Journal of

The Chemical Society,

Chemical Communications

NUMBER 8/1974

17 APRIL

Tenellin and Bassianin, Metabolites of *Beauveria* Species. Structure Elucidation with ¹⁵N- and Doubly ¹³C-Enriched Compounds using ¹³C Nuclear Magnetic Resonance Spectroscopy

By A. GAVIN MCINNES, DONALD G. SMITH, and CHI-KIT WAT

(Atlantic Regional Laboratory, National Research Council of Canada, Halifax, Nova Scotia B3H 3Z1, Canada)

and LEO C. VINING* and JEFFREY L. C. WRIGHT

(Department of Biology, Dalhousie University, Halifax, Nova Scotia, Canada)

Summary ¹³C n.m.r. spectroscopy and biosynthetic labelling with [¹⁵N]nitrate and [1,2-¹³C]acetate proved to be valuable adjuncts to established chemical and spectroscopic methods in elucidating the structures of tenellin and bassianin as 3-acyl derivatives of 1,4-dihydroxy-5*p*-hydroxyphenyl)-2(1*H*)-pyridone.

A PREVIOUS publication¹ contains preliminary u.v., i.r., and mass spectral data on two related pigments tenellin (I) $(C_{21}H_{23}NO_5)$, and bassianin $(C_{23}H_{25}NO_5)$, as well as conditions for their production and isolation from the insect pathogenic fungi *Beauveria tenella* (Delacroix) Siem., and *Beauveria bassiana* (Bals.) Vuill. We now report their structures.

Methylation of (I) with Ag₂O-MeI progressively methylates the hydroxamic acid [(II); m.p. 194°], phenol [(III); m.p. 96—98°], and enol [(IV); wax] hydroxy-groups. High resolution mass spectral studies, including precursor ion experiments, showed a major fragmentation pathway *via* initial loss of oxygen in (I), whereas in contrast fragmentation of (II), (III), and (IV) involved initial loss of CH₂O. This provides strong evidence for a cyclic hydroxamic acid unit² in (I).

Compound (I), $[\alpha]_{2}^{24} - 44.0^{\circ}$ (1% in acetone) with alkaline peroxide gave *p*-hydroxybenzoic acid; in refluxing KOH by retro-aldol cleavage it gave the 3-acetyl derivative

(V), m.p. $230-231^{\circ}$ (C₁₃H₁₁NO₅, m/e 261.0635), and two aldehydes, both racemic mixtures, characterized as the 2,4-dinitrophenylhydrazones of 2-methylbutanal, m.p.³ 128.5° and 2,4-dimethylhex-2-enal, m.p.⁴ 160-161°.

The ¹H n.m.r. signals [(CD_3)₂SO; 100 MHz] of (I) could be assigned as follows: δ 17.0br [s, 1H, -C(OH)=C(C=O)-], 11.5br (s, 1H, >NOH), 9.5br (s, 1H, ArOH; this and previous OH's temperature dependent, exchanged with D₂O), 8·10 (s, 1H, 6-H), 7·78 [AB, 2H, 8- and 9-H, Δν_{AB} 50·4, J_{AB} 15.3 Hz, 9-H long range coupled <0.5 Hz to 11-H (double resonance)], 7.07 (AA'BB', 4H, Δv_{AB} 49.6 Hz, J_{AB} 8.6 Hz, 1,4-subst. Ar), 5.92br [d, 1H, 11-H, J_{11.12} 9.6 Hz, coupled <0.3 Hz to vinylic Me (double resonance)], 2.50 (m, 1H, 12-H, Bu⁸- >CH-), 1.88br (s, 3H, 16-H, vinylic Me), 1.38 (m, 2H, 13-H, Bu⁸- >CH₂), 1.01 (d, 3H, 15-H, Bu⁸-Me, $J_{12,15}$ 6·3 Hz), and 0·85br (t, 3H, 14-H, remaining Bu^s-Me). The only significant differences in the spectra of derivatives were: (II), δ 11.5 (OH) replaced by δ 3.98 (>NOMe); (III), δ 11.5 and 9.5 (OH) replaced by δ 4.02 (>NOMe) and 3.79 (ArOMe); (V), all signals for H-8 replaced by acetyl Me at δ 2.70. Structural, stereochemical, and conformational features of (I) are thus retained in (II) and (III); (V) differs only in the contraction of an acyl to an acetyl substituent. A nuclear Overhauser enhancement (NOE) study⁵ [irradiated proton(s) in brackets] of (II) gave: 9-H {11-H} 22%; 8-H {11-H} 0%; 9-H {allylic Me} 0%; 8-H

{allylic Me} 20%; 6-H {>NOMe} 8%; 6-H {2'- and 6'-H} 26%; 3'- and 5'-H {2'- and 6'-H} 10%.

The preceding evidence requires that (I), (II), and (III) possess 4,6-dimethylocta-trans,trans-2,4-dienoyl and HOC₆- H_4C units, with 6-H close to the chemically equivalent aromatic hydrogens 2'-H and 6'-H, as well as to the Nmethoxy-group in (II) and (III), and hence the N-hydroxygroup in (I).



Pulse Fourier transform (8K; data accuracy ± 0.6 Hz) ¹³C n.m.r. spectral data [(CD₃)₂SO; internal reference Me₄Si, 5 kHz sweep width, 25.16 MHz] were obtained from protonnoise decoupled (p.n.d.) and high resolution spectra, and assignments were aided by off-resonance and single ¹H frequency decoupling experiments.^{6,7} Assignments for (I) were: δ 193·8 (C-7), 173·0 (C-4), 157·5 (C-2), 156·9 (C-4'), 150.8 (C-11, ${}^{1}J_{CH}$ 154.4 Hz), 149.7 (C-9, ${}^{1}J_{CH}$ 155.9 Hz), 140.0 (C-6, ¹*J*_{CH} 183.6 Hz), 132.6 (C-10), 130.2 (C-2', -6', ${}^{1}J_{CH}$ 160.0 Hz), 123.1 (C-8, ${}^{1}J_{CH}$ 167.5 Hz, 122.8 (C-1'), 115.0 (C-3', -5', ${}^{1}f_{CH}$ 160.0 Hz), 110.9 (C-5), 105.9 (C-3), 34.6 (C-12, ¹J_{CH} 125.4 Hz), 29.4 (C-13, ¹J_{CH} 126.0 Hz), 19.8 (C-15, ${}^{1}J_{CH}$ 125.8 Hz), 12.3 (C-16, ${}^{1}J_{CH}$ 126.2 Hz), 11.7 (C-14, ${}^{1}J_{CH}$ 125.2 Hz).

Addition of 90% enriched [1,2-13C] acetate to cultures of B. bassiana yielded (I) containing five discrete two-carbon units each doubly labelled with ¹³C. The presence of ¹³C-¹³C spin-spin coupling in the p.n.d. spectrum proved the existence of the following pairs of directly bonded carbons: C-2, C-3 (${}^{1}J_{CC}$ 75·3 Hz); C-7, C-8 (${}^{1}J_{CC}$ 55·9 Hz); C-9, C-10 $({}^{1}J_{\infty}$ 53.6 Hz); C-11, C-12 $({}^{1}J_{\infty}$ 43.2 Hz); C-13, C-14 $({}^{1}J_{CC} 34 \cdot 4 Hz).$

In addition the p.n.d. spectrum of (I) 95% enriched with ¹⁵N (obtained from cultures with K¹⁵NO₃ as the only nitrogen source) established that only C-6 $[^{1}J(^{15}NC)]$ 15.0 Hz], C-2 [¹J(¹⁵NC) 11.0 Hz], and C-3 [²J(¹⁵NC) 9.2 Hz] were spin-spin coupled to ¹⁵N, and only H-6 [²/(¹⁵NH) 1.0 Hz] was similarly coupled to nitrogen in the corresponding ¹H n.m.r. spectrum. These data confirm that the nitrogen in (I) is geminal to 6-H and C-3, and directly bonded to C-6 and C-2;8 the latter observation provides further evidence for a hydroxamic acid group.

The combined results therefore require that tenellin possesses a 1-hydroxy-2-pyridone ring substituted as shown in structure (I). Our NOE study places H-6 and the p-hydroxyphenyl substituent on adjacent carbons. C-4, C-3, and C-7 must constitute the enol form of a β -diketone system in order to account for their chemical shift,7a and the presence of the strongly hydrogen bonded hydroxygroup observed in the ¹H n.m.r. spectrum; the chemical shift of C-5 is consistent with it being deshielded by the aryl group and shielded by β -nitrogen and hydroxy substituents. Tenellin is therefore 3-(4,6-dimethyl-transtrans-octa-2,4-dienoyl)-1,4-dihydroxy-5-(p-hydroxyphenyl)-2(1H)-pyridone.

Parallel studies on bassianin gave a structure differing only in that the C-3 acyl substituent is extended by an additional conjugated trans-substituted -CH=CH- unit positioned between C-7 and C-8 of the structure (I).

(Received, 17th December 1973; Com. 1696.)

¹ S. H. El Basyouni, D. Brewer, and L. C. Vining, Canad. J. Bot., 1968, 46, 441.

¹S. H. El Basyouni, D. Brewer, and L. C. Vining, Canad. J. Bol., 1906, 40, 441.
² R. T. Coutts and K. W. Hindmarsh, Org. Mass Spectrometry, 1969, 2, 681.
³ J. D. Roberts and C. Green, J. Amer. Chem. Soc., 1946, 68, 214.
⁴ G. B. Jackman, A. Robertson, R. B. Travers, and W. B. Whalley, J. Chem. Soc., 1958, 1825.
⁵ J. H. Noggle and R. E. Schirmer 'The Nuclear Overhauser Effect,' Academic Press, New York, 1971.
⁶ G. C. Levy and G. L. Nelson, 'Carbon-13 Nuclear Magnetic Resonance for Organic Chemists,' Wiley-Interscience, New York, 1972.
⁷ J. B. Stothers, 'Carbon-13 NMR Spectroscopy,' Academic Press, New York, 1972; (a) p. 288.
⁸ R. L. Lichter, 'Determination of Organic Structures by Physical Methods,' eds. F. C. Nachod and J. J. Zuckerman, Academic Press, New York, 1971.

New York, 1971, 4, p. 195.