Certificate of Analysis(Ver.2.0) U-[¹³C₁₇]-Mycophenolic Acid in Acetonitrile

1. General information

This document is designed and the certified value(s) and uncertainty(ies) are determined in accordance with ISO Guide 31[1].

2. Description of the Reference Material (RM)

Name:	U-[¹³ C ₁₇]-Mycophenolic Acid in Acetonitrile	H H [13] C H H-C-H
Catalog number:	STD#3191U	H [13] C [13] C [13] H
CAS number:	1202866-92-9	
Formula:	¹³ C ₁₇ H ₂₀ O ₆	
Formula weight:	337.21	O √(13) C (13) H C (13) H
Lot #:	2B00A26	н
Starting material :	U-[¹³ C ₁₇]-Mycophenolic Acid,lot#l19906P,Pr	ibolab Pte.Ltd.
Solvent:	Acetonitrile,LiChrosolv [®] ,Merck	
Amount:	1.2mL	
Production date:	26/Jan/2022	
Expiry date:	25/Jul/2023	
Name of the supplier:	Pribolab Pte.Ltd.	

2.1 Intended use of the RM

- for laboratory use only
- internal standard[2]

2.2 Instruction for the correct use of the RM

The compound should be stored at $2-8^{\circ}$ C in a dark place.Before usage of the RM,the compound should be allowed to warm to temperature($20\pm3^{\circ}$ C).The recommended minimum sub-sample amount for all kinds of application is 100 µL.The expiry date of this RM is based on the current knowledge and holds only for proper storage conditions in the originally closed flasks/packages.

2.3 Hazardous situation

The normal laboratory safety precautions should be observed when working with this RM.Further details for the handing of this RM are available as safety data sheet.

Hazardous IngredientsConcentration in%Acetonitrile>99.9



Signal word Danger Hazard statement(s) H225,H302,H312,H319,H332

3. Certified values and their uncertainties

	U-[¹³ C ₁₇]-Mycophenolic	Acid in Acetonitrile	
	Compound	Mass conc	entration ^a
	U-[¹³ C ₁₇]-Mycophenolic Acid,96.27 atom% ¹³ C	Certified value ^b	Uncertainty ^c
Ľ		0.507µg/mL	±0.007µg/mL
а	Values are based on preparation data and conf	irmed experimentally by	HPLC-DAD
b	Mass concentration based on weighed amount	,purity and dilution step	
С	Expanded uncertainty U(k=2) of the value uc ac	cording to GUM[3]	

3.1 Calculation of uncertainty

The uncertainty of the calibrant solution was calculated on the basis of preparation [4].

Uncertainty components	Description	Standard uncertainty(U)	
Purity (P) of solid $U-[^{13}C_{17}]$ -MycophenolicAcid,96.27atom% ^{13}C Acid,96.27	P=98.9±1.1%	u(P)=0.6%	а
Weighing procedure Weighted sample: m _{ws} =1.282mg	U _(m) =0.0000008g+1.30*10 ⁻⁵ *m _{Toxin} u _(m) =U _(m) /2	u _(m) =0.0004mg	b
	calibration flask1: 250mL±0.15mL repeatability flask1: 0.03mL volume expansion solvent flask1	u(cal1)=0.06mL u(rep1)=0.03mL u(Vol.exp.1)=0.59mL u(V1)=0.59mL u(cal2)=0.06mL u(rep2)=0.03mL u(Vol.exp.2)=0.59mL u(V2)=0.59mL	c d e f
Dilution procedure Volumetric flask1:V _{f1} =250mL Volumetric flask2:V _{f2} =250mL One-mark glass pipette:V _P =25mL	calibration flask2: 250mL±0.15mL repeatability flask2: 0.03mL volume expansion solvent flask2		g h i j
	calibration pipette: 25mL±0.03mL volume expansion solvent pipette	u(cal3)=0.01mL u(Vol.exp.3)=0.06mL u(V3)=0.06mL	k I m

a Maximum tolerance of purity was divided by $\sqrt{3}$

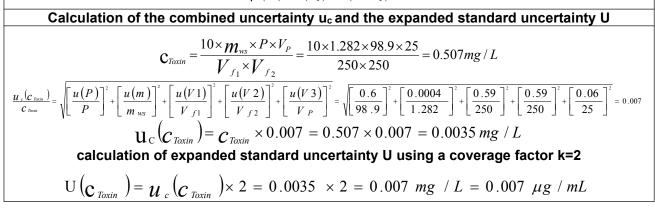
b Calculation of this u-value is based upon the uncertainty formula for the weighed amount as given in the calibration report from annual balance calibration

c,g,k A triangular distribution (division by $\sqrt{6}$ $\,$) was chosen for the calculation of u(cal)

d,h Based on a series of ten fill and weigh experiments on a typical 250mL flask; the value was used directly as a standard deviation

e,i,I Based on the density of 0.7857 g/cm³ at temperature T=20°C and a maximum temperature variation of ±3°C, of volume expansion, relative volume expansion coefficient of acetonitrile is 1370*10⁻⁶/°C[5], volume expansion term(rectangular distribution) was divided by $\sqrt{3}$

f,j,m The three contributions are combined to give the $u(V) = \sqrt{u(cal)^2 + u(rep)^2 + u(Vol.exp)^2}$



Compound	Isotopic distribution	
U-[¹³ C ₁₇]-Mycophenolic Acid		
U-[¹³ C ₁₆]-Mycophenolic Acid	9.47%	
U-[¹³ C ₁₅]-Mycophenolic Acid	17.59%	
U-[¹³ C ₁₄]-Mycophenolic Acid	2.44%	
U-[¹³ C ₁₃]-Mycophenolic Acid	2.84%	
Iculated isotopic enrichment level ^a :96.27 ato	m % ¹³ C	

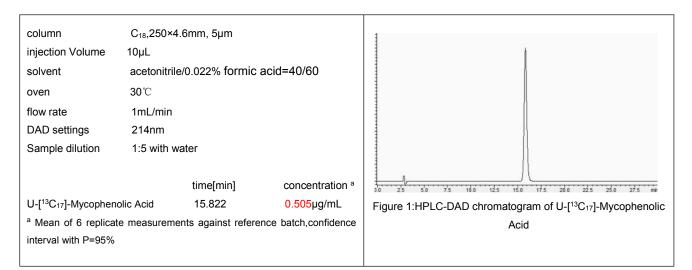
4.Isotopic enrichment and isotope pattern

5. Discussion of traceability

This calibrant is certified on the basis of gravimetric preparation [5]. Thus the certified value(mass concentration of U-[¹³C₁₇]-Mycophenolic Acid,96.27 atom% ¹³C is based on the weighed amount of the starting material and is therefore traceable to the stated purity of the solid raw material.High purity material represents a practical realization of concentration units,through conversion of mass to molar quality.

6. Confirmation of certified value by HPLC-DAD

The certified concentration of U-[$^{13}C_{17}$]-Mycophenolic Acid,96.27 atom% ^{13}C of the gravimetric prepared solution was confirmed by HPLC-DAD against an independently prepared reference batch of unlabeled Mycophenolic Acid .



7. Further information

The purchaser must determine the suitability of this product for its particular use. Pribolab makes no warranty of any kind, express or implied, other than its products meet all quality control standards set by Pribolab. We do not guarantee that the product can be used for a special application.

Inspected by

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Quality System Specialist

References:

- [1]ISO Guide 31:2015 1-18, "Reference materials contents of certificates, labels and accompanying documentation"
- [2]G. Häubl, F. Berthiller, R. Krska, R. Schuhmacher, "Suitability of a fully ¹³C isotope labelled internal standard for the determination of the mycotoxin deoxynivalenol by LC-MS/MS without clean-up", Anal. Bioanal. Chem. 384 (3), (2006), 692-696
- [3] International Organization for Standardization (ISO), (2008), "Guide to the expression of uncertainty in measurement", (GUM 1995 with minor corrections) 1st Ed. Geneva, Switzerland
- [4] R.D. Josephs, R. Krska, S. MacDonald, P. Wilson, H. Pettersson, J. AOAC Int. 86, 50-60, (2003), "Preparation of a Calibrant as Certified Reference Material for Determination of the Fusarium Mycotoxin Zearalenone"
- [5] E.W. Flick, (1998), "Industrial Solvents Handbook", 5th Ed., Noyes Data Corp. Westwood NJ