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Aerothionin, a Tetrabromo-compound from Aplysina aerophoba and Verongia thiona

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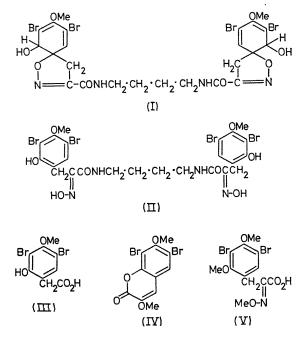
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Summary Aerothionin, a tetrabromo-compound from the sponges Aplysina aerophoba and Verongia thiona is shown to be the spiro-cyclohexadienylisoxazole derivative (I).

AEROTHIONIN, $C_{24}H_{26}Br_4N_4O_8$ (elemental analysis and osmometric M.W.), m.p. $134-137^{\circ}$ (dec.), $[\alpha]_D$ 252° (acetone), the major bromo-compound in the sponges *Aplysina aerophoba* and *Verongia thiona*, is shown to be (I). It forms a diacetate and a bistetrahydropyranyl ether, and has the following spectral properties; λ_{max} (MeOH) 284 nm (ϵ 12,660) (cf. aeroplysinin-I¹), ν_{max} (Nujol) 3335, 3160, 1675, 1660, and 1550 cm.⁻¹; n.m.r. (100 MHz, CD₃COCD₃) δ 7.58 (2H, bt, NHCH₂), 6.50 (2H, s, >C=CH), 5.37 (2H, d, *J* 8 Hz, OH), 4.18 (2H, d, *J* 8 Hz, >CHOH), 3.72 (6H, s,



† One OH signal was not observed.

 OCH_3), 3.84 and 3.14 (each 2H, d, J 18 Hz, CH_2), 3.34 (4H, m, NHCH₂), and 1.60 (4H, m, NHCH₂CH₂). These data show that aerothionin is a symmetrical molecule, the two halves being the same (not mirror images). The >CHOH structure attached to two tertiary carbon atoms was confirmed by addition of D₂O when the signal at δ 5.37 disappeared and the doublet at δ 4.18 collapsed to a singlet, and by the downfield shift (to δ 5.83) of the methine proton signal on acetylation. Addition of DCl eliminated the NH signal at δ 7.58 and simplified the multiplet at δ 3.34, while coupling between the signals at δ 3.34 and 1.60 was confirmed by double irradiation. This spectral evidence is fully consistent with structure (I) and is supported by the following reactions.

Treatment of aerothionin with dilute alkali gave the dihydric phenol (II), m.p. 189—191.5°, λ_{max} (MeOH) 292 nm (ϵ 5880); n.m.r., δ 10.93 (2H, s, exchangeable with D₂O, OH[†]), 7.99 (2H, b, NHCH₂), 7.59 (2H, s, ArH), 3.82 (10H, s, ArCH₂ and OCH₃), 3.40 and 1.62 (each 4H, m, NHCH₂·-CH₂). The relative positions of the aromatic substituents in (II) were established by cleavage in hot 6N-hydrochloric acid which gave (III) (identified as its methyl ester¹), and a lactone, converted by methylation with diazomethane into (IV); λ_{max} (MeOH) 286, 298, and 318 nm (ϵ 11,730, 12,090, and 9950), ν_{max} (Nujol) 1740 cm.⁻¹; n.m.r. (CDCl₃) δ 7.56 (1H, s, ArH), 6.69 (1H, s, >C=CH), and 3.93 (6H, s, OCH₃, split into two 3H singlets by addition of C₆D₆).

The phenol (II) formed a tetramethyl derivative, m.p. 130—131° (M^+ 871·9107 (quintet), $C_{28}H_{34}$ ⁷⁹Br₃⁸¹BrN₄O₈ requires 871·9092), which on hydrolysis with 25% aqueous ethanolic potassium hydroxide gave (V) (synthesised from methoxyamine and the arylpyruvic acid derived² from 3,5-dibromo-2,4-dimethoxybenzaldehyde and N-acetylglycine). Condensation of the acid chloride of (V) with 1,4-diaminobutane gave a product identical with the tetramethyl derivative of (II).

Like aerophysinin- I^1 and two other brominated compounds isolated³ from *Verongia* spp., aerothionin is probably biosynthesised from 3,5-dibromotyrosine which is condensed with a $C_4\mathchar`-N_2$ unit possibly derived from ornithine. The spiro-structure could arise either by nucleophilic attack by an oxime function on an arene epoxide,4 or conversion of the latter into a phenol followed by intramolecular phenol-oxime coupling.

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