Mimicking multipass transmembrane proteins: synthesis, assembly and folding of alternating amphiphilic multiblock molecules in liposomal membranes[†][‡]

Takahiro Muraoka,^a Tatsuya Shima,^a Tsutomu Hamada,^b Masamune Morita,^b Masahiro Takagi^b and Kazushi Kinbara^{*a}

Received 7th July 2010, Accepted 27th August 2010 DOI: 10.1039/c0cc02420a

Alternating amphiphilic multiblock molecules 1–4, involving fluorescent hydrophobic units, were designed as mimics for multipass transmembrane proteins. Fluorescence spectroscopy of 1–4 in liposomal membranes suggested the face-to-face stacking of the hydrophobic units to give folded structures as well as intermolecular assemblies.

Synthetic molecules mimicking the structure and function of proteins have been attracting keen interest for development of molecular devices and medicines.¹ Membrane proteins are important targets, where several synthetic ion channels,² including tubes,³ ribbons,⁴ and barrels as mimics of β-barrel structures⁵ have been developed so far. On the other hand, multipass transmembrane (MTM) proteins remain mostly unexplored targets, although they provide one of the major structural motifs of the membrane proteins.⁶ MTM proteins usually consist of α -helices connected by hydrophilic residues, where their ternary structures are stabilized by the helix-helix interaction.⁷ In this communication, as simple structural mimics of MTM proteins, we report alternating amphiphilic multiblock molecules 1-4, consisting of linearly connected hydrophilic and hydrophobic moieties (Scheme 1). We found that the hydrophobic units of 1-4, 1,4-bis(4-phenylethynyl)benzene (BPEB) units, tend to form face-to-face stacking in the membrane, to possibly give folded structures⁸ like MTM proteins (Fig. 1).

The BPEB unit is known to fluoresce at 390 nm upon excitation around 320 nm.⁹ Since the face-to-face stacking of BPEB units results in a bathochromic shift of the emission band ($\lambda_{em} = 440$ nm), the assembling events of this unit are able to be monitored by means of fluorescence spectroscopy. BPEB units of 2–4 were connected by hydrophilic units composed of a benzoate group bearing two tetraethylene glycol chains. 1–4 were prepared by repeating Williamson synthesis between phenolic OH and TEG tosylate (see Supplementary Information‡). Finally, hydroxyl groups were

introduced to both termini of 1–4. 1–4 tend to be soluble in relatively polar organic solvents such as THF, DMF and DMSO, while hardly soluble in water and non-polar solvents such as hexane and CH_2Cl_2 .

First, we investigated the aggregation behavior of BPEB units of 1-4 in solution. The fluorescence spectra of 1-4 in THF (2.0 × 10⁻⁵ M) showed two emission peaks ($\lambda_{ex} = 313$ nm) in all cases, where no obvious difference was observed in the spectral profiles among them (λ_{em} ; 1: 367, 384 nm, 2: 366, 384 nm, 3: 367, 385 nm, 4: 367, 384 nm) (Fig. 2, blue lines). These emission profiles indicate that the face-to-face stacking of BPEB units hardly occurs in THF.9 In sharp contrast, in a THF-water (1/9) mixture, 1-4 (2.0 \times 10⁻⁵ M) showed red-shifted emission peaks in each case ($\lambda_{ex} = 313$ nm, $\lambda_{\rm em} =$ 1: 426, 451 nm, 2: 427, 450 nm, 3: 450 nm, 4: 428, 452 nm) (Fig. 2, red lines). Moreover, absorption spectra of 1-4 showed a hypsochromic shift in a THF-water (1/9) mixture compared with those in THF (see Supplementary Information Fig. S1[‡]). These spectral changes strongly indicate that BPEB units form cofacial planar H-aggregates in aqueous conditions. Dynamic light-scattering (DLS) analyses of 1-4 $(2.0 \times 10^{-5} \text{ M})$ in a THF-water (1/9) mixture revealed the presence of particles with average hydrodynamic diameters of ca. 100 nm in each case (see Supplementary Information Fig. S2[‡]).

The aggregation behavior of 1-4 in the liposomal membranes was then investigated using giant unilamellar liposomes of



 ^a Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, 2-1-1, Katahira, Aoba-ku, Sendai 980-8577, Japan. E-mail: kinbara@tagen.tohoku.ac.jp; Fax: +81 22-217-5612; Tel: +81 22-217-5612

^b School of Materials Science, Japan Advanced Institute of Science

and Technology, 1-1, Asahidai, Nomi, Ishikawa 923-1292, Japan

[†] This article is part of the 'Emerging Investigators' themed issue for ChemComm.

[‡] Electronic supplementary information (ESI) available: Synthesis, UV-Vis absorption spectra and profiles of dynamic light scattering analyses of 1–4, and preparation of giant liposomes. See DOI: 10.1039/c0cc02420a



Fig. 1 Plausible models for the assemblies of **1** and **4** in a liposomal membrane. (a) **1** and (c) **4** in dilute conditions. (b) **1** and (d) **4** in dense conditions.



Fig. 2 Fluorescence spectra of (a) **1**, (b) **2**, (c) **3** and (d) **4** in THF (blue) and THF–water = 1/9 (red) with excitation at 313 nm.

1,2-dioleoyl-*sn*-glycero-3-phosphocholine (DOPC), which can be directly observed by optical microscopy.¹⁰ Actually, phase-contrast microscopy (200 μ M DOPC containing 1–4, [1–4]/[DOPC] = 0.10) clearly visualized μ m-sized unilamellar liposomes in water (Fig. 3a–d). Of importance, these liposomes were also visualized by fluorescence microscopy monitoring at > 420 nm ($\lambda_{ex} = 330-385$ nm) (Fig. 3e–h), which demonstrates successful incorporation of 1–4 into the liposomal membranes under these conditions.

Fluorescence spectroscopy of DOPC liposomes prepared with 1 in water (200 μ M DOPC containing 1, [1]/[DOPC] = 0.00050) showed emission peaks at 366 and 384 nm (λ_{ex} = 315 nm, Fig. 4a). The spectral profile is quite similar to that of 1 in THF (Fig. 2a, blue line). Namely, at this concentration, 1 does not form the face-to-face assemblies in



Fig. 3 Optical micrographs of DOPC giant unilamellar liposomes containing (a), (e) 1, (b), (f) 2, (c), (g) 3 and (d), (h) 4 ([1–4]/[DOPC] = 0.10). (a)–(d) Phase-contrast and (e)–(h) fluorescence micrographs (λ_{ex} ; 330–385 nm, λ_{obsd} ; >420 nm). Scale bars: 10 µm.



Fig. 4 Fluorescence spectra of 200 μ M DOPC giant unilamellar liposomes in water containing (a) 1, (b) 2, (c) 3 and (d) 4 with excitation at 315 nm. [1–4]/[DOPC] = 0.00050, 0.0050, 0.050 and 0.10.

the membrane (Fig. 1a). However, of interest, when the amount of 1 was increased, broad emission appeared at >400 nm with a decrease in the intensity of the original emission. At molar ratio [1]/[DOPC] = 0.10, the spectrum showed the maximum emission at 426 nm. Hence, the BPEB unit of 1 is capable of assembling in the membrane to form face-to-face stacking at relatively high concentrations (Fig. 1b). In sharp contrast to 1, 4 in the membrane showed emission bands at 428 and 448 nm ($\lambda_{ex} = 313$ nm) even at very low concentration ([4]/[DOPC] = 0.00050, Fig. 4d). Furthermore, 200-fold increase in the amount of 4 ([4]/[DOPC] = 0.10) resulted in only a slight change in the spectral profile; the fluorescence exhibited little dependency on the concentration of 4. Thus, BPEB units of 4 are likely to form intramolecular face-to-face stacking, allowing 4 to adopt a folded structure like MTM proteins (Fig. 1c). However, slightly intensified emission at $\lambda_{\rm em}$ > 480 nm (Fig. 4d) also suggests a possible contribution of intermolecular stacking at higher concentrations ([4]/[DOPC] > 0.0050) (Fig. 1d). Interestingly, not only 4 but also 2 bearing two BPEB units showed emission at 426 nm (λ_{ex} = 313 nm) even at a highly diluted condition ([2]/[DOPC] = 0.00050), while 3 showed red-shifted emission (425 nm) only at high concentrations ([3]/[DOPC] = 0.050–0.10). This apparent even–odd effect may suggest formation of dimeric assemblies of BPEB units, which play some roles in folding and assembling of 1–4 in the membrane. It is worthy of note that in spite of intermolecular assembling of 1–4 in the membrane, no µm-scale domain structure like lipid rafts¹¹ was observed by fluorescence microscopy of the liposomes including 1–4.§

In conclusion, the BPEB unit was found to have strong capabilities to be involved in a liposomal membrane and form face-to-face stacking. These features could be applied for controlled folding of molecules as was found for alternating amphiphilic multiblock molecules **2** and **4**, which can be regarded as structural mimics of two- and four-transmembrane proteins, respectively. MTM proteins exhibit diverse functions, such as cell-adhesion of four-transmembrane proteins, potassium channels,¹³ and signal transduction of seven-transmembrane proteins, G proteins.¹⁴ The functions of **1–4** are now under investigation.

We thank Prof. M. Shimomura and Dr T. Higuchi for assistance in DLS measurements. We also thank Ms Yuko Kishimoto for support in preparation and microscopy of liposomes. This work was performed under the Cooperative Research Program of "Network Joint Research Center for Materials and Devices (Institute of Multidisciplinary Research for Advanced Materials, Tohoku University)", and partially supported by the Ministry of Education, Science, Sports and Culture, Japan, Grant-in-Aid for Young Scientists S and Asahi Glass Foundation to K. K.

Notes and references

§ Circular dichroism spectra of **2** and **4** gave no significant signal even at high concentration ([**2**, **4**]/[DOPC] = 0.10, [DOPC] = 200μ M) in a long light path length quartz cuvette (100 mm).

- H.-A. Klok, Angew. Chem., Int. Ed., 2002, 41, 1509;
 S. M. Butterfield and J. Rebek, Jr., J. Am. Chem. Soc., 2006, 128, 15366;
 O. Khakshoor, B. Demeler and J. S. Nowick, J. Am. Chem. Soc., 2007, 129, 5558;
 J. A. Robinson, Chimia, 2007, 61, 84.
- 2 A. L. Sisson, M. R. Shar, S. Bhosale and S. Matile, *Chem. Soc. Rev.*, 2006, **35**, 1269; A. P. Davis, D. N. Sheppard and B. D. Smith, *Chem. Soc. Rev.*, 2007, **36**, 348; G. W. Gokel and N. Barkey, *New J. Chem.*, 2009, **33**, 947.

- I. Tabushi, Y. Kuroda and K. Yokota, *Tetrahedron Lett.*, 1982, 23, 4601; J.-H. Fuhrhop, U. Liman and U. Koesling, *J. Am. Chem. Soc.*, 1988, 110, 6840; M. R. Ghadiri, J. R. Granja and L. K. Buehler, *Nature*, 1994, 369, 301; T. M. Fyles and C. C. Tong, *New J. Chem.*, 2007, 31, 655; M. Jung, H. Kim, K. Baek and K. Kim, *Angew. Chem., Int. Ed.*, 2008, 47, 5755.
- 4 M. A. Schmitt, B. Weisblum and S. H. Gellman, J. Am. Chem. Soc., 2004, 126, 6848.
- W.-Y. Yang, J.-H. Ahn, Y.-S. Yoo, N.-K. Oh and M. Lee, Nat. Mater., 2005, 4, 399; C. P. Wilson and S. J. Webb, Chem. Commun., 2008, 4007; N. Sakai, J. Mareda and S. Matile, Acc. Chem. Res., 2008, 41, 1354; A. Satake, M. Yamamura, M. Oda and Y. Kobuke, J. Am. Chem. Soc., 2008, 130, 6314; R. E. Dawson, A. Hennig, D. P. Weimann, D. Emery, V. Ravikumar, J. Montenegro, T. Takeuchi, S. Gabutti, M. Mayor, J. Mareda, C. A. Schalley and S. Matile, Nat. Chem., 2010, 2, 533.
- 6 P. J. F. Henderson, Curr. Opin. Cell Biol., 1993, 5, 708; E. Wallin and G. von Heijne, Protein Sci., 1998, 7, 1029; D. Oesterhelt, Curr. Opin. Struct. Biol., 1998, 8, 489; S. Subramaniam, Curr. Opin. Struct. Biol., 1999, 9, 462.
- 7 J. H. Perlman, A.-O. Colson, W. Wang, K. Bence, R. Osman and M. C. Gershengorn, J. Biol. Chem., 1997, 272, 11937; G. V. Nikiforovich, M. Zhang, Q. Yang, G. Jagadeesh, H.-C. Chen, L. Hunyady, G. R. Marshall and K. J. Catt, Chem. Biol. Drug Des., 2006, 68, 239; Molecular Biology of the Cell, ed. B. Alberts, A. Johnson, J. Lewis, M. Raff and K. Roberts, Garland Science, New York, USA, 2007, 5th edn, p. 632; W.-K. Lee, J. J. Han, B.-S. Jin, D. W. Boo and Y. G. Yu, Biochem. Biophys. Res. Commun., 2009, 390, 815; J.-Y. Shim, Biophys. J., 2009, 96, 3251.
- 8 Y. Zhao and J. S. Moore, *Foldamers: Structure, Properties, and Applications*, ed. S. Hecht and I. Huc, Wiley-VCH, Weinheim, Germany, 2007, pp. 75–108; J. J. van Gorp, J. A. J. M. Vekemans and E. W. Meijer, *Chem. Commun.*, 2004, 60.
- 9 M. Levitus, K. Schmieder, H. Ricks, K. D. Shimizu, U. H. F. Bunz and M. A. Garcia-Garibay, J. Am. Chem. Soc., 2001, **123**, 4259.
- T. Hamada, Y. Miura, K. Ishii, S. Araki, K. Yoshikawa, M. Vestergaard and M. Takagi, J. Phys. Chem. B, 2007, 111, 10853; K. Ishii, T. Hamada, M. Hatakeyama, R. Sugimoto, T. Nagasaki and M. Takagi, ChemBioChem, 2009, 10, 251; M. Morita, M. Vestergaard, T. Hamada and M. Takagi, Biophys. Chem., 2010, 147, 81.
- 11 J. F. Hancock, Nat. Rev. Mol. Cell Biol., 2006, 7, 456; T. Hamada, M. Morita, Y. Kishimoto, Y. Komatsu, M. Vestergaard and M. Takagi, J. Phys. Chem. Lett., 2010, 1, 170.
- 12 L. Elkouby-Naor and T. Ben-Yosef, Int. Rev. Cell Mol. Biol., 2010, 279, 1.
- 13 D. A. Doyle, J. M. Cabral, R. A. Pfuetzner, A. Kuo, J. M. Gulbis, S. L. Cohen, B. T. Chait and R. MacKinnon, *Science*, 1998, 280, 69.
- 14 K. L. Pierce, R. T. Premont and R. J. Lefkowitz, Nat. Rev. Mol. Cell Biol., 2002, 3, 639.