

## ASYMMETRIC SYNTHESIS OF $\beta$ -LACTAMS BY CHIRAL ESTER ENOLATE - IMINE CONDENSATION

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**Summary:** Asymmetric cyclocondensation of *N,N*-bis(silyl)glycinates bearing chiral ester moieties with aldimines is found to proceed with extremely high enantioselectivity to give the corresponding chiral  $\beta$ -lactams in good to high yields. It is found that the *E/Z*-geometry of the chiral ester-enolate is responsible for *cis/trans* stereochemistry of the  $\beta$ -lactam formed. Effects of various chiral auxiliaries in the ester-enolates on the enantioselectivity of the reactions are examined.

We have been applying our " $\beta$ -Lactam Synthon Method" to the asymmetric synthesis of various non-protein amino acids and dipeptides containing non-protein amino acid residues,<sup>1</sup> which are potential enzyme inhibitors, fragments of peptide hormone analogues, and components of naturally occurring glycosphingolipids and antibiotics.<sup>2</sup> For the asymmetric synthesis of nonprotein amino acids through the " $\beta$ -Lactam Synthon Method" the development of newer and efficient routes to enantiomerically pure  $\beta$ -lactams are of particular importance. We describe here our preliminary results on the successful application of lithium chiral ester enolate - imine condensation strategy, which has been extensively studied by Hart et al.,<sup>3</sup> to the asymmetric synthesis of 3-amino- $\beta$ -lactams.

We carried out the reactions of chiral lithium ester enolates (4) generated *in situ* from *N,N*-bis(silyl)glycinates (3) with imines (1 and 2), which gave the corresponding chiral  $\beta$ -lactams (5-11) in fairly good isolated yields (eq. 1).<sup>4</sup> Results are summarized in Table 1.

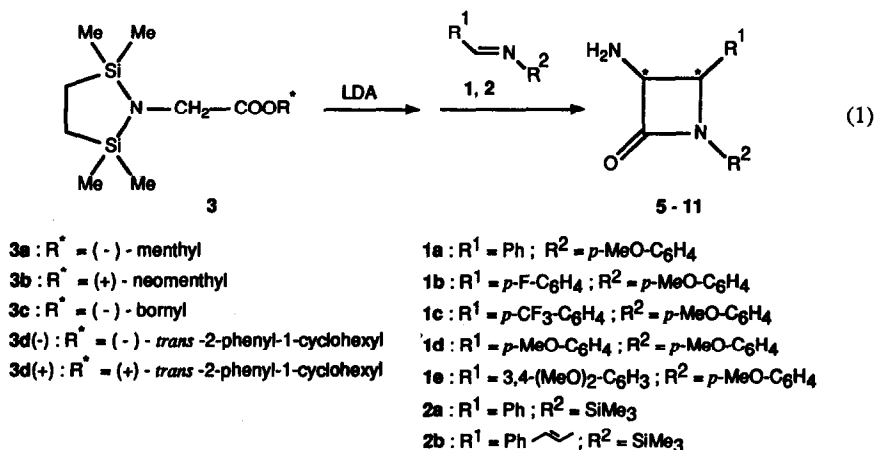


Table 1. Asymmetric Synthesis of  $\beta$ -Lactams (5-11) through Chiral Ester Enolate - Imine Condensation<sup>a</sup>

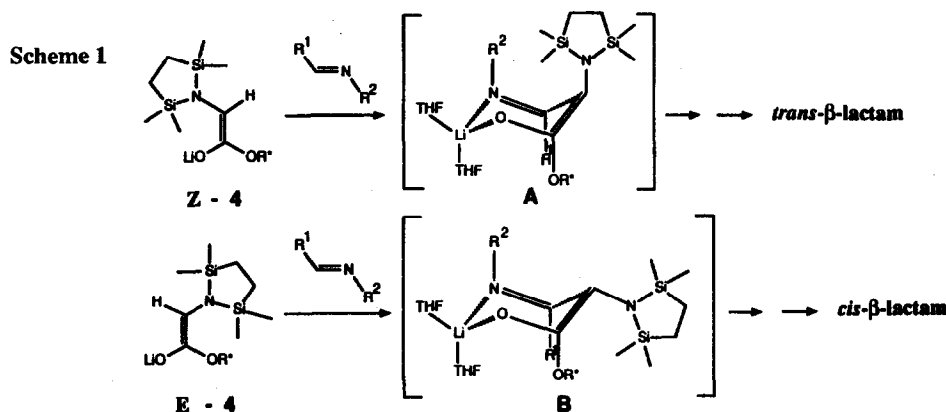
Entry	Ester	Imine	Conditions <sup>b</sup>	$\beta$ -Lactam	Isolated Yield (%) <sup>c</sup>	Product Ratio and Enantioselectivity <sup>d</sup>	
						trans (%)	cis (%)
1	3a	1a	A	5a	65	100 (>99% e.e.; 3 <i>R</i> ,4 <i>R</i> )	
2	3b	1a	A	5a 5b	65	26 (54% e.e.; 3 <i>S</i> ,4 <i>S</i> )	74 (21% e.e.; 3 <i>S</i> ,4 <i>R</i> )
3	3c	1a	A	5a 5b	53	37 (5% e.e.; 3 <i>R</i> ,4 <i>R</i> )	63 (2% e.e.; 3 <i>R</i> ,4 <i>S</i> )
4	3d(-)	1a	A	5a	58	100 (>99% e.e.; 3 <i>R</i> ,4 <i>R</i> )	
5	3d(+)	1a	A	5a	58	100 (>99% e.e.; 3 <i>S</i> ,4 <i>S</i> )	
6	3a	2a	B	6	38	100 (68% e.e.; 3 <i>R</i> ,4 <i>R</i> )	
7	3a	2b	B	7	48		100 (11% e.e.; 3 <i>S</i> ,4 <i>R</i> )
8	3d(-)	2b	B	7	46		100 (78% e.e.; 3 <i>S</i> ,4 <i>R</i> )
9	3a	1b	A	8	55	100 (>99% e.e.; 3 <i>R</i> ,4 <i>R</i> )	
10	3a	1c	A	9	59	100 (>99% e.e.; 3 <i>R</i> ,4 <i>R</i> )	
11	3a	1d	A	10a 10b	70	89 (>99% e.e.; 3 <i>R</i> ,4 <i>R</i> )	11 (38% e.e.; 3 <i>S</i> ,4 <i>R</i> )
12	3a	1e	A	11a 11b	54	91 (>99% e.e.; 3 <i>R</i> ,4 <i>R</i> )	9 (27% e.e.; 3 <i>S</i> ,4 <i>R</i> )

<sup>a</sup>All reactions were run with 2.0 mmole of 3, 2.20 mmole of LDA, and 2.0 mmole of 1 or 2 in 6 ml of THF. After the mixture was stirred at -78°C and/or -50°C for a given period of time, the reaction mixture was allowed to warm gradually to ambient temperature. The reaction was quenched with 0.1*N* HCl. <sup>b</sup>Condition A, at -78°C for 4h; Condition B, at -78°C for 4 h and at -50°C for 72 h. <sup>c</sup>Yield of  $\beta$ -lactam(s) after passing crude reaction mixture through a short column (15 cm, 10 g of silica gel) to eliminate unreacted starting materials, i.e., chiral auxiliary, aldehyde, and amine. <sup>d</sup>Enantiomeric purity was determined by the Mosher's MTPA method<sup>5</sup> on <sup>1</sup>H NMR and/or <sup>19</sup>F NMR. Absolute configurations were determined based on chemical correlation (specific rotation) with authentic samples (for 4-aryl- $\beta$ -lactams, their conversion to the corresponding  $\alpha$ -amino acid amides by hydrogenolysis<sup>1</sup> was used), and also based on the NMR chemical shift correlation of 3-MTPA-amino- $\beta$ -lactams.

As Table 1 shows, the reactions of 3a (*R*\* = (-)-menthyl) and 3d (*R*\* = (-)- or (+)-*trans*-2-phenyl-1-cyclohexyl<sup>6</sup>) with 1a-c give exclusively the corresponding *trans*- $\beta$ -lactams (5a, 8, and 9) in fairly good yields with extremely high enantiomeric purity (entries, 1, 4, 5, 9 and 10). The reactions of 3a with 1d and 1e also give *trans*- $\beta$ -lactams (10a and 11a) as the predominant products with >99% e.e. accompanied by a small amount of *cis*- $\beta$ -lactams (10b and 11b) (Entries 11 and 12). When (+)-neomenthyl group is used as the chiral auxiliary, the reaction gives a 1:3 mixture of *trans*:*cis* isomers (5a and 5b) with *S* configuration at the C-3 positions (Entry 2), which is opposite to that of 5a obtained by using (-)-menthyl group as the chiral auxiliary (Entry 1). A mixture of *trans*- and *cis*- $\beta$ -lactams (5a and 5b) is also obtained on using (+)-bornyl group as the chiral auxiliary, in which 3*R*-isomers are formed with very low enantiomeric purity (Entry 3). Accordingly, it is obvious that the use of (-)- and (+)-*trans*-2-phenyl-1-cyclohexyl as well as (-)-menthyl groups as the chiral auxiliaries is the most efficient. The substituent on the imine nitrogen also exerts a marked influence on

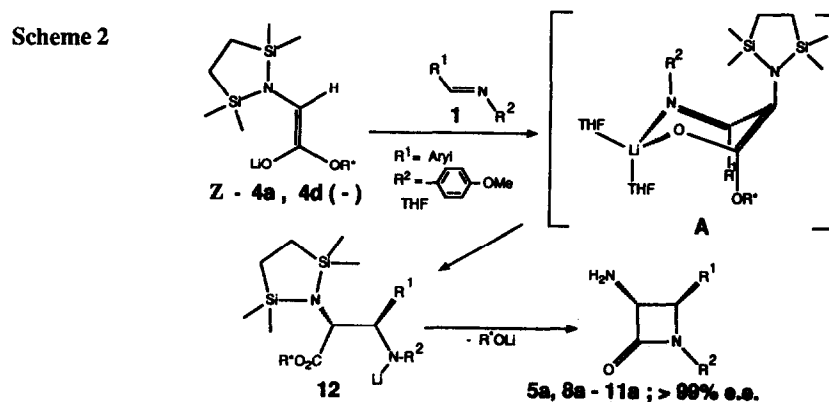
asymmetric induction, viz., 4-methoxyphenyl gives >99% e.e. for **5a** (Entry 1) whereas trimethylsilyl (TMS) induces only 68% e.e. for **6a** (Entry 6). In contrast with the cases of arylimines (**1**), the reactions of **3a** and **3d** with *N*-silylcinnamaldimine (**2b**) give *cis*- $\beta$ -lactam (**7**; *3S,4R*) exclusively (Entries 7 and 8), and **3d** gives much higher asymmetric induction (78% e.e.) than **3a**.

The formation of *trans*- and *cis*- $\beta$ -lactams can be explained by taking into account the stereochemistry of the lithium enolates (*Z*- and *E*-**4**) and transition states, **A** and **B**, as shown in Scheme 1.<sup>3,4c,7</sup>

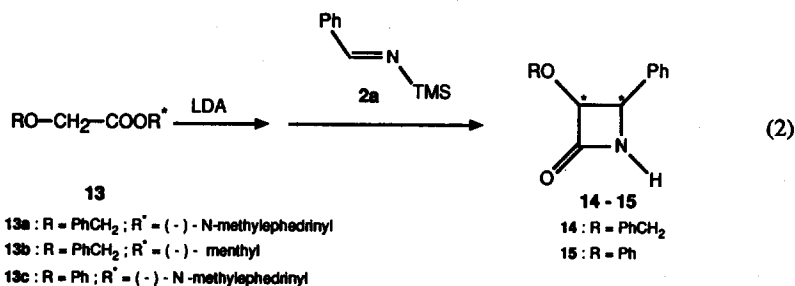


As Scheme 1 shows, the reaction of *Z*-enolate proceeds through the transition state **A** to give *trans*- $\beta$ -lactams, while the reaction of *E*-enolate gives *cis*- $\beta$ -lactams via the transition state **B**.<sup>3,4c,7</sup> When (-)-menthyl and (-)- or (+)-2-phenyl-1-cyclohexyl groups are used as the chiral auxiliaries, *trans*- $\beta$ -lactams are formed predominantly or exclusively, which strongly suggests that *Z*-enolates (**Z-4a,d**) react much faster than *E*-enolates (**E-4a,d**) and thus the reactions proceed through the transition state **A**. It is also indicated that the imines (**1a-e**) approach exclusively from the *re*-face of **Z-4a** and **Z-4d**(-) (*si*-face of **Z-4d**(+)) to give *N*-lithiated- $\beta$ -amino esters (**12a**), which then cyclize to afford the corresponding *trans*- $\beta$ -lactams (**5a**, **8a-11a**) with >99% e.e. (Scheme 2).

On the other hand, the fact that the reactions of **4a** and **4d**(-) with **2b** ( $R^1 = \textit{trans}$ -styryl,  $R^2 = \text{TMS}$ ) exclusively give *cis*- $\beta$ -lactam **7b**, strongly suggests that these reactions proceed through the transition state **B**, i.e., the *E*-enolates (**E-4a,d**) react with **2b** exclusively.



We also looked at the reactions of the lithium enolates of (-)-N-methylephedrinyl and (-)-menthyl O-protected hydroxyacetates (**13a-c**) with **2a** (eq. 2).<sup>8</sup> The reactions gave exclusively the corresponding *cis*- $\beta$ -lactams, **14** and **15**: **14** from **13a**, 20% yield, 67% e.e. (3*R*,4*S*); **14** from **13b**, 15% yield, 18% e.e. (3*S*,4*R*); **15** from **13c**, 25%, 20% e.e. (3*R*,4*S*).



Further studies on the mechanisms and applications of the chiral enolate - imine condensation are actively in progress.

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- The reactions were carried out in a manner similar to those shown in Table 1 except the fact that the reaction mixture was stirred at -78°C for 4 h and then at -50°C for 48 h.