林缘香料

Safety Data Sheet MSDS / SDS

According to the UN GHS revision 9

Version: 1.0 Creation Date: July 15, 2019 Revision Date: July 15, 2019

SECTION 1: Identification

1.1 GHS Product identifier

Product name 2-ethylhexan-1-ol

MF:C8H18O

l³c CH³

1.2 Other means of identification

Product number

Other names 2-Ethylhexyl Alcohol; 2-Aethyl-naphthalin; Octyl Alcohol

1.3 Recommended use of the chemical and restrictions on use

Identified uses Industrial and scientific research use.

Uses advised against no data available

1.4 Supplier's details

Company Jiangxi LinQ Spices Co.,Ltd.

Address Building15#,Xinghai Gardon,TianLi Square,QingYuan District,

Ji'An City, Jiang Xi Province

Telephone (+86)0796-8287629

SECTION 2: Hazard identification

2.1 Classification of the substance or mixture

Skin irritation, Category 2 Eye irritation, Category 2

Acute toxicity - Category 4, Inhalation

Specific target organ toxicity – single exposure, Category 3

2.2 GHS label elements, including precautionary statements

Pictogram(s)



Signal word

Warning

Hazard statement(s)

H315 Causes skin irritation H319 Causes serious eye irritation

H332 Harmful if inhaled

H335 May cause respiratory irritation

Precautionary statement(s)

Prevention

P264 Wash ... thoroughly after handling.

P280 Wear protective gloves/protective clothing/eye protection/face

protection/hearing protection/...

P261 Avoid breathing dust/fume/gas/mist/vapours/spray. P271 Use only outdoors or in a well-ventilated area.

Response P302+P352 IF ON SKIN: Wash with plenty of water/...

P321 Specific treatment (see ... on this label).

P332+P317 If skin irritation occurs: Get medical help.

P362+P364 Take off contaminated clothing and wash it before reuse. P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P304+P340 IF INHALED: Remove person to fresh air and keep comfortable

for breathing.

P317 Get medical help.

P319 Get medical help if you feel unwell.

Storage P403+P233 Store in a well-ventilated place. Keep container tightly closed.

P405 Store locked up.

Disposal P501 Dispose of contents/container to an appropriate treatment and disposal

facility in accordance with applicable laws and regulations, and product

characteristics at time of disposal.

2.3 Other hazards which do not result in classification

no data available

SECTION 3: Composition/information on ingredients

3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
2-ethylhexan-1-ol	2-ethylhexan-1-ol	104-76-7	203-234-3	100%

SECTION 4: First-aid measures

4.1 Description of necessary first-aid measures

If inhaled

Fresh air, rest. Refer for medical attention.

Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap.

Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

Following ingestion

Rinse mouth. Refer for medical attention.

4.2 Most important symptoms/effects, acute and delayed

Anesthesia, nausea, headache, dizziness; mildly irritating to skin and eyes. (USCG, 1999)

4.3 Indication of immediate medical attention and special treatment needed, if necessary

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand-valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR as necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. Higher alcohols (>3 carbons) and related compounds

SECTION 5: Fire-fighting measures

5.1 Suitable extinguishing media

Suitable extinguishing media: Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.

5.2 Specific hazards arising from the chemical

This chemical is combustible. (NTP, 1992)

5.3 Special protective actions for fire-fighters

Use water spray, alcohol-resistant foam, dry powder, carbon dioxide.

SECTION 6: Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

6.2 Environmental precautions

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

6.3 Methods and materials for containment and cleaning up

Use personal protective equipment. Avoid breathing vapors, mist or gas. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Beware of vapors accumulating to form explosive concentrations. Vapors can accumulate in low areas.

SECTION 7: Handling and storage

7.1 **Precautions for safe handling**

NO open flames. Above 73°C use a closed system, ventilation and explosion-proof electrical equipment. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

7.2 Conditions for safe storage, including any incompatibilities

Separated from strong oxidants and strong bases. Store in an area without drain or sewer access. Ventilation along the floor. Keep container tightly closed in a dry and well-ventilated place. Containers which are opened must be carefully resealed and kept upright to prevent leakage.

SECTION 8: Exposure controls/personal protection

8.1 **Control parameters**

Occupational Exposure limit values

EU-OEL: 5.4 mg/m3, 1 ppm as TWA.MAK: 54 mg/m3, 10 ppm; peak limitation category: I(1); pregnancy risk group: C

Biological limit values

no data available

8.2 **Appropriate engineering controls**

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear safety spectacles or eye protection in combination with breathing protection.

Skin protection

Protective gloves.

Respiratory protection

Use ventilation, local exhaust or breathing protection.

Thermal hazards

no data available

SECTION 9: Physical and chemical properties and safety characteristics

Physical state Liquid. Colour Colourless.

Mild, oily, sweet, slightly floral odor reminiscent of rose Odour

186 °C. Atm. press.:101.3 kPa.

75 °C. Atm. press.:101.3 kPa.

280 °C. Atm. press.:1 017 hPa.

-89 °C. Melting point/freezing point

Boiling point or initial boiling

point and boiling range

Combustible.

volume

Flammability

Lower and upper explosion limit/flammability limit

Flash point

Auto-ignition temperature **Decomposition temperature**

no data available

no data available

dynamic viscosity (in mPa s) = 9.7. Temperature:20°C.;dynamic viscosity (in Kinematic viscosity

mPa s) = 4.3. Temperature: $40^{\circ}C$.

less than 1 mg/mL at 64° F (NTP, 1992) Solubility

Partition coefficient n-

octanol/water

log Pow = 2.9. Temperature:25 °C. Remarks:Water grade 1 with 0.01 mS/m conductivity, similiar to pH7, measured was pH 5.9, but weak reliability due

Lower flammable limit: 0.88% by volume; Upper flammable limit: 9.7% by

to the low ion strength of the water.

380 Pa. Temperature:50 °C.;120 Pa. Temperature:25 °C.;93 Pa. Vapour pressure

Temperature: 20 °C.

Density and/or relative density 0.832. Temperature:20 °C.

Relative vapour density 4.49 (vs air) **Particle characteristics** no data available

SECTION 10: Stability and reactivity

10.1 Reactivity

Reacts violently with strong oxidants and strong bases. Decomposes on heating and on burning.

10.2 Chemical stability

no data available

10.3 Possibility of hazardous reactions

If material on fire or involved in fire: Do not extinguish fire unless flow can be stopped. Use water in flooding quantities as fog. Solid streams of water may be ineffective. Cool all affected containers with flooding quantities of water. Apply water from as far a distance as possible. Use foam, dry chemical, or carbon dioxide.2-ETHYL HEXANOL is an alcohol. Flammable and/or toxic gases are generated by the combination of alcohols with alkali metals, nitrides, and strong reducing agents. They react with oxoacids and carboxylic acids to form esters plus water. Oxidizing agents convert them to aldehydes or ketones. Alcohols exhibit both weak acid and weak base behavior. They may initiate the polymerization of isocyanates and epoxides. This compound is incompatible with strong oxidizing agents and strong acids. (NTP, 1992)

10.4 Conditions to avoid

no data available

10.5 Incompatible materials

A dangerous fire hazard when exposed to heat or flame; can react vigorously with oxidizing materials.

10.6 Hazardous decomposition products

When heated to decomp it emits acrid smoke and fumes.

SECTION 11: Toxicological information

Acute toxicity

• Oral: LD50 - rat (male) - ca. 2 047 mg/kg bw.

• Inhalation: LC50 - rat (male/female) - > 0.89 - <= 5.3 mg/L air.

• Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

no data available

Reproductive toxicity

no data available

STOT-single exposure

The substance is irritating to the eyes, skin and respiratory tract. The substance may cause effects on the central nervous system. If swallowed the substance may cause vomiting and could result in aspiration pneumonitis.

STOT-repeated exposure

The substance defats the skin, which may cause dryness or cracking.

Aspiration hazard

A harmful contamination of the air can be reached rather quickly on evaporation of this substance at 20°C , on spraying much faster.

SECTION 12: Ecological information

12.1 Toxicity

- Toxicity to fish: LC50 Leuciscus idus melanotus 17.1 mg/L 96 h.
- Toxicity to daphnia and other aquatic invertebrates: EC50 Daphnia magna 39 mg/L 48 h.
- Toxicity to algae: EC50 Desmodesmus subspicatus (previous name: Scenedesmus subspicatus) 11.5 mg/L
 72 h.
- Toxicity to microorganisms: NOEC activated sludge, domestic -> 300 mg/L 24 h.

12.2 Persistence and degradability

AEROBIC: 2-Ethylhexanol, present at 100 mg/L, reached 89.5% of its theoretical BOD in 2 weeks using an activated sludge inoculum at 30 mg/L in the Japanese MITI test(1). In BOD tests, 100% degradation occurred in 24 hours using an industrial wastewater (acclimated) inoculum, while no degradation occurred after 24 hours and 100% degradation occurred after 135 hours using a municipal wastewater (unacclimated) seed(2). In a BOD test using a sewage seed, 26%, 75%, 78%, and 86% of 2-ethylhexanol's theoretical BOD was exerted after 5, 10, 15, and 20 days, respectively(3). When the test was repeated in synthetic seawater, the respective results were 58%, 64%, 84%, and 100% of the theoretical BOD consumed(3). After 5 days, 41.5% of theoretical BOD was consumed using an inoculum of mixed microbial cultures(4). 2-Ethylhexanol was degraded 31% when incubated with soil from a waste disposal site for 21 days; 52% degradation occurred when the sample was amended with a source of nitrogen(5). When 2-ethylhexanol was added to river water at 18-19 deg C, it started to degrade after a 3-4 day lag, after 8 and 9 days the theoretical BOD consumed was 30% and 59%, respectively(6).

12.3 Bioaccumulative potential

An estimated BCF of 30 was calculated in fish for 2-ethylhexanol(SRC), using an estimated log Kow of 2.73(1) and a regression-derived equation(1). According to a classification scheme(2), this BCF suggests the potential for bioconcentration in aquatic organisms is low(SRC).

12.4 Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc of 2-ethylhexanol can be estimated to be 35(SRC). According to a classification scheme(2), this estimated Koc value suggests that 2-ethylhexanol is expected to have very high mobility in soil.

12.5 Other adverse effects

no data available

SECTION 13: Disposal considerations

13.1 Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incincration with flue gas scrubbing is possible for combustible packaging materials.

SECTION 14: Transport information

14.1 UN Number

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

IATA: Not dangerous goods. (For reference only, please check.)

14.2 UN Proper Shipping Name

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

IATA: Not dangerous goods. (For reference only, please check.)

14.3 Transport hazard class(es)

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

IATA: Not dangerous goods. (For reference only, please check.)

14.4 Packing group, if applicable

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

IATA: Not dangerous goods. (For reference only, please check.)

14.5 Environmental hazards

ADR/RID: No IMDG: No IATA: No

14.6 Special precautions for user

no data available

14.7 Transport in bulk according to IMO instruments

no data available

SECTION 15: Regulatory information

15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
2-ethylhexan-1-ol	2-ethylhexan-1-ol	104-76-7	203-234-3
European Inventory of I	Listed.		
EC Inventory	Listed.		
United States Toxic Sub	Listed.		
China Catalog of Hazar	Not Listed.		
New Zealand Inventory	Listed.		
Philippines Inventory of	Listed.		
Vietnam National Chem	Listed.		
Chinese Chemical Inven	Listed.		
Korea Existing Chemica	Listed.		

SECTION 16: Other information

Information on revision

Creation Date July 15, 2019 **Revision Date** July 15, 2019

Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

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- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en
- CÂMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemlDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/

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