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Studies on the Constituents of Asclepiadaceae Plants. LIV. 1) The Structures of Glaucoside-F and -G from the Chinese Drug "Pai-ch'ien," Cynanchum glaucescens Hand-Mazz

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The glycosides of the Chinese crude drug "Pai-ch'ien" have been further investigated. Two new glycosides named glaucoside-F (8) and -G (9) were isolated and their structures were characterized on the bases of spectroscopic evidence and analyses of their hydrolysates by thin-layer chromatography (TLC). They were found to possess α -L-cymaropyranose at the terminal of their sugar chains, like glaucoside-B (4), -C (5), -D (6), and -E (7).

Keywords——glaucoside-F,-G; ¹³C-NMR; "Pai-ch'ien"; *Cynanchum glaucescens*; Asclepiadaceae

We have already reported in a previous paper²⁾ five glycosides named glaucoside-A (3), -B (4), -C (5), -D (6), and -E (7) isolated from Chinese crude drug "Pai-ch'ien''³) 芫花叶白前, dried root of *Cynanchum glaucescens* Hand-Mazz (Asclepiadaceae), which has been used as an antitussive and expectorant in China. Compounds 4, 5, 6, and 7 possess unique sugar chains, in that the terminal sugars of their sugar chains are α -linked L-cymaropyranose, while the other linkages are all β . This paper deals with the isolation and structural elucidation of two new

Chart 1

glycosides named glaucoside-F (8) and -G (9).

The less polar portion of the crude glycosides was subjected to repeated silica gel column chromatography with various solvent systems to yield 8 and 9 as amorphous white powders.

Glaucoside-F (8) has the molecular formula $C_{42}H_{64}O_{15}$, and gave glaucogenin-A (1), oleandrose, and cymarose on hydrolysis; these were identified by TLC comparison with authentic samples. The proton nuclear magnetic resonance spectrum (¹H-NMR) in deuterochloroform (CDCl₃) showed signals due to the sugars: three secondary methyls at δ 1.24, 1.27, and 1.31; three methoxyl methyls at δ 3.39, 3.42, and 3.48; and three anomeric protons at δ 4.49, 4.96 (each 1H, dd, J=10, 2 Hz), and 4.80 (1H, br d, J=4 Hz), indicating the presence of

Table I. ¹³C-NMR Chemical Shifts for 1, 2, 8, 9, and 10 (ppm in C_5D_5N)

| | 1 | 8 | 2 | 9 | | 10 |
|-------------|-------|------------|-------|-------------|------|-------|
| C 1 | AF F | 44.5 | 36.6 | 36.4 | | |
| C – 1 | 45.5 | | | 29.9 | | |
| C - 2 | 72.4 | 69.6(-2.8) | 30.0 | | | |
| C - 3 | 76.7 | 85.0(+8.3) | 78.1 | 78.0 | | |
| C - 4 | 40.1 | 37.5(-2.6) | 39.0 | 38.9 | | |
| C - 5 | 140.9 | 139.4 | 140.7 | 140.3 | | |
| C - 6 | 120.0 | 120.4 | 120.4 | 120.1 | | |
| C-7 | 30.1 | 29.9 | 30.0 | 29.0 | | |
| C - 8 | 53.2 | 52.9 | 53.3 | 53.1 | | |
| C-9 | 40.4 | 40.1 | 40.7 | 40.6 | | |
| C-10 | 40.4 | 39.3 | 38.7 | 38.6 | | |
| C –11 | 23.9 | 23.8 | 23.9 | 23.9 | | |
| C –12 | 28.2 | 28.3 | 28.4 | 28.4 | | |
| C-13 | 118.5 | 118.1 | 118.4 | 118.2 | | |
| C –14 | 175.4 | 174.8 | 175.4 | 175.0 | | |
| C -15 | 67.8 | 67.5 | 67.7 | 67.7 | | |
| C-16 | 75.5 | 75.3 | 75.5 | 75.4 | | |
| C-17 | 56.2 | 56.0 | 56.2 | 56.0 | | |
| C-18 | 143.8 | 143.5 | 143.8 | 143.5 | | |
| C-19 | 19.2 | 18.9 | 18.6 | 18.6 | | |
| C -20 | 114.3 | 113.9 | 114.3 | 114.1 | | |
| C -21 | 24.8 | 24.7 | 24.8 | 24.8 | | |
| C - 1' | | 98.6^{a} | 102.4 | 102.0 | | |
| C – 2' | | 37.3 | 75.0 | 74.4 | | |
| C - 3' | | 81.4 | 88.0 | 85.6(-2.4) | | |
| C - 4' | | 82.0 | 75.9 | 82.5(+6.6) | | |
| C - 5' | | 73.0 | 72.6 | 71.4(-1.2) | | |
| C- 6' | | 18.5 | 17.9 | 17.8 | | |
| -QMe | | 57.2 | 60.8 | 60.3 | | |
| C- 1" | | 98.8^{a} | 00.0 | 98.6% | | |
| C-1 $C-2''$ | | 36.8 | | 38.4 | | |
| C - 2" | | 77.6 | | | | |
| C - 3" | | | | 69.0 | | |
| | | 82.2 | | 80.6 | | |
| C – 5" | | 69.3 | | 67.7 | | |
| C - 6" | | 18.5 | | $18.4^{c)}$ | | |
| -QMe | | 58.2 | | 00.00 | 0.1 | 07. 0 |
| C – 1′′′ | | 98.0 | | 98.26) | C-1 | 97.6 |
| C - 2''' | | 31.9 | | 32.1 | C-2 | 31.9 |
| C – 3′′′ | | 76.1 | | 76.3 | C -3 | 76.5 |
| C - 4''' | | 73.1 | | 72.6 | C -4 | 73.2 |
| C – 5′′′ | | 66.0 | | 66.9 | C -5 | 65.2 |
| C-6''' | | 18.5 | | 18.5°) | C-6 | 18.9 |
| –QMe | | 56.3 | | 56.6 | -QMe | 56.7 |
| | | | | | | 54.7 |

a-c) Assignments may be interchanged.

two β -linkages and one α -linkage. The field desorption mass spectrum (FD-MS) of 8 displayed prominent fragment ion peaks at m/z: 664, 520, and 376 besides the molecular ion peak (m/z): 808); the former peaks may be attributed to successive loss of three sugars starting from the terminal as reported by Shulten and co-workers.⁴⁾ In the ¹³C nuclear magnetic resonance spectrum (13C-NMR) of 8 in pentadeuteropyridine (C₅D₅N) (Table I), glycosidation shifts⁵⁾ were observed at C-2 (-2.8 ppm), C-3 (+8.3), and C-4 (-2.6) in the aglycone moiety, as in the cases of 3, 4, 5, and 6, so that the position at which the sugar is linked should be the C-3 hydroxyl group of the aglycone. The presence of sugar carbon signals corresponding to those for methyl α -L-cymaropyranoside²⁾ (10) located the α -linked cymaropyranose at the terminal of the sugar chain in 8, while the other sugars were assigned as β -linked cymaro- and oleandropyranose on the basis of glycosidation shifts at C-4. These spectroscopic data provided no further information on the sequence of sugars. Fortunately, however, acid hydrolysis of 8 under mild conditions gave glaucoside-A (3) in the partial hydrolysate; it was identified by comparison with an authentic sample. The occurrence of L-cymarose, 6) p-oleandrose, 7) and D-digitoxose⁸⁾ in the hydrolysate of this material had been confirmed;²⁾ therefore the structure of 8 was established as glaucogenin-A $3-O-\alpha-L-cymaropyranosyl-(1\rightarrow 4)-\beta-L-cymaropyranosyl (1\rightarrow 4)$ - β -D-oleandropyranoside. The lower hydrolyzability of the β -D- $(1\rightarrow 3)$ -linkage compared with the others cannot be accounted for at present.

Glaucoside-G (9) has the molecular formula $C_{41}H_{62}O_{15}$, and gave glaucogenin-C mono-p-thevetoside (2), digitoxose, and cymarose on hydrolysis. The ¹H-NMR spectrum showed signals due to three sugars: three secondary methyls at δ 1.25, 1.26×2; two methoxyl methyls at δ 3.26 and 3.42; and three anomeric protons at δ 4.33 (1H, d, J=7.3 Hz), 4.91 (1H, br d, J=4 Hz), and 4.95 (1H, dd, J=9, 2 Hz), also suggesting the presence of two β -linkages and one α -linkage. The ¹³C-NMR spectrum of 9 showed signal groups assignable to 2 glycosylated at the C-4 hydroxyl group of thevetose, β -linked digitoxopyranose glycosylated at the C-4 hydroxyl group, and α -linked cymaropyranose, so that the thevetose, digitoxose, and cymarose should be attached to the aglycone in that order. This was supported by the prominent fragment peaks at m/z: 650, 520, and 360 in the FD-MS of 9. Thus, the structure glaucogenin-C 3-O- α -L-cymaropyranosyl-(1 \rightarrow 4)- β -D-digitoxopyranosyl-(1 \rightarrow 4)- β -D-thevetopyranoside was assigned to glaucoside-G (9).

It should be noted that the terminal sugars of six trisaccharide glycosides so far obtained from this drug are all α -linked L-cymaropyranose, while the other linkages are all β . The more polar portion of the glycosides is currently under investigation.

Experimental

Melting points were determined on a Kofler hot stage apparatus and are uncorrected. Optical rotations were measured with a JASCO DIP-4 digital polarimeter at room temperature. Infrared (IR) spectra were recorded on a JASCO A-102 spectrometer. $^1\text{H-NMR}$ spectra were run on a JEOL FX-200 (200 MHz) in CDCl₃ solution and $^{13}\text{C-NMR}$ spectra on a JEOL FX-100 (25 MHz) in C₅D₅N solution with tetramethylsilane as a standard. Electron impact (EI)-MS were determined with a JEOL JMS-D-300 mass spectrometer and FD-MS with a JEOL JMS-01SG-2. TLC was performed on Merck precoated plates (Kieselgel 60 F₂₅₄), and silica gel column chromatography on Wakogel C-200 (200 mesh) or C-300 (300 mesh).

Isolation of 8 and 9——A part of the hexane-benzene (1:1) and benzene soluble portion of the crude glycosides (40 g) reported in the previous paper¹⁾ was applied to a column of silica gel (400 g of Wakogel C-200), and the material was eluted with solvents of increasing polarity from benzene-acetone (8:1) to acetone. The fraction eluted with benzene-acetone (5:1) contained 3 and 4. Fraction 3 (8.0 g), which contained 5, 6, 7 and 8, was obtained by further elution with the same solvent. Fraction 4 (9.3 g), eluted with benzene-acetone (4:1), contained 6 and 9. Fraction 3 (8.0 g) was rechromatographed with 1.5% methanol (MeOH) in chloroform (CHCl₃) to yield five fractions (fractions A to E). Fraction C (2.24 g) contained 5 and 7, and fraction D (1.07 g) contained 6. Fraction B (640 mg), containing 8, was further rechromatographed with hexane-EtOAc (1:2) to give a fraction (137 mg) containing mainly 8, which was rechromatographed with 2% MeOH in benzene to furnish 8 (128 mg). Fraction 4 (9.30 g) was subjected to rechromatography with 3% MeOH in CHCl₃, hexane-EtOAc-MeOH (25: 25: 2), and 3% MeOH in benzene

in that order to give 9 (87 mg). Compounds 8 and 9 were obtained as amorphous white powders. The Rf values of 8 and 9 on TLC with 5% MeOH in CHCl₃ were 0.67 and 0.52, respectively.

Glaucoside-F (8)——An amorphous powder, mp 110—113°C, $[\alpha]_D$ —17.4° (c = 1.21, CHCl $_3$). Anal. Calcd for C $_{42}$ H $_{64}$ O $_{15}$ ·H $_2$ O: C, 61.00; H, 8.05. Found: 61.00; H, 7.80. IR $\nu_{\max}^{\text{CHCl}_3}$ cm $^{-1}$: 3550, 3450, 1730, 1710, 1655, 1310, 1050, 1010, 880. FD-MS m/z: 808 (M $^+$, base peak), 664, (M $^+$ —144), 520 (664—144), 376 (520—144). 1 H-NMR (CDCl $_3$) δ : 0.94 (3H, s, 19-CH $_3$), 1.00 (1H, t, J = 12 Hz, 1-CH $_4$), 1.24, 1.27, and 1.31 (each 3H, d, J = 6 Hz, 5′-, 5″-, and 5″′-CH $_3$), 1.53 (3H, s, 21-CH $_3$), 1.74 (1H, dt, J = 12, 5 Hz, 4-CH $_4$), 3.39, 3.42, and 3.48 (each 3H, s, 3′-, 3″-, and 3″′-OCH $_3$), 3.84 (1H, dd, J = 10, 9 Hz, 15-CH $_4$), 4.16 (1H, dd, J = 9, 7 Hz, 15-CH $_4$), 4.49 (1H, dd, J = 10, 2 Hz, 1′-CH), 4.80 (1H, br d, J = 4 Hz, 1″′-CH), 4.96 (1H, dd, J = 10, 2 Hz, 1″-CH), 5.30 (1H, ddd, J = 10, 8, 7 Hz, 16-CH), 5.42 (1H, d, J = 5 Hz, 6-CH), 6.27 (1H, d, J = 2 Hz, 18-CH). 13 C-NMR: see Table I.

Glaucoside-G (9)—An amorphous powder, mp 117—123°C, $[\alpha]_D$ —29.6° $(c=0.81, \text{CHCl}_3)$. Anal. Calcd for $C_{41}H_{62}O_{15}$: C, 61.95; H, 7.68. Found: C, 61.74; H, 7.92. IR $\nu_{\max}^{\text{CHCl}_3}$ cm⁻¹: 3550, 3300, 1730, 1710, 1655, 1310, 1100, 100, 930, 860. FD-MS m/z: 795 (M^++H) , base peak), 794 (M^+) , 650 (M^+-144) , 520 (650—130), 360 (520—160). ¹H-NMR (CDCl $_3$) δ : 0.92 (3H, s, 19-CH $_3$), 1.25, 1.26 (3H and 6H, respectively, each d, J=6.4, 6.8 Hz, 6'-, 6"-, and 6"'-CH $_3$), 1.53 (3H, s, 21-CH $_3$), 3.26, 3.42 (each 3H, s, 3'-and 3"'-OCH $_3$), 3.84 (1H, dd, J=10, 9 Hz, 15-CH $_3$), 4.16 (1H, dd, J=9, 7 Hz, 15-CH $_4$), 4.33 (1H, d, J=7.3 Hz, 1'-CH), 4.91 (1H, br d, J=4 Hz, 1"'-CH), 4.95 (1H, dd, J=9, 2 Hz, 1"-CH), 5.35 (1H, ddd, J=10, 8, 7 Hz, 17-CH), 5.39 (1H, d, J=5 Hz, 6-CH), 6.25 (1H, d, J=2 Hz, 18-CH). ¹³C-NMR: see Table I.

Partial Acidic Hydrolysis of 8——A solution of 36 mg of 8 in 3 ml of MeOH was treated with 1 ml of 0.2 N $\rm H_2SO_4$, and kept at around 60°C for 20 min. TLC analysis with CHCl₃-acetone (5: 1) revealed the formation of 3 (Rf, 0.37) and 1 (Rf, 0.22) as well as methyl glycosides. The solution was neutralized with saturated aqueous $\rm Ba(OH)_2$ and the precipitates were filtered off. The filtrate was concentrated and the residue was subjected to silica gel column chromatography with CHCl₃-acetone (15: 1) to give 3 (5.6 mg) as an amorphous white powder, mp 85—90°C, $\rm [\alpha]_D$ +8.93° ($\rm c=0.56$, CHCl₃). IR $\rm \nu_{max}^{\rm CHCl_3}$ cm⁻¹: 3600, 3350, 1730, 1710, 1655, 1310, 1165, 1070, 950. $\rm ^{1}H$ -NMR (CDCl₃) $\rm \delta$: 0.95 (3H, s, 19-CH₃), 1.02 (1H, t, $\rm J=12$ Hz, 1-CH_a), 1.36 (3H, d, $\rm J=6.4$ Hz, 6'-CH₃), 1.54 (3H, s, 21-CH₃), 3.74 (1H, ddd, $\rm J=12$, 10, 5 Hz, 2-CH_β), 3.85 (1H, dd, $\rm J=10$, 9 Hz, 15-CH_β), 4.16 (1H, dd, $\rm J=9$, 7 Hz, 15-CH_a), 4.55 (1H, dd, $\rm J=10$, 2 Hz, 1'-CH), 5.35 (1H, ddd, $\rm J=10$, 8, 7 Hz, 17-CH), 5.45 (1H, d, $\rm J=4.5$ Hz, 6-CH). The $\rm [\alpha]_D$, IR, MS, and $\rm ^{1}H$ -NMR data are identical with those of 3. The identity of these two compounds was further confirmed by TLC with three solvent systems: Rf 0.56 (5% MeOH in CHCl₃); Rf 0.57 (benzene-acetone (5: 3)); Rf 0.68 (3% ethanol in methylene chloride).

Acidic Hydrolysis of 8 and 9——A solution of 3 mg of 8 in 2 ml of MeOH was treated with 2 ml of $0.1 \, \mathrm{N} \, \mathrm{H}_2 \mathrm{SO}_4$, and kept at around $60^{\circ}\mathrm{C}$ for 30 min, then the solution was diluted with 2 ml of water and concentrated to 1/2 the initial volume. The solution was again kept at around $60^{\circ}\mathrm{C}$ for a further 30 min, then neutralized with saturated $\mathrm{Ba}(\mathrm{OH})_2$, and the precipitates were filtered off. The filtrate was concentrated to give a yellow syrup, which was analyzed by TLC with three solvent systems: solvent A, CHCl₃-MeOH (9: 1); solvent B, methylene chloride-ethanol (9: 1); and solvent C, benzene-acetone (5: 3). The Rf values of 1, 2, cymarose, oleandrose, and digitoxose were 0.51, 0.55, 0.47, 0.43, and 0.21 with solvent A, 0.53, 0.56, 0.42, 0.33 and 0.21 with solvent B; and 0.43, 0.49, 0.43, 0.31, and 0.17 with solvent C, respectively. When 8 was hydrolyzed, 1, oleandrose, and cymarose were identified by TLC comparisons with authentic samples (solvents A, B, and C). Similarly, 3 mg of 9 was hydrolyzed, and 2, cymarose, and digitoxose were identified.

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