REACTION OF THE METHYL ESTER OF  $\alpha$ -METHOXYPERFLUOROPROPIONIC ACID WITH 2,6-DIMETHYLPHENOL

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 $\alpha$ -Chlorodialkyl ethers are efficient chloroalkylating agents [1]. The behavior of  $\alpha$ -fluoropolyfluorodialkyl ethers in this reaction has not been studied. In the reaction of the methyl ester of  $\alpha$ -methoxyperfluoropropionic acid (I) with 2,6-dimethylphenol (II), we showed that  $\alpha$ -fluoropolyfluorodialkyl ethers in the presence of a catalyst C-alkylate aromatic compounds. Thus, heating (I) and (II) in the presence of  $\text{Et}_2\text{O}\cdot\text{BF}_3$  (the mole ratio of these reagents was 2:1:0.3) at reflux for 4-5 h gave bisphenol (III) in 65% yield.

Methyl ester of  $\alpha,\alpha$ -bis(3,5-dimethyl-4-hydroxyphenyl)trifluoropropionic acid (III), mp 167-169°C (CCl<sub>4</sub>), R<sub>f</sub> 0.45 (1:3 acetone-CCl<sub>4</sub>). <sup>13</sup>C NMR spectrum ( $\delta$ , ppm, acetone, TMS): 170.14 (C=0), 154.04 (C<sup>4</sup>, C<sup>4</sup>'), 130.34 (C<sup>2</sup>, C<sup>2</sup>', C<sup>6</sup>, C<sup>6</sup>'), 127.70 (C<sup>1</sup>, C<sup>1</sup>'), 124.26 (C<sup>3</sup>, C<sup>3</sup>', C<sup>5</sup>, C<sup>5</sup>'), 67.26 (C\*-CF<sub>3</sub>, <sup>2</sup>J<sub>C-F</sub> = 31.50 Hz), 53.11 (OCH<sub>3</sub>), 16.71 (CH<sub>3</sub>). PMR spectrum in acetone-d<sub>6</sub> ( $\delta$ , ppm, TMS): 7.50 s (2H, 2OH), 6.85 s (4H, H<sup>2</sup>, H<sup>2</sup>', H<sup>6</sup>, H<sup>6</sup>'), 3.78 s (3H, OCH<sub>3</sub>), 2.30 s (12H, 4CH<sub>3</sub>). <sup>19</sup>F NMR spectrum in acetone-d<sub>6</sub> ( $\delta$ , ppm, CF<sub>3</sub>CO<sub>2</sub>H): -16.20 s. Found, %: C 62.56; H 5.53; F 14.75. C<sub>20</sub>H<sub>21</sub>F<sub>3</sub>O<sub>4</sub>. Calculated, %: C 62.83; H 5.50; F 14.92.

## LITERATURE CITED

1. R. C. Fuson, E. C. Horning, M. L. Ward, et al., J. Am. Chem. Soc., 64, 30 (1942).

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