CHEMISTRY LETTERS, pp. 385-386, 1986.

NOVEL PHOTOREARRANGEMENT OF 3-PHENYL-2-PROPENYLGERMANES

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3-Phenyl-2-propenylgermanes,  $R_3GeCH_2CH=CHPh$  (R=Me, Et, Pr, Bu, Ph) were found to rearrange to 1-phenyl-2-propenylgermanes,  $R_3GeCH(Ph)CH=CH_2$  upon irradiation with low pressure mercury lamp in cyclohexane.

There have been a number of studies on the photochemical rearrangement of organosilanes.<sup>1)</sup> For example, silyl substituted phenylpropynes underwent photochemical [1,3]-rearrangement to afford silyl substituted 3-phenylpropadienes.<sup>2)</sup> On the contrary, very little is known about the photochemical behavior of the organogermanium compounds.<sup>1)</sup>

We wish to report here that irradiation of 3-phenyl-2-propenylgermanes,  $R_3GeCH_2CH=CHPh$  (R=Me, Et, Pr, Bu, Ph), (<u>1a-e</u>) gave 1-phenyl-2-propenylgermanes,  $R_3GeCH(Ph)CH=CH_2$ ,(<u>2a-e</u>); this is the first example of the photochemical [1,3]rearrangement of germyl groups in germylolefins.

The starting compounds, <u>1b</u> and <u>1e</u>, were prepared by Grignard method.<sup>3)</sup>  $R_3GeBr + PhCH=CHCH_2MgCl \longrightarrow R_3GeCH_2CH=CHPh$ 

Compounds <u>1a</u>, <u>1c</u> and <u>1d</u> were synthesized by reaction of 3-phenyl-2-propenyltrichlorogermane' (<u>3</u>), which was prepared from trichlorogermane and cinnamyl chloride (75%), with appropriate alkylmagnesium halides.

 $PhCH=CHCH_2Cl + HGeCl_3 \rightarrow Cl_3GeCH_2CH=CHPh \rightarrow R_3GeCH_2CH=CHPh$ 

 $\frac{3(75\%)}{2} RMgX \frac{1a}{1a}; R=Me, \frac{1c}{R}; R=Pr, \frac{1d}{R}; R=Bu^{5}$ Photolysis of 3-phenyl-2-propenylgermanes (<u>1a-e</u>) upon irradiation with low pressure mercury lamp in cyclohexane afforded their isomers, 1-phenyl-2-propenylgermanes (<u>2a-e</u>).<sup>6</sup>

 $R_3 GeCH_2 CH=CHPh \longrightarrow R_3 GeCHCH=CH_2$ <u>1</u> Ph <u>2</u>

For example, 3-phenyl-2-propenyltriethylgermane (<u>1b</u>) (0.25 mmol) was irradiated externally with 160 W low pressure mercury lamp in cyclohexane (10 ml) until <u>1b</u> and <u>2b</u> attained to a constant ratio (1b/2b=9/91; ca. 22 h) and the resulting mixture was analyzed by GLC. <u>2b</u> was isolated by preparative GLC, and its structure was confirmed by <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, and MS spectra [<u>2b</u>;<sup>13</sup>C-NMR,  $\delta$  111.0 (t, J<sub>C-H</sub>=156.9 Hz), terminal carbon of C=CH<sub>2</sub>; MS, m/e=278 (M<sup>+</sup>)].

Irradiation of <u>2b</u> under similar conditions gave the mixture of <u>1b</u> and <u>2b</u> in almost same ratio (<u>1b/2b</u>=8/92; ca. 22 h). Thus, <u>1</u> and <u>2</u> are in photoequilibrium under the irradiation conditions, and the equilibrium lies toward the compound <u>2</u>. While starting <u>1</u> was E-form, recovered <u>1</u> was a mixture of E-Z isomer (for example, E/Z ratio of <u>1b</u> after the irradiation was 1:2.15 ).

Tabl	.e 1. Pho	Photorearrangement of <u>1</u> to <u>2</u> <sup>a)</sup>					
Entry		Irr. Time/h	Product ratio	$(1/2)^{b}$	)		
1	Me	21	19/81				
2	Et	22	9/91				
3	Pr	21	16/84				
4	Bu	21	14/86				
5	Ph	22	29/71				
	rradiator	a with lo		orauru	1.200	in	

The results of the photolysis of 1a-e were summarized in Table 1.

a) Irradiated with low pressure mercury lamp in cyclohexane; concentration, 2.5-3.0 x 10<sup>-2</sup>mol/1.

b) Since no product other than photoisomer was detected, conversion yield was considered as quantitative.

It is well known that germyl radicals easily add to olefins.<sup>4)</sup> However, in the present photorearrangement, the multi-step reaction mechanism involving the dissociation of  $\underline{1}$  to form germyl free radical and recombination of the geminate radical pair may be denied, since no hexyltriethylgermanes were detected by GC-MS upon irradiation in the presence of 1-hexene (5.0 mol/mol of  $\underline{1}$ ) under the similar conditions to entry 2 in Table 1. In every case of the photoreaction of  $\underline{1a}$ -e, reaction is very clean and formation of by-products such as digermane was not observed by GLC. Also no thermal isomerization of  $\underline{1}$  was observed even after heating for 41 h at 150°C in cyclohexane solution in a sealed ampoule. These facts and existence of photoequilibrium indicate the present rearrangement proceeds via orbital symmetry allowed concerted [1,3]-sigmatropic mechanism.

We thank to Asai Germanium Research Institute for donation of tetrachlorogermane and trichlorogermane.

## References

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- 5) Bp(°C/mmHg): <u>1a</u>, 93-93.5/3; <u>1c</u>, 122-124/0.3; <u>1d</u>, 150-151/0.3. Mp: <u>1e</u>, 79.5-81°C.
- 6) All products gave satisfactory spectral data. Data of  $\underline{2b}, \underline{e}$  are shown as example.  $\underline{2b}$ : <sup>1</sup>H-NMR;  $\delta 0.33-1.27(m, 15H)$ , 3.33(d, J=9.6 Hz, 1H), 4.90(d, J=11.4 Hz, 1H), 4.94(d, J=16.8 Hz, 1H), 6.28(m, 1H), 7.00-7.57(m, 5H); <sup>13</sup>C-NMR;  $\delta 3.5$ , 8.8, 42.0,  $111.0(t, J_{C-H}=156.9 Hz)$ , 124.3, 126.8, 128.2, 129.0, 143.1. MS;  $m/e=278 (M^+)$ .  $\underline{2e}$ : <sup>1</sup>H-NMR;  $\delta 4.01(d, J=10.2 Hz, 1H)$ , 4.91(d, J=16.2 Hz, 1H), 4.94(d, J=12.6 Hz, 1H), 6.30(m, 1H), 6.83-7.77(m, 20 H); <sup>13</sup>C-NMR;  $\delta 43.3$ ,  $113.5(t, J_{C-H}=158.7)$

Hz), 125.3, 127.9, 128.2, 128.4, 128.9, 134.0, 135.5, 138.3, 140.8. MS; m/e= 422(M<sup>+</sup>).

( Received December 27, 1985 )