

SHORT
COMMUNICATIONS

One-Pot Synthesis of 2-Aryl-4-methyl-3-nitro-2,3-dihydro-1,5-benzothiazepines

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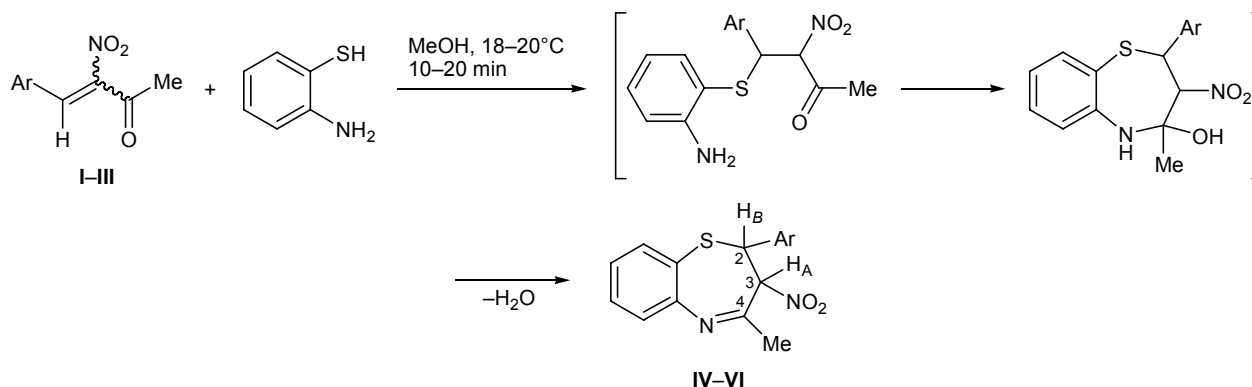
Wide application in medical practice of compounds whose molecules contain seven-membered thiazepine rings (examples are diltiazem which exhibits antian-ginal, hypotensive, and antiarrhythmic activity and neuroleptic quetiapine [1, 2]), strongly stimulated studies aimed at developing new methods of synthesis of thiazepine derivatives and extending their series.

With the goal of obtaining new nitro-substituted thiazepine structures we examined reactions of acces-sible and reactive 4-aryl-3-nitrobut-3-en-2-ones **I–III** [3] with *o*-aminobenzenethiol. Unlike other α,β -unsat-urated ketones which react with *o*-aminobenzenethiol under fairly drastic conditions (heating in boiling methanol for several hours in the presence of acid [4] or base catalyst [5] or heating in higher-boiling sol-vents, such as toluene, in the absence of catalyst [6]) to produce seven-membered heterocycles, reactions of nitro enones **I–III** with *o*-aminobenzenethiol occurred very readily at 18–20°C in methanol (without a cata-lyst) and were complete in 10–20 min. Crystalline 2-aryl-4-methyl-3-nitro-2,3-dihydro-1,5-benzothiazep-

ines **IV–VI** separated from the reaction solution, and their yields attained 98%. Presumably, the process follows nucleophilic addition pattern with subsequent heterocyclization of S-adducts.

Compounds **IV–VI** were isolated as colorless or yellowish crystalline substances which were diastereo-isomerically pure. Their structure was confirmed by spectral methods. The IR spectra of functionally sub-stituted 1,5-benzothiazepines **IV–VI** contained strong absorption bands due to stretching vibrations of uncon-jugated nitro group (1560, 1360 cm^{-1}), and $\text{C}=\text{N}$ stretching vibrations had a frequency of 1645 cm^{-1} .

The ^1H NMR spectra of **IV–VI** were consistent with the assumed structure. Compound **V** displayed in the ^1H NMR spectrum (CDCl_3) two clearly defined doublets from H_A and H_B at δ 5.45 and 5.25 ppm, respectively, with a coupling constant $^3J_{AB}$ of 11.90 Hz; protons in the methyl and methoxy groups resonated as singlets at δ 2.43 and 3.76 ppm, respectively; aromatic protons appeared as doublets at δ 7.11 and 6.79 ppm and multiplets at δ 7.57, 7.52, 7.29, and 7.20 ppm. In



I, IV, Ar = Ph; **II, V**, Ar = 4-MeOC₆H₄; **III, VI**, Ar = 4-Me₂NC₆H₄.

the ^{13}C NMR spectrum of **V** (recorded with decoupling from protons), the downfield signal at δ_{C} 162.68 ppm was assigned to the C=N carbon atom (C^4), and signals at δ_{C} 58.43 and 91.30 ppm were attributed to C^2 and C^3 . The assignment of signals in the ^1H and ^{13}C NMR spectra was proved by HMQC and HMBC correlation experiments.

Initial 4-aryl-3-nitrobut-3-en-2-ones **I–III** were synthesized according to the procedures reported in [3], and *o*-aminobenzenethiol was prepared as described in [7].

4-Methyl-3-nitro-2-phenyl-2,3-dihydro-1,5-benzothiazepine (IV). Yield 81%, colorless crystals, mp 136–137°C (from petroleum ether). IR spectrum, ν , cm^{-1} : 1645 (C=N); 1560, 1360 (NO_2). ^1H NMR spectrum (CDCl_3), δ , ppm: 2.44 s (3H, CH_3), 5.27 d (1H, H_B , $^3J_{AB} = 11.90$ Hz), 5.52 d (1H, H_A , $^3J_{AB} = 11.90$ Hz), 7.17–7.58 m (9H, H_{arom}). ^{13}C – $\{^1\text{H}\}$ NMR spectrum (CDCl_3), δ_{C} , ppm: 21.99 (CH_3), 58.70 (C^2), 90.98 (C^3); 120.50, 124.56, 126.39, 126.60, 129.02, 129.23, 131.04, 135.28, 139.89, 150.09 (C_{arom}); 162.76 (C=N). Found, %: C 64.61; H 4.91; N 9.15. $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_2\text{S}$. Calculated, %: C 64.41; H 4.73; N 9.39.

2-(4-Methoxyphenyl)-4-methyl-3-nitro-2,3-dihydro-1,5-benzothiazepine (V). Yield 98%, light yellow crystals, mp 132–134°C (from ethanol). IR spectrum, ν , cm^{-1} : 1645 (C=N); 1560, 1360 (NO_2). ^1H NMR spectrum (CDCl_3), δ , ppm: 2.43 s (3H, CH_3), 3.76 s (3H, OCH_3), 5.25 d (1H, H_B , $^3J_{AB} = 11.90$ Hz), 5.45 d (1H, H_A , $^3J_{AB} = 11.90$ Hz), 6.79 d and 7.11 d (2H each, C_6H_4); 7.20 t, 7.29 d, 7.52 t, and 7.57 d (4H, 6-H, 7-H, 8-H, 9-H). ^{13}C – $\{^1\text{H}\}$ NMR spectrum (CDCl_3), δ_{C} , ppm: 22.05 (CH_3), 55.38 (OCH_3), 58.43 (C^2), 91.30 (C^3); 114.49, 120.59, 121.68, 124.54, 126.54, 127.69, 130.93, 132.11, 135.24, 150.00, 159.90 (C_{arom}); 162.68 (C=N). Found, %: N 8.80. $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}_3\text{S}$. Calculated, %: N 8.53.

2-(4-Dimethylaminophenyl)-4-methyl-3-nitro-2,3-dihydro-1,5-benzothiazepine (VI). Yield 98%,

light orange crystals, mp 144–146°C (from ethanol). IR spectrum, ν , cm^{-1} : 1645 (C=N); 1560, 1360 (NO_2). ^1H NMR spectrum (CDCl_3), δ , ppm: 2.43 s (3H, CH_3), 2.91 s (6H, NCH_3), 5.23 d (1H, H_B , $^3J_{AB} = 11.90$ Hz), 5.46 d (1H, H_A , $^3J_{AB} = 11.90$ Hz), 6.57 d and 7.03 d (2H each, C_6H_4); 7.18 t, 7.28 d, 7.50 t, 7.58 d (4H, 6-H, 7-H, 8-H, 9-H). ^{13}C – $\{^1\text{H}\}$ NMR spectrum (CDCl_3), δ_{C} , ppm: 22.09 (CH_3), 40.41 (NCH_3), 59.01 (C^2), 91.46 (C^3); 112.41, 121.00, 124.44, 126.41, 126.29, 130.67, 135.30, 150.04, 150.70, 150.00 (C_{arom}); 162.74 (C=N). Found, %: N 12.01. $\text{C}_{18}\text{H}_{19}\text{N}_3\text{O}_2\text{S}$. Calculated, %: N 12.31.

The ^1H and ^{13}C NMR spectra, including HMQC and HMBC experiments, were recorded on a Jeol JNM-ECX400A spectrometer at 100.53 (^{13}C) and 399.78 MHz (^1H) using the residual solvent signal (CHCl_3) as internal reference. The IR spectra were measured from solutions in CHCl_3 ($c = 0.1$ – 0.001 M) on a Shimadzu IR Prestige-21 spectrometer with Fourier transform. The elemental compositions were determined on a EuroVector EA 3022 CHN Dual analyzer.

REFERENCES

1. Mashkovskii, M.D., *Lekarstvennye sredstva* (Drugs), Moscow: Novaya Volna, 2007, pp. 72, 433.
2. *Registr lekarstvennykh sredstv Rossii. Entsiklopediya lekarstv* (Drug Register of Russia. Encyclopedia of Drugs), RLS-2002, 2002, pp. 9, 298, 411.
3. Fel'gendler, A.V., Aboskalova, N.I., and Berestovitskaya, V.M., *Russ. J. Gen. Chem.*, 2000, vol. 70, p. 1087.
4. Khanna, M.S., Kumar, D., Garg, C.P., and Kapoor, R.P., *Indian J. Chem., Sect. B*, 1995, vol. 34, p. 333.
5. Orlov, V.D., Kolos, N.N., and Ruzhitskaya, N.N., *Khim. Geterotsikl. Soedin.*, 1983, p. 1638.
6. Gupta, A.K., Singh, V.K., and Pant, U.C., *Indian J. Chem., Sect. B*, 1983, vol. 22, p. 1057.
7. *Metody polucheniya khimicheskikh reaktivov i preparatov* (Methods for Preparation of Chemicals), Moscow: IREA, 1964, vol. 9, p. 26.