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The reaction of unsymmetrical N,N-dialkylamidines with the esters of dialkylthioboric acids gives dialkylborylamidines (DBA), in which the boron is attached to the imino nitrogen of the amidine.

$$\begin{split} R_2BSR^3 + HN = & C \sqrt{\frac{R^1}{NR_2^2}} \xrightarrow{60-90^\circ} R_2B - N = & C \sqrt{\frac{R^1}{NR_2^2}} + R^3SH \\ R^3 = \textit{n-C}_3H_7 \text{ or } C_4H_6; \ R = R^1 = \textit{n-C}_3H_7, \ R^2 = C_8H_5 \ (I) \\ R = \textit{i-C}_3H_7, \ R^1 = C_6H_5, \ R^2 = CH_3 \ (I1) \end{split}$$

The following compounds were synthesized with this procedure: N,N-diethyl-N'-di-n-propylborylbutyramidine (I), 70% yield, bp 95-97° (3 mm), and N,N-dimethyl-N-diisopropylborylbenzamidine (II), 90% yield, bp 110-112° (2.5 mm). Compounds (I)-(II) are monomeric (by cryoscopy in benzene,  $^{11}B$  NMR), and their IR spectra contain an intense band at 1800 cm<sup>-1</sup>, which is characteristic for monomeric compounds that contain the >B-N=C < group [1].

The DBA can be obtained from the N-unsubstituted, monosubstituted and symmetrically disubstituted amidines by refluxing a mixture of the trialkylborane with the amidine in either THF or benzene (see [2] for the reaction of tri-n-propylborane with N,N'-diphenylacetamidine). Compound (III), with bp 119-121° (2.5 mm), which was obtained in 87% yield from triisopropylborane and N-isopropylbenzamidine, and compound (IV), with bp 151-154° (1 mm), which was synthesized in 80% yield from triisobutylborane and N-phenylbenzamidine, were isolated in the monomeric form. It is possible for (III) and (IV) to exist as the tautomers (A), (B), and (C).

$$R_{3}B + HN = C \begin{pmatrix} R^{1} & -RH & -R_{2}B - N = C & R^{1} & (A) \\ NHR^{2} & THF & -R_{2}B - N + C & R^{1} & (B) \\ -R_{2}B - N - C & NH & (C) & (C) \\ R^{2} & -R_{2}B - N - C & NH & (C) \\ R^{2} & R = R^{2} = i \cdot C_{3}H_{7}, R^{1} = C_{6}H_{5} \text{ (III)}; R : i \cdot C_{4}H_{9}, R^{1} = R^{2} = C_{6}H_{5} \text{ (IV)}$$

The presence in the IR spectrum of (III) of an intense band at  $1800 \text{ cm}^{-1}$  (>B-N=C<) and a relatively weak band at  $1650 \text{ cm}^{-1}$  ( N-C=N-) testifies to the predominance of tautomer (A). In contrast, in the spectrum of (IV) the  $1635 \text{ cm}^{-1}$  band is much more intense than the  $1805 \text{ cm}^{-1}$  band, and consequently the amount of tautomer (A) in the mixture is smaller than that of (B) or (C).

The purity of DBA (I)-(IV) was confirmed by the elemental analysis and NMR spectral data.

## LITERATURE CITED

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