Material Safety Data Sheet for Sodium Thiosulfate

Part 1 Chemical Name

Chinese name of the chemical: Sodium thiosulfate

Common name of the chemical: Sodium hyposulfite

English Name: sodium

Technical data sheet code: 106516 CAS No:

Company: Hainan Hua Yangshun Import and Export Co, Ltd.

Address: Longqiao town, Longhua district, Haikou, Hainan, China.

Phone: +86 18754130113

Part 2 Composition/Information on Ingredients

Ingredients Sodium thiosulfate content 99.999%

CAS No: 7772-98-7

Part 3 Hazards Identification

Hazard Category:

Invasion route: inhalation, skin contact, eye contact, ingestion

Environmental hazard: slight water pollutant

Explosion hazard: non-combustible material

Part 4 First Aid Measures

Skin contact: 1. Rinse with plenty of water for at least 20 minutes.

2. Remove contaminated clothing immediately

Eye contact: 1. Hold the upper and lower eyelids open and rinse with water for 20 minutes. 2. Notify an ophthalmologist and seek medical attention immediately.

Inhalation: Move the patient to fresh air.

Ingestion: Seek medical attention immediately.

Part 5 Firefighting Measures

Hazardous Characteristics: Non-combustible material

Harmful combustion products: No data Nitrogen Fire extinguishing method: Fire extinguisher

Part 6 Leakage Emergency Treatment

Emergency Treatment: 1. Before the contaminated area is completely cleaned, restrict personnel from entering the contaminated area.

Part 7 Operation and Storage

Operation Precautions: No special requirements Storage Precautions: Store in a closed container

Part 8 Contact Control/Personal Protection

| Engineering | Closed operation, provide adequate local exhaust. Provide safe | | |
|--|--|--|--|
| control: | showers and eyewash equipment. | | |
| Respiratory | When the dust concentration in the air is too high, it is | | |
| protection: | recommended to wear a filter-type dust respirator. | | |
| Eye protection: | Wear chemical safety goggles | | |
| Body protection: Protective clothing should be selected according to the concentration of harmful substances in the environment | | | |
| Hand protection: | and protection: Wear chemical protective gloves | | |
| Other protection: | Take off contaminated clothing as soon as possible after work | | |
| | | | |

Part 9 Physical and Chemical Properties

| Appearance and properties: | transparent monoclinic crystals. | | |
|----------------------------|----------------------------------|----------------------------------|----------------|
| pH: | 6. 0—7. 5 (25℃) | | |
| Melting point (℃): | 40~45 | Relative density (water = 1): | 1.729(11℃) |
| Boiling point | 100 | Relative vapor density(air = 1): | 1.01g/mLat 25° |
| Molecular formula: | Na2S2O3.5H2O | Molecular weight: | 248. 18 |
| Main | | | |

| Ingredients: | | | |
|---|--|------------------------------|---------|
| Saturated vapor pressure (kPa): | No data | Heat of combustion(kJ/mol): | No data |
| Critical temperature (°C): | No data | Critical pressure(MPa): | No data |
| Logarithm of octanol/water partition coefficient: | | | |
| Flash point (℃): | No data | Upper Explosion limit%(V/V): | No data |
| Ignition temperature (℃): | No data | lower Explosion limit%(V/V): | No data |
| Solubility: | Soluble in water and turpentine, hardly soluble in ethanol. | | |
| Main application: | It has a reducing effect. It is used as a photographic fixer, dechlorinating agent and analytical reagent, and is used in leather tanning and extracting silver from ores. It also has an anti-allergic effect. It is clinically used for skin pruritus, chronic urticaria, drug eruption, cyanide and arsenic poisoning, etc. | | |
| Other physical and chemical properties: | | | |

Section 10. Stability and reactivity

| Stability: | Stable Stable |
|------------|---------------|
| Stability: | Busic |

| Incompatible materials: | Nitrogen, peroxide, acid |
|-------------------------|---|
| Conditions to avoid | Reactive with nitrite and peroxide (explosion |
| contact: | hazard) |
| Polymerization hazard: | Does not occur |
| Decomposition products: | Sulfur oxides |
| | |

Part 11 Toxicological Information

| Acute toxicity: | Most of the ecological impact data for this substance are not available. |
|------------------|--|
| Carcinogenicity: | |

Part 12 Ecological Information

| Ecotoxicological | Most of the ecological impact data for this | |
|------------------------|---|--|
| toxicity: | substance are not available. | |
| Biodegradability: | No data | |
| Non-biodegradability: | No data | |
| Other harmful effects: | COD: 0. 322g/g | |
| | | |

Part 13 Disposal

| Waste nature: | Hazardous waste |
|--------------------|---|
| Waste disposal | Entrust the disposal to a nationally approved waste |
| method: | company |
| Waste precautions: | Disposal in accordance with current regulations |
| | |

Part 14 Transport Information

| Dangerous goods number: | 024 |
|-------------------------|------------------------|
| Packing method: | Double layer packaging |
| | |

Part 15 Regulatory Information

The 《Regulations on the Safety Management of Hazardous Chemicals》 (issued by the State Council on March 2, 2011) has made corresponding provisions for the safe production, use, storage, transportation, loading and unloading of hazardous chemicals.

Part 16 Other information

It weathers in dry air above 33°C, decomposes at 48°C, and decomposes into sodium sulfide and sodium sulfate when burned. The density of the anhydrous substance is 1.667. The aqueous solution reacts weakly alkaline. It decomposes in strong acid and precipitates sulfur and sulfur dioxide.