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A New β -Arylethylamine Synthesis by Aryl Aldehyde Homologation

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Summary β -Arylethylamines can be generated in good yield by condensation of an aryl aldehyde with methoxyacetonitrile followed by demethylation of the α -methylcinnamonitrile product with sodium benzylthiolate in the presence of an amine and subsequent reduction with lithium aluminium hydride.

As part of our work on alkaloid biogenesis, we required a versatile and efficient method for the generation of monoand di- β -arylethylamines from the available vanillin/isovanillin units. We now report a process which offers advantages over the commonly used cyanide^{1,2} and nitrostyrene³ routes. The method is exemplified by the synthesis of the compound (6) (Scheme).

O-Benzylisovanillin was treated with an equivalent of the sodium salt of methoxyacetonitrile in excess of methoxyacetonitrile and dimethylformamide (N_2) at 110° to give α -methoxycinnamonitrile[†] (1), m.p. 85–87° (81%). The enol ether function was demethylated with sodium benzylthiolate (1 equiv.) in dimethylformamide under N₂ at 110°, during 5 min, benzylamine being added concurrently with the thiol. The presumed thiol ester intermediate (3) [derived from the aroyl cyanide (2)] and/or the cyanide (2)were captured by the added benzylamine (1 equiv.) to give the benzylamide (4) (95%). This was reduced to the amine (75%) with lithium aluminium hydride in tetrahydrofuran under reflux. This secondary amine was then utilized as the nucleophilic component in a second acylation step (as above) to form the tertiary amide (6) (80%). Reduction and debenzylation of this as previously described⁴ gave the diphenethylamine (7). The overall yield from O-benzylisovanillin was 27%.

In contrast to previous syntheses, the two aralkyl units are derived from a common intermediate with a consequent

† All new compounds gave correct microanalysis and the expected spectral data.

¹ K. Kindler and W. Peschke, Arch. Pharm., 1932, 270, 410.

² M. H. Tsao, J. Amer. Chem. Soc., 1951, 73, 5495.

 ³ F. Benington and R. D. Morin, J. Amer. Chem. Soc., 1951, 73, 1353.
 ⁴ D. H. R. Barton, R. James, G. W. Kirby, D. W. Turner, and D. A. Widdowson, J. Chem. Soc. (C), 1968, 1529.
 ⁵ E. Aufderhaar, J. E. Baldwin, D. H. R. Barton, D. J. Faulkner, and M. Slaytor, J. Chem. Soc. (C), 1971, 2175; see especially G. Stork and L. Maldonado, J. Amer. Chem. Soc., 1971, 93, 5286.

diminution in the number of steps involved. The yields are comparable.4

$$Ar^{1}CHO + MeOCH_{2}CN$$
 (i) $Ar^{1}CH = C(OMe)CN$
(1)



$$\xrightarrow{(iv)} \operatorname{Ar}^{1} \operatorname{CH}_{2} \operatorname{CH}_{2} \operatorname{NHCH}_{2} \operatorname{Ph} \xrightarrow{(v)} \operatorname{Ar}^{1} \operatorname{CH}_{2} \operatorname{CH}_{2} \operatorname{N(CH}_{2} \operatorname{Ph}) \operatorname{COCH}_{2} \operatorname{Ar}^{1}$$
(5)
(6)



(i) NaH-DMF; (ii) BzS-Na+; (iii) PhCH₂NH₂; Scheme. (iv) LiAlH_4 ; (v) (1) + BzS⁻ Na⁺; (vi) LiAlH_4 ; (vii) $\text{H}_2/\text{Pd}/\text{C}$.

In addition, as will be exemplified later, this route permits the efficient synthesis of 1-benzyl-1,2,3,4-tetrahydroisoquinolines. The use of the anions of cyanohydrin ethers in synthesis has recent precedent.⁵

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