# **Certificate of Analysis**<sub>(Ver.1.0)</sub>

# $U-[^{13}C_{36}]$ -Enniatin A,U-[ $^{13}C_{35}$ ]-Enniatin A<sub>1</sub>,

# U-[¹³C₃₃]-Enniatin B,U-[¹³C₃₄]-Enniatin B₁ in Acetonitrile

#### 1. General information

This document is designed and the certified value(s) and uncertainty(ies) are determined in accordance with ISO Guide 31[1].

### 2. Description of the Reference Material (RM)

Name:  $U-[^{13}C_{36}]$ -Enniatin A,U- $[^{13}C_{35}]$ -Enniatin A<sub>1</sub>,U- $[^{13}C_{33}]$ -Enniatin B,U- $[^{13}C_{34}]$ -Enniatin B<sub>1</sub> in

**Acetonitrile** 

Catalog number: STD#3219U

CAS number: U-[13C36]-Enniatin A:2503-13-1(unlabeled);U-[13C35]-Enniatin A:4530-21-6(unlabeled);

U-[13C33]-Enniatin B:917-13-5(unlabeled);U-[13C34]-Enniatin B1:19914-20-6(unlabeled)

Formula:  $U-[^{13}C_{36}]$ -Enniatin A: $^{13}C_{36}H_{63}N_3O_9$ ;  $U-[^{13}C_{35}]$ -Enniatin A: $^{13}C_{35}H_{61}N_3O_9$ ;

 $\text{U-[$^{13}$C$_{33}]-Enniatin B:$^{13}$C$_{33}$H$_{57}$N$_3$O$_{9};$U-[$^{13}$C$_{34}]-Enniatin B$_{1:}$^{13}$C$_{34}$H$_{59}$N$_3$O$_{9}}$ 

Formula weight: U-[13C36]-Enniatin A:717.90;U-[13C35]-Enniatin A:702.87;

 $\label{eq:U-[13C33]-Enniatin B:672.82; U-[13C34]-Enniatin B_1:687.85} \\$ 

Lot #: 2A01A16

Starting material: U-[13C<sub>36</sub>]-Enniatin A:lot#J200110P,Pribolab Pte. Ltd.;

U-[13C35]-Enniatin A1:lot#J200110P,Pribolab Pte. Ltd.;

U-[13C33]-Enniatin B,lot#J200110P,Pribolab Pte. Ltd.;

U-[13C34]-Enniatin B1,lot#J200110P,Pribolab Pte. Ltd.

Solvent: Acetonitrile, LiChrosolv®, Merck

Amount: 1.2mL

Production date: 16/Nov/2021

Expiry date: 16/Nov/2022

Name of the supplier: Pribolab Pte.Ltd.

#### 2.1 Intended use of the RM

- for laboratory use only

- internal standard[2]

#### 2.2 Instruction for the correct use of the RM

The compound should be stored at 2-8°C in a dark place.Before usage of the RM,the compound should be allowed to warm to temperature( $20\pm3$ °C).The recommended minimum sub-sample amount for all kinds of application is 100 µL.The expiry date of this RM is based on the current knowledge and holds only for proper storage conditions in the originally closed flasks.

#### 2.3 Hazardous situation

The normal laboratory safety precautions should be observed when working with this RM.Further details for the handing of this RM are available as safety data sheet.

Hazardous IngredientsConcentration in%PictogramsSignal wordHazard statement(s)Acetonitrile>99.9DangerH225,H302,H312,H319,H332

#### 3. Certified values and their uncertainties

$U-[^{13}C_{36}]$ -Enniatin A,U- $[^{13}C_{35}]$ -Enniatin A <sub>1</sub> ,U- $[^{13}C_{33}]$ -Enniatin B,U- $[^{13}C_{34}]$ -Enniatin B <sub>1</sub> in Acetonitrile			
Compound	Mass concentration <sup>a</sup>		
	Certified value b	Uncertainty <sup>c</sup>	
U-[ <sup>13</sup> C <sub>36</sub> ]-Enniatin A, 99.15 atom% <sup>13</sup> C	2.02µg/mL	±0.04µg/mL	
U-[13C35]-Enniatin A1,98.86 atom% 13C	2.03µg/mL	±0.04µg/mL	
U-[ <sup>13</sup> C <sub>33</sub> ]-Enniatin B,98.64 atom% <sup>13</sup> C	2.12µg/mL	±0.03µg/mL	
U-[ <sup>13</sup> C <sub>34</sub> ]-Enniatin B <sub>1</sub> ,98.52 atom% <sup>13</sup> C	2.12µg/mL	±0.04μg/mL	

- a Values are based on preparation data and confirmed experimentally by HPLC-DAD
- b Mass concentration based on weighed amount ,purity and dilution step
- c Expanded uncertainty U(k=2) of the value u<sub>c</sub> according to GUM[3]

# 4.Isotopic enrichment and isotope pattern

Isotope pattern <sup>a</sup>			
Compound	Isotopic distribution	Compound	Isotopic distribution
U-[13C36]-Enniatin A	73.10%	U-[ <sup>13</sup> C <sub>35</sub> ]-Enniatin A <sub>1</sub>	66.23%
U-[ <sup>13</sup> C <sub>35</sub> ]-Enniatin A	23.25%	U-[13C34]-Enniatin A1	27.81%
U-[ <sup>13</sup> C <sub>34</sub> ]-Enniatin A	3.65%	U-[¹³C₃₃]-Enniatin A₁	5.96%
U-[13C33]-Enniatin B	63.29%	U-[¹³C₃₄]-Enniatin B₁	59.88%
U-[13C32]-Enniatin B	28.48%	U-[¹³C₃₃]-Enniatin B₁	29.94%
U-[ <sup>13</sup> C <sub>31</sub> ]-Enniatin B	8.23%	U-[¹³C₃₂]-Enniatin B₁	10.18%

**Calculated isotopic enrichment level** <sup>a</sup>:U-[<sup>13</sup>C<sub>36</sub>]-Enniatin A **99.15 atom%** <sup>13</sup>C;U-[<sup>13</sup>C<sub>35</sub>]-Enniatin A<sub>1</sub> **98.86 atom%** <sup>13</sup>C;U-[<sup>13</sup>C<sub>33</sub>]-Enniatin B **98.64 atom%** <sup>13</sup>C;U-[<sup>13</sup>C<sub>34</sub>]-Enniatin B<sub>1</sub> **98.52 atom%** <sup>13</sup>C

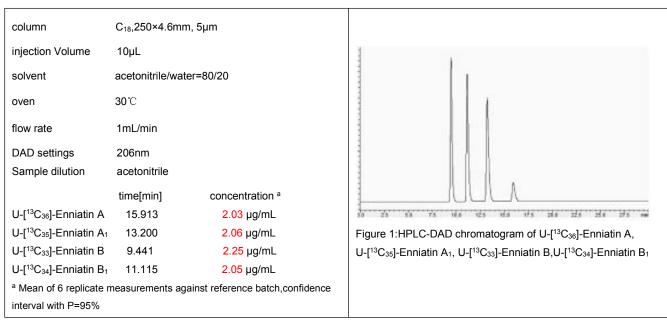
<sup>&</sup>lt;sup>a</sup> Approximation based on LC-MS/MS data

## 5.Discussion of traceability

This calibrant is certified on the basis of gravimetric preparation [4]. Thus the certified value(mass concentration of U-[ $^{13}C_{36}$ ]-Enniatin A, U-[ $^{13}C_{35}$ ]-Enniatin A<sub>1</sub>, U-[ $^{13}C_{33}$ ]-Enniatin B,U-[ $^{13}C_{34}$ ]-Enniatin B<sub>1</sub> is based on the weighed amount of the starting material and is therefore traceable to the stated purity of the solid raw material. High purity material represents a practical realization of concentration units, through conversion of mass to molar quantity.

## 6. Confirmation of certified value by HPLC-DAD

The certified concentration of the gravimetric prepared solution was confirmed by HPLC-DAD against an independently prepared reference batch of unlabeled Enniatin A, Enniatin B,Enniatin B<sub>1</sub>.



#### 7. Further information

The purchaser must determine the suitability of this product for its particular use. Pribolab makes no warranty of any kind, express or implied, other than its products meet all quality control standards set by Pribolab. We do not guarantee that the product can be used for a special application.

Inspected by

Quality System Specialist

### References:

- [1]ISO Guide 31:2015 1-18, "Reference materials contents of certificates, labels and accompanying documentation"
- [2]G. Häubl, F. Berthiller, R. Krska, R. Schuhmacher, "Suitability of a fully <sup>13</sup>C isotope labelled internal standard for the determination of the mycotoxin deoxynivalenol by LC-MS/MS without clean-up", Anal. Bioanal. Chem. 384 (3), (2006), 692-696
- [3] International Organization for Standardization (ISO), (2008), "Guide to the expression of uncertainty in measurement", (GUM 1995 with minor corrections) 1st Ed. Geneva, Switzerland
- [4] E.W. Flick, (1998), "Industrial Solvents Handbook", 5th Ed., Noyes Data Corp. Westwood NJ