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## SODIUM SILICOFLUORIDE

# **Material Safety Data Sheet**

Date updated 25. Mar, 2018

## **CHEMICAL PRODUCT AND COMPANY IDENTIFICATION**

Date concerning producer

Wuhan Xinru Chemical Industry Co., ltd

Add: 2805 , Building 1, Fuxing huiyu Fuxing City(North Area),

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#### 1. Product Identification

Synonyms: sodium silicofluoride

CAS No.: 16893-85-9

Molecular Weight: 188.06 Chemical Formula: Na2SiF6

UN NO.:2674

## 2. Composition/Information on Ingredients

Ingredient CAS No Percent Hazardous Sodium Silicofluoride 16893-85-9 98.5min Yes

## 3. Hazards Identification

### **EMERGENCY OVERVIEW**

Toxic by inhalation and if swallowed. Harmful if in contact with skin.

### 4. First Aid Measures

**GENERAL ADVICE:** 

Show this safety data sheet to the

doctor in attendance

**INGESTION:** 

Drink plenty of water. Induce vomiting immediately and call a physician. If possible drink milk afterwards.

**EYE CONTACT:** 

In case of contact with eyes, rinse immediately with plenty of water and seek medical advise

### **INHALATION:**

Move to fresh air. Give oxygen. Consult a physician after significant exposure

### **SKIN CONTACT:**

Wash off immediately with soap and plenty of water removing all contaminated clothes and shoes. Consult a physician.

**PROTECTION OF FIRST AIDERS:** 

Ensure adequate ventilation. Use personal protective equipment.

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## 5. Fire Fighting Measures

SUITABLE EXTINGUISHING MEDIA: Non-flammable. Carbon dioxide (CO<sub>2</sub>), dry chemical EXTINGUISHING MEDIA NOT SUITABLE: Water.

SPECIFIC FIRE-FIGHTING HAZARDS: Hydrogen fluoride can be produced during a fire.

**SPECIFIC METHODS: Standard procedure for chemical fires.** 

PROTECTION FOR FIRE FIGHTERS: In case of fire, wear a self-contained breathing apparatus.

#### 6. Accidental Release Measures

PERSONAL PRECAUTIONS: Ensure adequate ventilation. Use personal protective equipment. Evacuate personnel to safe areas.

ENVIRONMENTAL PRECAUTIONS: Prevent product from entering drains. METHODS FOR CLEANING UP: Shovel into suitable container for disposal.

## 7. Handling and Storage

#### **TECHNICAL MEASURE TO PREVENT USER EXPOSURE**

Use only in area provided with appropriate exhaust ventilation. Avoid dust formation.

SAFE HANDLING PRECAUTIONS Wear personal protective equipment (eyes).

#### **TECHNICAL STORAGE MEASURES**

25kg or 50kg woven polypropylene bags with waterproof inner liners.

#### STORAGE CONDITIONS

Keep containers tightly closed in a dry, cool and well ventilated area.

**INCOMPATIBLE PRODUCTS Storage category A (away from acids).** 

#### 8. Exposure Controls/Personal Protection

#### APPROPRIATE ENGINEERING MEASURES

Ensure adequate ventilation, especially in confined areas.

**SPECIFIC CONTROL PARAMETER (EXPOSURE LIMITS)** 

OSHA Permitted exposure level (PEL Final) 8 hr TWA 2.5mg/m

**PPE: RESPIRATORY PROTECTION** 

In case of respirable dust and/or fumes, use a self-contained breathing

apparatus. PPE: HAND Rubber or plastic gloves.

PPE: EYE Tightly fitting safety goggles

**PPE: SKIN AND BODY** 

Dust impervious protective suit. Remove and wash contaminated clothing before re-use

**HYGIENE MEASURES** 

Avoid contact with skin, eyes and clothing. When using - do not eat, drink or smoke. Wash hands before breaks and immediately after handling the product.

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## 9. Physical and Chemical Properties

FORM: COLOUR: pH:

Crystalline. White powder. Between 6 and 7.

ODOR: FLASH POINT: DECOMPOSITION

Slightly pungent. Does not flash. TEMPERATURE:

Starts at 500°C.

MOLECULAR WEIGHT: RELATIVE DENSITY: SOLUBILITY:

188.06 2.68 Water: 0.64g/100g water at

20°C

TYPICAL POUR DENSITY: TYPICAL TAP DENSITY:

3 1340kg/m 1590kg/m

## 10. Stability and Reactivity

STABILITY: CONDITIONS TO AVOID:

Stable at normal conditions. To avoid thermal decomposition, do

Decomposition starts at 500°C. not overheat.

HAZARDOUS DECOMPOSITION MATERIALS TO

PRODUCTS: AVOID: Metals.

Sodium Fluoride, Silicon Tetra fluoride and Hydrogen fluoride

## 11. Toxicological Information

ACUTE TOXICITY: SENSITIZATION: LD50/oral/rat = 125 mg/kg. No data available.

Irritation of skin: mild/500 mg/rabbit
Irritation of eyes: severe/100 mg/rabbit

LOCAL EFFECTS: CHRONIC TOXICITY:

**Ingestion may cause gastrointestinal irritation,** Continuous or intermittent exposure to inorganic fluorides

nausea, vomiting and diarrhoea. Inhalation of can lead to appreciable accumulation of fluoride in

dust may cause shortness of breath, tightness of and, to development of osteosclerosis and

other bone

the chest, a sore throat and cough. changes.

**HUMAN EXPERIENCE:** 

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No evidence of association between fluoride ingestion and mortality from cancer in humans. (Internal Agency of Research on Cancer).

## 12. Ecological Information

MOBILITY: ECOTOXICITY: BIOACCUMULATION: No data available. No data available.

PERSISTENCE AND FURTHER INFORMATION:
DEGRADABILITY: Highly toxic to foliage.

No data available. Lethal to Tinca Vulgaris at 50 ppm.

## 13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste facility. Although not a listed RCRA hazardous waste, this material may exhibit one or more characteristics of a hazardous waste and require appropriate analysis to determine specific disposal requirements. Processing, use or contamination of this product may change the waste management options.

## 14. Transport Information

ADR/RID: 6.1 Labels: Harmful Proper shipping name:

Sodium fluorosilicate

IMDG: Labels: Harmful Proper shipping name:
Class: 6.1 Subrisk: None Sodium fluorosilicate

Packaging group: III

IATA: Labels: Harmful Proper shipping name: Class: 6.1 Subrisk: None Sodium fluorosilicate

Packaging group: III

## 15. Regulatory Information

T: Toxic

R23/24/25 : Toxic by inhalation, in contact with skin and if

swallowed. S1/2: Keep locked-up and out of reach of children.

S26: In case of contact with eyes, rinse immediately with plenty of water and seek medical advise. S44: If you feel unwell, seek medical advice (show label where possible).

## 16. Other Information

#### **RECOMMENDED USE:**

- 1. Fluoridation of drinking water.
- 2. Laundry sours.
- 3. Opalescent glass.

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- 4. Vitreous enamel frits.
- 5. Metallurgy of aluminium and verblium.
- 6. Insecticides and rodenticides.
- 7. Leather and wood preservatives.

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