Synthesis of 6H,11H-Pyrrolo[1,2-b][2,5]benzodiazocine: a New Nitrogen Heterocyclic System

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Summary Vilsmeier-Haack formylation of 1-(2-acetamidomethylbenzyl)pyrrole, followed by alkaline hydrolysis, afforded directly 6H,11H-pyrrolo[1,2-b][2,5]benzodiazocine.

In continuation of studies¹ on the synthesis of new heterocyclic systems potentially useful as starting materials for

derivatives of pharmaceutical significance, we describe the preparation of a previously unreported ring system 6H,-11H-pyrrolo[1,2-b][2,5]benzodiazocine (I). 1-(2-Cyanobenzyl)pyrrole (II), m.p. 43-44° (petrol), was formed in 75% yield by reaction of o-bromomethylbenzonitrile with pyrrole potassium salt in tetrahydrofuran. Lithium aluminium hydride reduction of (II) in anhydrous ether gave oily 1-(2-aminomethylbenzyl)pyrrole (III), which was characterized as its picrate, m.p. $168-169^{\circ}$ (water). On treatment with cold acetic anhydride the compound (III) was easily transformed into 1-(2-acetamidomethylbenzyl)pyrrole (IV), m.p. 99-100° (water), which was then formylated by the Vilsmeier-Haack method at room temperature giving 1-(2-acetamidomethylbenzyl)pyrrole-2-carboxaldehyde (V), m.p. 95—96° (cyclohexane).

Alkaline hydrolysis of (V) afforded directly 6H,11H-pyrrolo[1,2-b][2,5]benzodiazocine (I), m.p. 290—291° (ethylacetate).

All the compounds gave satisfactory elemental analyses and spectral data.

We thank the Italian National Council of Research (C.N.R.) for financial aid.

(Received, October 18th, 1971; Com. 1802.)

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