

(100 c.c.) was added and the hirkture was shaken in a closed mask for 5 hours at 20. After 12 hours, shaking was repeated for a further 2 hours. Annonium chloride, which separated, was filtered off and the filtrate evaporated *in vacuo* at $30-35^{\circ}$ (bath). The residue was dissolved in chloroform (30 c.c.), the solution washed with water, 10% sodium hydroxide solution, and water, and the chloroform was distilled off. The oily residue (5·2 g.) is the crude base (VIII; R = H). The *picrate*, after recrystallisation from absolute ethyl alcohol, melted at $209-209\cdot5^{\circ}$ (Found : C, 55·8; H, 4·0; N, 15·5.

 $C_{21}H_{19}O_2N_5$ requires C, 55.6; H, 4.2; N, 15.5%). NN-Dimethyl- $\beta\beta$ -diphenylpropionamidine (VIII; R = Me).—The crude imidate hydrochloride (4.5 g.) was dissolved in ethyl alcohol (20 c.c.), 18% ethyl-alcoholic dimethylamine solution (4.5 c.c.) was added and the mixture was set aside for 24 hours at room temperature. After evaporation of alcohol the residue was recrystallised from ethyl alcohol-ether. 3.2 G. of *hydrochloride*, m. p. 249°, were obtained (Found : C, 70.5; H, 7.3; N, 9.5. $C_{17}H_{21}N_2Cl$ requires C, 70.7; H, 7.3; N, 9.7%). NN-Pentamethylene- $\beta\beta$ -diphenylpropionamidine (VIII; NR₂ = N<[CH₂]₅).—To the solution of

crude imidate hydrochloride (6.2 g.) in absolute ethyl alcohol (30 c.c.) piperidine (2.0 g.) was added and the mixture set aside for 12 hours at room temperature. It was then evaporated in vacuo to dryness and gave, as above, 4.2 g. of hydrochloride, m. p. 250° (Found : C, 73.1; H, 7.6. C20H25N2Cl requires

C, 73.0; H, 7.7%). Ethyl Diphenylacetimidate Hydrochloride.—A solution of diphenylacetonitrile (6.6 g.), m. p. 74°, prepared by dehydration of diphenylacetamide (Neure, Annalen, 1889, 250, 142), in a mixture of chloroform (30 c.c.) and absolute ethyl alcohol (5.0 c.c.) was saturated with hydrogen chloride at 0° and set aside for 14 days at room temperature in a closed flask. After evaporation of the solvents a crude product suitable for the preparation of the amidine was obtained.

Diphenylacetamidine (IX).—(a) A suspension of the above imidate hydrochloride in absolute ethyl alcohol (20 c.c.) was mixed with 8% ethyl-alcoholic ammonia (18.5 c.c.), and the mixture shaken for 2 hours alcohol (20 c.c.) was mixed with 8% ethyl-alcoholic ammonia (18.5 c.c.), and the mixture shared to 1 - note at 20°. After 12 hours, shaking was repeated for an hour. Ammonium chloride was filtered off and the filtrate evaporated under reduced pressure. The remaining oil was dissolved in chloroform (50 c.c.), the mixture shared with 5N-sodium hydroxide and water, and the chloroform evaporated. The the solution washed with 5N-sodium hydroxide and water, and the chloroform evaporated. The residue is the crude amidine, m. p. $95-97^{\circ}$ (7·1 g.). It gave a *picrate* which after recrystallisation from ethyl alcohol melted at $224-225^{\circ}$ (decomp.) (Found : C, 54·8; H, 3·6. C₂₀H₁₇O₇N₅ requires C, 54.7; H, 3.9%). The hydrochloride could not be obtained crystalline. (b) A mixture of diphenylacetonitrile (2.5 g.) and dried ammonium thiocyanate (4.0 g.) was heated

for 5 hours at 180°. After cooling, the product was treated with hot water (10 c.c.), and the solution filtered. The filtrate was made alkaline with 5N-sodium hydroxide (10 c.c.) and extracted with chloroform. After evaporation of the solvent the residue gave a picrate (0.4 g.), m. p. 224-225°, identical with the above product.

2-Benzhydryl-4: 5-dihydroglyoxaline (X).—A mixture of diphenylacetonitrile (9.6 g.) and 2-amino-ethylammonium toluene-p-sulphonate (11.6 g.) was heated for 3 hours at 200—210°, during which ammonia was evolved. The cooled mass was extracted with hot water (500 ml.), the mixture made alkaline with 5N-sodium hydroxide (50 c.c.) and extracted with chloroform (200 c.c.). After drying of the solution the chloroform was evaporated off. The residue (61%) is the base, which after recrystallisation from acetone melts at $133-135^{\circ}$ (Aspinall, J. Amer. Chem. Soc., 1939, **61**, 3195, gives 137°). From the base were prepared the picrate, crystallising from ethyl alcohol and melting at 184.5° (Aspinall, *loc. cit.*, gives 185°) (Found : C, 56.6; H, 4.4; N, 15.6. Calc. for $C_{22}H_{19}O_7N_5$: C, 56.8; H, 4-1; N, 15.1%), and hydrochloride, crystallising from ethyl alcohol-ether and melting at 180—182° (Ciba, Austrian P, 150,307; Zentr., 1937, II, 3039, gives 192—193°) (Found : N, 10.0; Cl, 12.5. Calc. for $C_{16}H_{17}N_2Cl$: N, 10.3; Cl, 13.0%).

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