Certificate of Analysis_(Ver.2.0)

Fumonisin B₁,B₂ in Acetonitrile/Water(1/1)

1. General information

This document is designed and the certified value(s) and uncertainty(ies) are determined in accordance with ISO Guide 31^[1].

2. Description of the Reference Material (RM)

Product name: Fumonisin B₁,B₂ in Acetonitrile/Water(1/1)

Product number: STD#2011
CAS number: 2B00C01

Formula: Fumonisin B_1 :116355-83-0;Fumonisin B_2 :116355-84-1;

Formula weight: Fumonisin B₁:C₃₄H₅₉NO₁₅; Fumonisin B₂:C₃₄H₅₉NO₁₄

 $Lot\#: Fumonisin \ B_1:721.83; Fumonisin \ B_2:705.83$

 $Result\ concentration: \qquad FumonisinB_1:49.99\pm0.90\mu g/mL; \qquad FumonisinB_2:50.13\pm1.00\mu g/mL$

Starting material: FB₁,lot#18102Z,Pribolab Pte. Ltd.; FB₂,lot#17103J,Pribolab Pte. Ltd.;

Matrix: Acetonitrile, LiChrosolv®, Merck

Amount: 5.0mL

Production date: 02,Mar,2022
Expiry date: 01,Sep,2023

Name of the supplier: Pribolab Pte. Ltd.

2.1 Intended use of the RM

- for laboratory use only
- calibration of analytical instruments

2.2 Instruction for the correct use of the RM

The compound should be stored at 2-8°C in a dark place. Before usage of the RM,the compound should be allowed to warm to temperature(20±3°C). The recommended minimum sub - sample amount for all kinds of application is 100 μ L. The expiry date of this RM is based on the current knowledge and holds only for proper storage conditions in the originally closed flasks/ packages.

2.3 Hazardous situation

The normal laboratory safety precautions should be observed when working with this RM.Further details for the handing of this RM are available as safety data sheet.

Hazardous IngredientsConcentration in%PictogramsSignal wordHazard statement(s)Acetonitrile>50DangerH225,H302,H312,H319,H332

3. Certified values and their uncertainties

Fumonisin B ₁ ,B ₂ in Acetonitrile/Water(1/1)				
	Compound	Mass concentration ^a		
		Certified value ^b	Uncertainty ^c	
	Fumonisin B ₁	49.99μg/mL	±0.90μg/mL	
	Fumonisin B ₂	50.13μg/mL	±1.00μg/mL	
а	Mass concentration based on weighed amount ,purity and dilution steps			
)	Values are based on preparation data and confirmed experimentally by HPLC-FLD			
С	Expanded uncertainty U(k=2) of the value u _c ac	ccording to GUM ^[2]		

3.1 Calculation of uncertainty

After the concentration of the gravimetric prepared solution was confirmed by HPLC-FLD, the uncertainty of the calibrant was calculated on the basis of preparation^[3].

Uncertainty components	Description	Standard uncertaint	Standard uncertainty _(U)	
Purity(P)of solid Fumonisin B ₁ Fumonisin B ₂	P ₁ =98.5±1.5% P ₂ =98.2±1.8%	u(P ₁)= <mark>0.9</mark> % u(P ₂)=1.0%	а	
Weighing procedure; Weighted sample: mFumonisin B1=5.075mg mFumonisin B2=5.105mg	U _(m) =0.0000008g+1.30*10 ⁻⁵ *m _{Toxin} u _(m) =U _(m) /2	u _(m) =0.0004mg	b	
Dilution procedure Volumetric flask :V _f =100mL	Calibration:100mL ± 0.1mL Repeatability : 0.03mL Volume expansion solvent	u(cal)=0.04mL u(rep)=0.03mL u(Vol.exp.1)=0.24mL u(v)=0.25mL	c d e f	

- a Maximum tolerance of purity was divided by $\sqrt{3}$
- b Calculation of this u-value is based upon the uncertainty formula for the weighed amount as given in the calibration report from annual balance calibration
- c A triangular distribution(division by $\sqrt{6}$)was chosen for the calculation of u(cal)
- d Based on a series of ten fill and weigh experiments on a typical 100mL flask; the value was used directly as a standard deviation
- e Based on the density of 0.7857 g/cm³ at temperature T=20 °C and a maximum temperature variation of ± 3 °C, of volume expansion, relative volume expansion coefficient of acetonitrile is 1370*10-6/°C[7], volume expansion term(rectangular distribution) was divided by $\sqrt{3}$
- f The three contributions are combined to give the u(V)= $\sqrt{u(cal)^2 + u(rep)^2 + u(Volexp)^2}$

Calculation of the combined uncertainty uc and the expanded standard uncertainty U

$$C_{Toxin} = \frac{10 \times m_{ws} \times P_{1}}{V_{f}} = \frac{10 \times 5.075 \times 98.5}{100} = 49.99 \,\text{mg} / L$$

$$\frac{\mathbf{u}_{c}(C_{Toxin})}{\mathbf{c}_{Toxin}} = \sqrt{\left[\frac{u(P)}{P}\right]^{2} + \left[\frac{u(m)}{m_{ws}}\right]^{2} + \left[\frac{u(V)}{V_{f}}\right]^{2}} = \sqrt{\left[\frac{0.9}{98.5}\right]^{2} + \left[\frac{0.0004}{5.075}\right]^{2} + \left[\frac{0.25}{100}\right]^{2}} = 0.009$$

$$u_{c}(C_{Toxin}) = C_{Toxin} \times 0.009 = 49.99 \times 0.009 = 0.45 \,\text{mg} / L$$

calculation of expanded standard uncertainty U using a coverage factor k=2

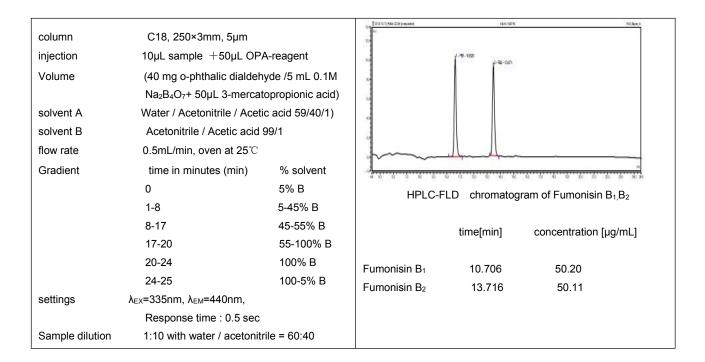
$$U(\mathbf{c}_{Toxin}) = \mu_c(\mathbf{c}_{Toxin}) \times 2 = 0.45 \times 2 = 0.90 \mu g / mL$$

4. Discussion of traceability

This calibrant is certified on the basis of gravimetric preparation^[4]. Thus the certified value(mass concentration of Fumonisin B₁,B₂ are based on the weighed amount of the starting material and is therefore traceable to the stated purity of the solid raw material. High purity material represents a practical realization of concentration units, through conversion of mass to molar quantity.

5. Confirmation of certified value by HPLC-FLD

The certified concentration of Fumonisin B_1,B_2 of the gravimetric prepared solution were confirmed by HPLC-FLD against an independently prepared reference batch of Fumonisin B_1,B_2 .



6. Further information

The purchaser must determine the suitability of this product for its particular use. Pribolab makes no warranty of any kind, express or implied, other than its products meet all quality control standards set by Pribolab. We do not guarantee that the product can be used for a special application.

Inspected by

Quality System Specialist

References:

- [1] ISO Guide 31, 1-7, (2000), "Reference Materials Contents of Certificates and Labels"
- [2] International Organization for Standardization (ISO), (2008), "Guide to the Expression of Uncertainty in Measurements", (GUM 1995 with minor corrections) 1st Ed. Geneva, Switzerland
- [3] R.D. Josephs, R. Krska, S. MacDonald, P. Wilson, H. Pettersson, J. AOAC Int. 86, 50-60. (2003), "Preparation of a Calibrant as Certified Reference Material for Determination of the Fusarium Mycotoxin, Zearalenone"
- [4] E.W. Flick, (1998), "Industrial Solvents Handbook ",5rd Ed., Noyes Data Corp. Westwood NJ