

## 1,2- or 1,4-Addition in the Interaction of Quinolines and Organolithium Compounds

By C. E. CRAWFORTH\*

(Department of Chemistry and Biology, Harris College, Preston PR1 2TQ)

O. METH-COHN

(Department of Chemistry and Applied Chemistry, University of Salford, Salford M5 4WT)

and C. A. RUSSELL

(The Open University, Walton Hall, Walton, Bletchley, Bucks.)

**Summary** The reaction of 4-methylquinoline with organolithium reagents followed by ethyl chloroformate leads to *N*-ethoxycarbonyl-1,2-dihydroquinolines, and not the 1,4-analogues as recently postulated.

ADDITION reactions of pyridines<sup>1</sup> or quinolines<sup>2</sup> with organolithium reagents result in 2-substitution. Several workers have produced evidence in support of these reactions proceeding through 1,2-addition.<sup>3</sup> Recently, however, it

was proposed, largely from spectral data of the derivatives (Ia and b), that reactions between 4-methylquinoline and phenyl- or *n*-butyl-lithium proceeded *via* 1,4-addition followed by rearrangement.<sup>4</sup> We have repeated these experiments, obtaining compounds with identical m.p. and b.p.,<sup>4</sup> which are now reformulated as the corresponding 1,2-dihydro-derivatives (IIIa,b).

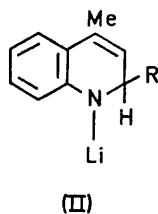
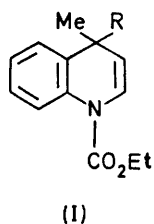
Authentic 1,2-dihydro-4-methylquinoline (IVc)<sup>5</sup> gave the *N*-ethoxycarbonyl-derivative (IIIC) when treated with ethyl

TABLE

*N.m.r. spectral data of (IIIa—c) and (IVa—c)*  
 $\tau$  (CDCl<sub>3</sub>; Me<sub>4</sub>Si)

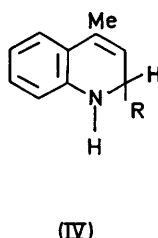
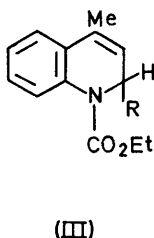
Compound	2-H	3-H	4-Me	Coupling constants (in Hz)		
				$J_{2,3}$	$J_{2,4-Me}$	$J_{3,4-Me}$
(IIIa)		3.96(m)	7.89(t)	—	—	—
(IIIa) <sup>a</sup>	1.56(m)	3.85(dq)	7.71(t)	6.4	1.0	1.0
(IIIb)	5.13(q)	4.20(dq)	7.98(q)	6.2	1.0	1.3
(IIIc)	5.71(m)	4.23(m)	7.97(q)	4.5	1.6	1.6
(IVa)		4.63(m)	8.02(t)	5.0	1.2	1.2
(IVb)	5.93(m)	4.66(m)	8.06(t)	5.0	1.5	1.5
(IVc)	6.12(m)	4.81(m)	8.23(q)	5.0	2.0	2.0

<sup>a</sup> With Eu(fod)<sub>3</sub> (0.25 mol. equiv.).



a, R = Ph ;

b, R = Bu<sup>n</sup>



a, R = Ph ;

b, R = Bu<sup>n</sup> ;

c, R = H

chloroformate. The relative positions, intensities, and multiplicities of the 2-H, 3-H, and 4-Me n.m.r. signals of these compounds clearly indicate that (IIIc) contains the 1,2-dihydroquinoline nucleus (see Table). The n.m.r. spectrum of the *n*-butyl derivative (Ib or IIIb) shows that the positions of the 2-H and 3-H signals are little affected by alkyl substitution. Decoupling experiments show that the 2-H signal is coupled to 3-H and the  $\alpha$ -protons of the 2-butyl group:  $\tau$  5.13 (q,  $J_{2,3}$  6.0,  $J_{2,\alpha}$  6.2 Hz), with smaller coupling to methyl  $J_{2,4-Me}$  1.0 Hz, and that due to 3-H,  $\tau$  4.20 (dq), is coupled to 3-H and 4-Me ( $J_{2,3}$  6.0,  $J_{3,4-Me}$  1.3 Hz). Structure (IIIb) is therefore indicated. The 2-H and 3-H signals of the phenyl derivative (Ia or IIIa) overlap. The large downfield shift of the 2-H signal is typical of that due to deshielding by an adjacent phenyl group<sup>6</sup> and indicates structure (IIIa). In the presence of the shift reagent<sup>7</sup> Eu(fod)<sub>3</sub> (fod: 1,1,1,2,2,3,3-heptafluoro-7,7-dimethyl-4,6-octanedionato-) however, the 2-H and 3-H signals were separated, that for 2-H to lowest field, producing a much simplified spectrum from which the respective coupling constants were easily obtained (see Table).

These structural assignments are further supported by the similarity between the spectra of the parent heterocycles (IVa,b), [prepared from 4-methylquinoline and phenyl- or n-butyl-lithium respectively, followed by careful aqueous hydrolysis] and reported analogues.<sup>8</sup> There

seems little doubt that addition of organolithium reagents to quinolines proceed *via* 1,2-addition.

(Received, 26th November 1971; Com. 2031.)

<sup>1</sup> K. Ziegler and H. Zeiser, *Ber.*, 1930, **63**, 1847; *Annalen*, 1931, **485**, 174.

<sup>2</sup> D. S. Tarbel, J. F. Bunnet, R. B. Carlin, and V. P. Wystrach, *J. Amer. Chem. Soc.*, 1945, **67**, 1582; R. A. Abramovitch, K. S. Ahmed, and C. S. Giam, *Canad. J. Chem.*, 1963, **41**, 1752.

<sup>3</sup> R. Foster and C. A. Fyfe, *Tetrahedron*, 1969, **25**, 1489; G. Fraenkel and J. G. Cooper, *Tetrahedron Letters*, 1968, 1825; C. S. Giam and J. L. Stout, *Chem. Comm.*, 1969, 142; K. Blaha and O. Cervinka, in 'Advances in Heterocyclic Chemistry', ed. A. R. Katritzky and A. J. Boulton, Academic Press, London, 1966, vol. 6, p. 222; R. A. Abramovitch and J. G. Saha, *ibid.*, p. 274, and references cited therein.

<sup>4</sup> Y. Otsuji, K. Yutani, and E. Imoto, *Bull. Chem. Soc. Japan*, 1971, **44**, 520.

<sup>5</sup> K. W. Rosenmund, F. Zymalkowski, and N. Schwartz, *Chem. Ber.*, 1954, **87**, 1229.

<sup>6</sup> 'An Introduction to Spectroscopic Methods for the Identification of Organic Compounds', ed. F. Scheinmann, Pergamon Press, Oxford, 1970, vol. 1, p. 59.

<sup>7</sup> 'Perkin Elmer NMR Quarterly', 1971, No. 1, p. 2.

<sup>8</sup> R. Bramley and M. D. Johnson, *J. Chem. Soc.*, 1965, 1372; T. W. Bunting and W. G. Meathrel, *Tetrahedron Letters*, 1971, 133; W. S. Johnson and B. G. Buell, *J. Amer. Chem. Soc.*, 1952, **74**, 4517.