

5. E. V. Isaeva, A. M. Agranovich, N. A. Dobrynina, A. M. Evseev, V. G. Gontar', and S. P. Gladkikh, *Vestn. Mosk. Gos. Univ., Ser. Khim.*, **26**, No. 6, 571 (1985).
6. C. F. Wells, *Trans. Faraday Soc.*, **62**, 2815 (1966).

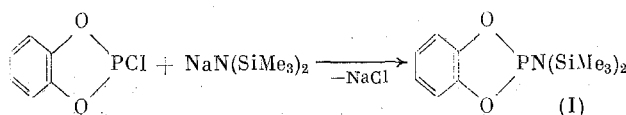
REACTION OF 2-[N,N-BIS(TRIMETHYLSILYL)]AMINO-4,5-BENZO-1,3,2-DIOXAPHOSPHOLANE WITH CARBOXYLIC ACID HALIDES

M. A. Pudovik, L. K. Kibardina, and A. N. Pudovik

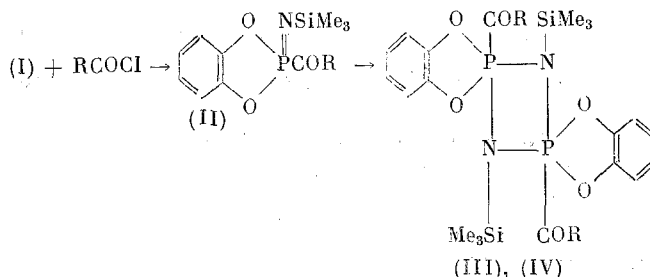
UDC 542.91:547.1'118

The reaction of N-alkyl and N-aryl-N-trimethylsilylamino-1,3,2-oxazaphospholanes with acetyl chloride proceeds with the participation of the nitrogen atom of the P-N system and leads to N,N-diphosphorylated primary amines. In the case of N,N-disilylated P(III) amido derivatives, the P-acylation products, namely, N-silylated imidophosphonates, are formed. Both reaction pathways proceed concurrently for N-acetyl-N-trimethylsilylaminophosphites [1].

In order to study the effect of the nature of the cyclic fragment on the direction of the acylation, we synthesized 2-[N,N-bis(trimethylsilyl)]-amino-4,5-benzo-1,3,2-dioxaphospholane (I) by the reaction of 2-chloro-4,5-benzo-1,3,2-dioxaphospholane with the sodium salt of hexamethyldisilazane.



Phospholane (I), in contrast to the analogous oxazaphospholanes, reacts with AcCl and PhCOCl to form not P-acetyl-N-trimethylsilylimidophosphonates (II) [1], but rather the products of their dimerization, namely, diazadiphosphetidines (III) and (IV) with $\delta^{32}\text{P} = -58$ and -54 ppm, respectively.



The IR spectra have carbonyl bands at 1700 (III) and 1660 cm^{-1} (IV) and trimethylsilyl bands at 1255 cm^{-1} (III) and (IV) but lack absorption at 1300-1350 cm^{-1} characteristic for the $\text{P}=\text{N}-\text{Si}$ group [2-4]. Thus, the introduction of a dioxaphenylene fragment at the phosphorus atom, which stabilizes the trigonal bipyramidal structure, results in the first example of the dimerization of the N-trimethylsilylphosphoimide bond [4].

EXPERIMENTAL

The ^{31}P NMR spectra were taken on a KGU-4 spectrometer at 10.2 MHz relative to 85% H_3PO_4 . The IR spectra were taken on a UR-20 spectrometer for KBr pellets.

A. E. Arbuzov Institute of Organic and Physical Chemistry, Kazan Branch, Academy of Sciences of the USSR. Translated from *Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya*, No. 4, pp. 939-940, April, 1986. Original article submitted July 16, 1986.

2-[N,N-Bis(trimethylsilyl)]amino-4,5-benzo-1,3,2-dioxaphospholane (I). A sample of 17.5 g catechol chlorophosphite was added dropwise with stirring to an ethereal solution of 18.3 g of the sodium salt of hexamethyldisilazane cooled to -20°C . The mixture was warmed to $\sim 20^{\circ}\text{C}$ and maintained for 12 h. After separation of the precipitate and removal of the solvent, distillation gave 16.3 g (55%) (I), bp 78°C (0.05 mm), n_D^{20} 1.5190, d_4^{20} 1.0601, ^{31}P 174 ppm. Found, %: C 48.53; H 7.33; P 10.14; Si 18.32. $\text{C}_{12}\text{H}_{22}\text{NO}_2\text{PSi}_2$. Calculated, %: C 48.16; H 7.35; P 10.36; Si 18.69.

1,3-Bis(trimethylsilyl)-2,4,6-bis(dihydroxyphenylene)-2,4-diacetyl-1,3,2 λ^5 ,4 λ^5 -diazadiphosphetidine (III). A mixture of 3 g (I) and 0.8 g acetyl chloride in 20 ml dry benzene was maintained at 20°C for 24 h to give 1.76 g (65%) (III), mp 132°C . Found, %: C 49.19; H 6.03; N 5.33; P 11.66; Si 10.51. $\text{C}_{22}\text{H}_{32}\text{N}_2\text{O}_6\text{P}_2\text{Si}_2$. Calculated, %: C 49.07; H 6.00; N 5.24; P 11.52; Si 10.41.

1,3-Bis(trimethylsilyl)-2,4,6-bis(dihydroxyphenylene)-2,4-dibenzoyl-1,3,2 λ^5 ,4 λ^5 -diazadiphosphetidine (IV). A sample of 3 g (I) was mixed with 1.41 g PhCOCl . The mixture crystallized after 12 h. Washing with benzene gave 2.01 g (61%) (IV), mp 157°C . Found, %: H 5.54; N 4.11; P 8.87; Si 8.10. $\text{C}_{30}\text{H}_{36}\text{N}_2\text{O}_6\text{P}_2\text{Si}_2$. Calculated, %: H 5.43; N 4.23; P 9.36; Si 8.45.

CONCLUSION

The reaction of 2-[N,N-bis(trimethylsilyl)]amino-4,5-benzo-1,3,2-dioxaphospholane with acetyl and benzoyl chlorides leads to the formation of silicon-containing diazadiphosphetidines.

LITERATURE CITED

1. M. A. Pudovik, L. K. Kibardin, M. D. Medvedeva, and A. N. Pudovik, *Zh. Obshch. Khim.*, **49**, 988 (1979).
2. A. M. Pinchuk, M. G. Suleimanova, and L. P. Filonenko, *Zh. Obshch. Khim.*, **42**, 2115 (1972).
3. O. J. Scherer and G. Schieder, *Chem. Ber.*, **101**, 4184 (1968).
4. A. F. Grapov, L. V. Razvodovskaya, and N. N. Mel'nikov, *Usp. Khim.*, **50**, 606 (1981).

SYNTHESIS OF NEW ORGANOSILICON AND ORGANOGERMANIUM DERIVATIVES

OF DIALLYLISOCYANURATE AND CYANURIC ACID

A. S. Gordetsov, A. P. Kozina, T. N. Brevnova, UDC 542.91:547.1'128.547.491.8
N. F. Cherepennikova, V. V. Semenov, S. E. Skobeleva,
T. I. Chulkova, O. S. Rukevich, and Yu. I. Dergunov

In a continuation of work on the synthesis and investigation of the properties of heteroorganic symm-triazines [1] and in a search for new biologically active compounds and inhibitors of the nitrification of nitrogen fertilizers, we are the first to report the preparation of a series of derivatives of diallylisocyanurate and cyanuric acid containing disilanyl, disiloxane or carbofunctional organogermanium fragments. Thus, the heterogeneous reactions of the silver salt of diallylisocyanurate with pentamethylchlorodisilane or β -triallylgermylpropionyl chloride in ether or benzene solution in a sealed ampule at $\sim 20^{\circ}\text{C}$ over 70-80 h with rapid stirring (method A) gave high yields of the corresponding pentamethyldisilanyldiallylisocyanurate (I), β -(triethylgermyl)- (II) and β -(tributylgermyl)propionyldiallylisocyanurates (III).

Institute of Chemistry, Academy of Sciences of the USSR, Gorkii. S. M. Kirov Gorkii Medical Institute. Translated from *Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya*, No. 4, pp. 941-943, April, 1987. Original article submitted July 18, 1986.