

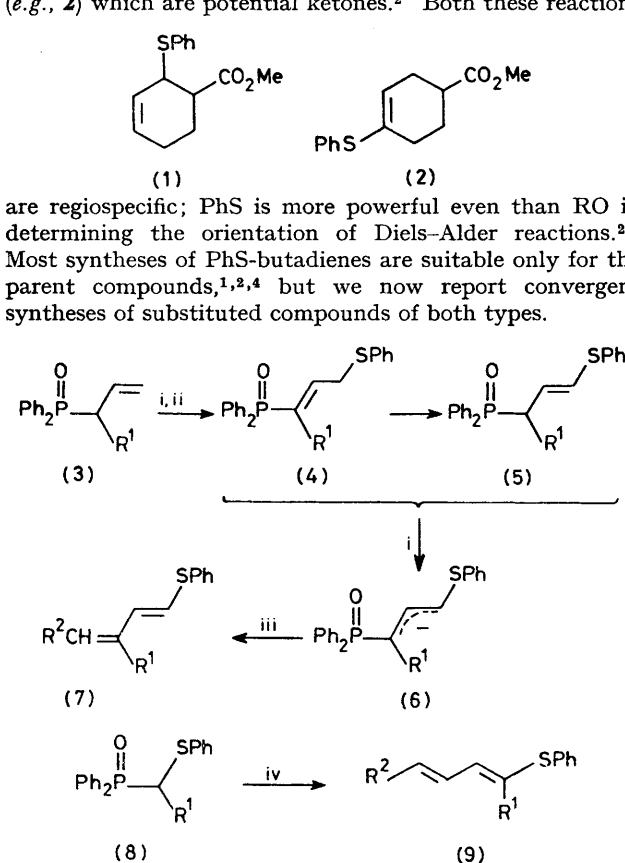
New Syntheses of 1- and 2-Phenylthiobutadienes

By PHILIP BLATCHER, J. IAN GRAYSON, and STUART WARREN*
(University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW)

Summary The Horner–Wittig reaction with sulphenylated allylphosphine oxides and the rearrangement and oxidation of α -hydroxy-bis(phenylthio) acetals are used to make 1- and 2-phenylthiobutadienes respectively.

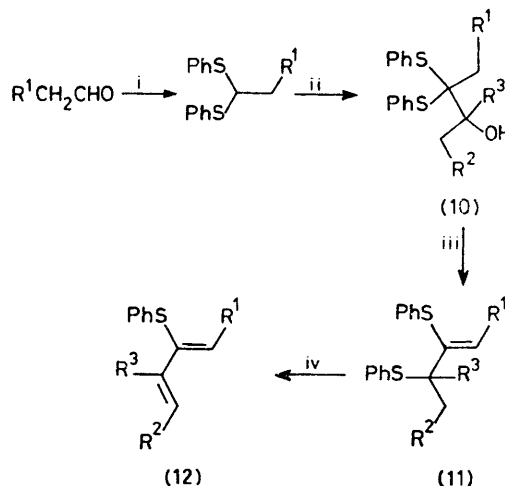
DIELS–ALDER reactions with 1-phenylthio (1-PhS) butadiene give cyclic allyl sulphides (*e.g.*, **1**) from which allylic alcohols, cyclic dienes, or aromatic compounds can be made.¹ 2-PhS butadiene on the other hand gives cyclic vinyl sulphides (*e.g.*, **2**) which are potential ketones.² Both these reactions

1) only the γ -adduct (**4**) is formed (*e.g.*, $R^1 = \text{Me}$, 71%) as the α -adduct gives (**4**) by a [1,3] PhS shift.⁶ The vinyl sulphide (**5**) is formed from (**4**) under the reaction conditions but this does not matter as (**4**) and (**5**) both give the same anion (**6**) which reacts with aldehydes to give the 3,4-disubstituted dienes (**7**; *e.g.*, $R^1 = \text{Me}$, $R^2 = \text{Ph}$, 68%) by the Horner–Wittig reaction.⁷ For 1,4-dialkyl derivatives (**9**) it is better to use reagents (**8**)⁸ on enals (*e.g.*, $R^1 = \text{Pr}^i$, $R^2 = \text{Ph}$, 92%;⁸ $R^1 = \text{H}$, $R^2 = \text{Me}$, 96%).



SCHEME 1. i, BuLi; ii, Ph_2S_2 ; iii, R^2CHO ; iv, (1) i, (2) $\text{R}^2\text{CH}=\text{CHCHO}$.

1-Alkylallylphosphine oxides (**3**), readily available by the rearrangement of phosphinite esters,⁵ form anions with BuLi which give mixtures of α - and γ -adducts with most electrophiles. However, with diphenyl disulphide (Scheme



SCHEME 2. i, PhSH, HCl; ii, BuLi, $\text{R}^2\text{CH}_2\text{COR}^3$; iii, SOCl_2 , Et_3N ; iv, (1) NaIO_4 , (2) heat.

The route to substituted 2-PhS butadienes (**12**) (Scheme 2) starts with the alcohols (**10**) made in good yield from bis-(phenylthio) acetals and aldehydes or ketones.⁹ On treatment of these alcohols with thionyl chloride and triethylamine, one PhS group migrates, assisted no doubt by the other, to give compounds (**11**) containing vinylic and allylic PhS groups (*e.g.*, $R^1 = \text{Me}$, $R^2 = R^3 = \text{H}$, 90%; $R^1 = \text{Bu}^i$, $R^2 = \text{H}$, $R^3 = \text{Me}$, 76%). The vinylic PhS group is conjugated enough for the allylic PhS group to be selectively oxidised to the sulfoxide; thermal elimination of this sulfoxide then gives the 2-PhS diene (**12**), *e.g.*, $R^1 = \text{Ph}$, $R^2 = \text{Me}$, $R^3 = \text{H}$, 62% overall from (**11**).

We have carried out Diels–Alder reactions of both types of diene with acrylate esters and they give substituted derivatives of (**1**) and (**2**), *e.g.*, from (**9**; $R^1 = \text{H}$, $R^2 = \text{Me}$), 90%, from (**12**; $R^1 = \text{Bu}^i$, $R^2 = \text{H}$, $R^3 = \text{Me}$), 78%.

(Received, 5th May 1978; Com. 484.)

¹ D. A. Evans, C. A. Bryan, and C. L. Sims, *J. Amer. Chem. Soc.*, 1972, **94**, 2891.

² T. Cohen, A. J. Mura, D. W. Shull, E. R. Fogel, R. J. Ruffner, and J. R. Falck, *J. Org. Chem.*, 1976, **41**, 3218.

³ B. M. Trost, J. Ippen, and W. C. Vladuchick, *J. Amer. Chem. Soc.*, 1977, **99**, 8116.

⁴ K.-D. Gundermann and P. Holtmann, *Angew. Chem. Internat. Edn.*, 1966, **5**, 668; R. H. Everhardus, A. Peterse, P. Vermeer, L. Brandsma, and J. F. Arens, *Rec. Trav. chim.*, 1974, **93**, 90, but see R. H. Everhardus, R. Gräling, and L. Brandsma, *ibid.*, 1978, **97**, 69; A. de Groot and B. J. M. Jansen, *Synthesis*, 1978, 52.

⁵ M. P. Savage and S. Trippett, *J. Chem. Soc. (C)*, 1966, 1842.

⁶ P. Brownbridge and S. Warren, *J.C.S. Perkin I*, 1976, 2125; H. Kwart and T. J. George, *J. Amer. Chem. Soc.*, 1977, **99**, 5214.

⁷ A. H. Davidson and S. Warren, *J.C.S. Perkin I*, 1976, 639.

⁸ J. I. Grayson and S. Warren, *J.C.S. Perkin I*, 1977, 2263.

⁹ P. Blatcher, J. I. Grayson, and S. Warren, *J.C.S. Chem. Comm.*, 1976, 547; P. Blatcher and S. Warren, *ibid.*, p. 1055; *J.C.S. Perkin I*, submitted for publication.