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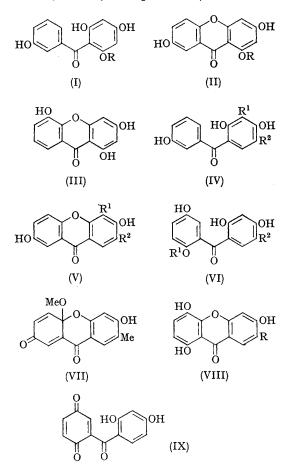
Biogenetic-type Syntheses of Xanthones

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In the course of investigations directed towards the synthesis of naturally occurring derivatives of xanthones (*e.g.*, ergoflavin¹), we have examined potential biogenetic-type approaches to these compounds, including the oxidative coupling² of various benzophenones.

Thus, oxidation of 2,8',4,6-tetrahydroxybenzophenone (I; R = H) with potassium ferricyanide in aqueous acetone containing sodium hydrogen carbonate, gave the xanthone (II; R = H) in 62% yield. The isomeric xanthone (III) arising by ortho-coupling of (I; R = H) could not be detected. Similarly, 2,3',4-trihydroxy-6-methoxybenzophenone (I; R = Me) gave only (II; R = Me) in 45% yield, whilst 2,3',4-trihydroxy-5-methyl-(IV; $R^1 = H$, $R^2 = Me$) and 2,3',4-trihydroxy-2methyl-benzophenone (IV; $R^1 = Me$, $R^2 = H$) furnished the xanthones (V; $R^1 = H$, $R^2 = Me$) and (V; $R^1 = Me$, $R^2 = H$) respectively, unaccompanied by any *ortho*-coupled products.

Ferricyanide oxidation of 2,4,5'-trihydroxy-2'methoxy-5-methylbenzophenone (VI; $\mathbb{R}^1 = Me$,



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 $R^2 = Me$) in aqueous sodium carbonate gave only the *para*-coupled dienone (VII) in 51% yield. Treatment of (VII) with hydrochloric acid-acetic acid gave (quantitatively) 4-chloro-2,6-dihydroxy-7-methylxanthone, whilst aromatisation with zincacetic acid furnished (V; $R^1 = H$, $R^2 = Me$) in 95% yield.

The 1,4-dihydroxyanthones of type (VIII) may arise, inter alia, by ortho-oxidative coupling of a benzophenone type (VI; R = H) or alternatively by way of the sequence (VI; R = H), (IX), and (VIII). We have demonstrated the feasibility of both pathways. Thus, e.g., oxidation of 2,2',4,5'tetrahydroxybenzophenone (VI; $R^1 = R^2 = H$) with potassium ferricyanide in aqueous sodium carbonate gave 1,4,6-trihydroxyanthone (VIII; R = H) (14% yield). The action of chromic oxidesulphuric acid upon (VI; $R^1 = R^2 = H$) gave (VIII; R = H) in 33% yield, most probably by way of the intermediate quinone (IX). Oxidation of (VI; $R^1 = R^2 = H$) with 2,3-dichloro-5,6-dicyanop-benzoquinone in benzene at 0° rapidly gave the unstable quinone (IX) (70%) which cyclised quantitatively to the xanthone (VIII; R = H) on solution in warm methanol. Other xanthones, e.g., (VIII; $\mathbf{R} = \mathbf{M}\mathbf{e}$), were similarly prepared from the corresponding benzophenones. The action of tetrachloranil in boiling benzene-methanol converts e.g., (VI; $R^1 = H$, $R^2 = Me$) directly into (VIII; R = Me).

The majority of the benzophenones were synthesised by a modification of the process described by Usgaonkar and Jadhav.³

All new compounds had the requisite spectral and analytical properties. The structures of several xanthones obtained by oxidative coupling were also confirmed by alternative, conventional syntheses.

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¹ J. W. ApSimon, J. A. Corran, N. G. Creasey, K. Y. Sim, and W. B. Whalley, J. Chem. Soc., 1965, 4130.

² J. R. Lewis and B. H. Warrington, J. Chem. Soc., 1964, 5074; J. E. Atkinson and J. R. Lewis, preceding Communication.

³ U. R. Usgaonkar and G. V. Jadhav, J. Indian. Chem. Soc., 1963, 40, 27.