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Organic Preparations and Procedures International: The New Journal for Organic Synthesis

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/uopp20

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To cite this article: Sachin S. Patil , Sachin V. Patil & Vivek D. Bobade (2011) An Efficient Synthesis of Zolpidem, Organic Preparations and Procedures International: The New Journal for Organic Synthesis, 43:2, 260-264, DOI: <u>10.1080/00304948.2011.564558</u>

To link to this article: <u>http://dx.doi.org/10.1080/00304948.2011.564558</u>

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An Efficient Synthesis of Zolpidem

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Imidazo[1,2-a]pyridines (IPs) have received considerable attention because of their interesting therapeutic properties,¹ including antibacterial,² antifungal,^{3,4} antiviral,^{5,6} antiulcer,⁷ and anti-inflammatory behavior.⁸ They have also been identified as selective cyclin-dependent kinase inhibitors,⁹ calcium channel blockers,¹⁰ β -amyloidal formation inhibitors,¹¹ and benzodiazepine receptor agonists,¹² and constitute a novel class of orally active nonpeptidic bradykinin B₂ receptor antagonists.¹³ Drug formulations containing imidazo[1,2a]pyridines such as *alpidem* (anxolytic), *zolpidem* (hypnotic), and *zolimidine* (antiulcer) are currently marketed.



Zolpidem (7) is a prescription medication used for the short-term treatment of insomnia, as well as some brain disorders. It is a short-acting non-benzodiazepine hypnotic that potentiates γ -aminobutyric acid (GABA), an inhibitory neurotransmitter, by binding to GABA_A receptors at the same location as benzodiazepines.¹⁴ Although its hypnotic effects are similar to those drugs of the benzodiazepine class, it is molecularly distinct from benzodiazepines and is classified as an imidazopyridine.

One reported synthesis^{15,16} of zolpidem proceeds in overall 19% yield from 2-bromo-4'-methylacetophenone (1) as depicted in *Scheme 1*.

Submitted February 26, 2010.

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(a) 2-Amino-5-methylpyridine, EtOH, NaHCO₃, Δ. (b) Acetic acid, aq. Dimethylamine, formalin, RT (c) Acetone, CH₃I, Δ (d) NaCN, EtOH, Δ (e) KOH, EtOH /Δ (f) Carbonyldiimidazole, THF, dimethylamine / RT

Scheme 1

In a related procedure,¹⁷ an alternative preparation of **5** involves a Vilsmeier-Haack formylation of **2** followed by sodium borohydride reduction of the aldehyde to the alcohol. Tosylation of the alcohol and subsequent treatment with cyanide ion gave **5** 30% yield from **2**.

Another literature method^{18,19} involved bromination of *N*,*N*-dimethyl-4-oxo-4-tolylbutanamide (**8**) to 3-bromo-*N*,*N*-dimethyl-4-oxo-4-tolylbutanamide (**9**) followed by condensation of resultant bromo derivative **9** with 2-amino-5-methylpyridine to give zolpidem (**7**) in 17% yield (*Scheme 2*).



(a) Bromine, acetic acid (b) 2-Amino-5-methylpyridine

Scheme 2

However, these procedures suffer from several disadvantages from a commercial production standpoint such as (*a*) difficulty in handling the lachrymatory bromo compounds **1** and **9** on a large scale which requires precaution to avoid exposure, (*b*) multistep synthesis for compound **6** as illustrated in *Scheme 1* (3 steps from compound **3**), (*c*) isolation (drying operations) of intermediates such as compounds **1**, **4** and **5** and the work-up procedures which lead to increased cycle time and hence increase in the manufacturing cost, (*d*) the work-up procedure of the cyanation step is hazardous and requires safety measures as it may contain residual cyanide (*e*) in the hydrolysis step (**5** to **6**), compound **2** is formed as by-product (*f*) the use of carbonyldiimidazole (CDI) is expensive and is a highly moisture sensitive reagent and the reaction does not proceed to completion (*g*) the low overall yields in these procedures make them less viable for commercial production.



Method A: (i) HTIB, CH₃CN, reflux 2 h; (ii) 2-Amino-5-methylpyridine, reflux

Method B: (i) (PhIO) + p-TsOH/CH₃CN, reflux 2 h; (ii) 2-Amino-5-methypyridine, reflux

Scheme 3

We herein report a novel synthesis of zolpidem (7) by condensation of α -tosyloxyketone **10**,²⁰ prepared *in situ* by reaction of *N*,*N*-dimethyl-4-oxo-4-tolylbutanamide **8** and [hydroxy(tosyloxy)iodo]benzene (HTIB), with 2-amino-5-methylpyridine to give zolpidem (*Method A*). [Hydroxy(tosyloxy)iodo]benzene (HTIB), known as Koser's reagent has been gaining increasing popularity as versatile reagent for the preparation of α -tosyloxy ketones from enolizable ketones.^{21–23} We also established that it is possible to use a combination of iodosobenzene and *p*-toluenesulfonic acid [(PhIO)n + p-TsOH] in place of HTIB. This combination reagent generates HTIB *in situ*, which then reacts with ketone **8** to give the intermediary α -tosyloxyketone **10** (*Method B*) which further reacted *in situ* with 2-amino-5-methylpyridine to give zolpidem; however, the overall yield in this method was less than that of *Method A*.

In conclusion this investigation describes a simple and efficient one-pot procedure for the synthesis of zolpidem under mild conditions.

Experimental Section

Melting points are uncorrected and were determined on a Buchi Melting Point apparatus. ¹H NMR spectra were obtained in CDCl₃ using TMS as internal standard on Brucker spectrometer at 300MHz. Mass spectra were recorded on Shimadzu LCMS-QP8000, LC-MS and AB-4000 Q-trap LC-MS/MS. The elemental analysis was performed on Perkin-Elmer model 2400 CHNS/O analyzer. The reaction was monitored by TLC on silica gel plates.

Procedure for the Synthesis of Zolpidem

Method A

A solution of *N*,*N*-dimethyl-4-oxo-4-tolylbutanamide (**8**, 2.19 g, 1.0 mmol)²⁴ and hydroxy(tosyloxy)iodo]benzene (HTIB) (3.98 g, 1.0 mmol) in acetonitrile (25 mL) was refluxed for 2 hrs. After the completion of reaction, as monitored on TLC (EtOAc:hexane, 80:20), 2-amino-5-methylpyridine (1.1 g, 1.0 mmol) was added to the reaction mixture which was further refluxed for 2h. After completion of reaction as indicated by TLC, (solvent system CHCl₃:methanol 9:1) acetonitrile was distilled off. The residue was dissolved in dichloromethane (25 mL) and washed with a 5% aqueous sodium carbonate (20 mL) and water (20 mL). The organic phase was dried over MgSO₄, concentrated *in vacuo* and the solid residue was crystallized from toluene, dried at 50°C under vacuum, to yield zolpidem as a yellow solid (1.84 g, 60%) mp.192–194°C, *lit.*¹⁵ mp. 194–195°C.

Method B

A solution of iodosobenzene (2.20 g, 10 mmol), *p*-toluenesulfonic acid (1.72 g, 10 mmol) in acetonitrile (20 mL) was stirred at room temperature for 5 min and N,N-diethyl-4-oxo-4-tolylbutanamide (2.19 g, 1.0 mmol) was added and the mixture was refluxed for 2h. To this solution, 2-amino-5-methylpyridine (1.1g, 1.0 mmol) was added and the resulting mixture was further refluxed for 2 hrs. The product was isolated (1.44 g, 47%) as described in the procedure given in *Method A*.

¹H NMR (CDCl₃): δ 2.25 (s, 3H, ArCH₃), 2.34 (s, 3H, ArCH₃), 2.85 (s, 3H, NCH₃), 2.94 (s, 3H, NCH₃), 4.16 (s, 2H, COCH₂), 7.05 (d, 2H, ArH), 7.43–7.66 (m, 4H, ArH), 8.11 (s, 1H, ArH); ¹³C NMR (CDCl₃): δ 17.92, 20.79, 29.31, 35.34, 37.03, 113.79, 115.37, 121.82, 130.26, 137.27, 142.11, 142.81, 167.63. MS: M⁺ 308, 235, 219, 92

Anal. Calcd. For C₁₉H₂₁N₃O: C, 74.25; H, 6.88; N, 13.67; Found: C, 74.10; H, 6.74; N, 13.79.

Acknowledgement

The authors thank the University Grants Commission, New Delhi for financial assistance. They also thank IIT Mumbai Department of Chemistry for providing spectral data analyses. The author is also grateful to Mr. Omkar Herlekar, Director of Omkar Chemical Pvt. Ltd, Ambernath, for providing gift sample of diacetoxyiodobenzene (DIB) and iodosobenzene.

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