

## Facile Synthesis of Various 1-Azabicyclo[n.4.0]alkanes via Beckmann Rearrangement/Allylsilane Cyclization

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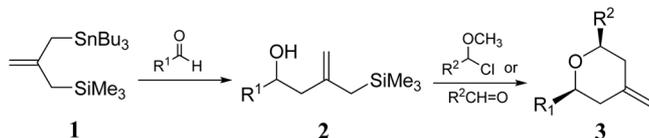
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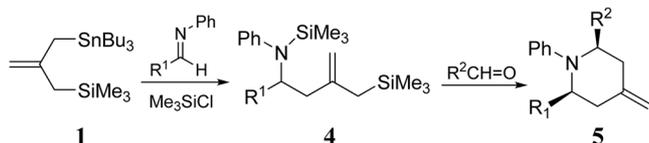
**Key Words :** Allylsilane, Beckmann rearrangement, 3-Stannyl-2-(silylmethyl)propene, *N*-Heterocycle

The carbon-carbon bond formation by the reaction of allylsilanes with electrophiles has been widely used in organic synthesis.<sup>1</sup> Particularly, intramolecular cyclization of allylsilanes bearing an electrophilic terminus has an extensive application for the highly regio- and stereo-selective synthesis of various ring compounds.

The bismetallic reagent 3-stannyl-2-(silylmethyl)propene **1**<sup>2</sup> should be a versatile conjunctive reagent since the allylstannane and the allylsilane moieties of **1** could be manipulated sequentially and in a controlled manner.<sup>3</sup> Indeed, the allylstannane moiety of **1** selectively react with an aldehyde to yield hydroxy allylsilane **2**.<sup>4</sup> The reactions of **2** with either vinyl ethers or  $\alpha$ -halo ethers give acetals which are subsequently cyclized to afford 2,6-*cis*-disubstituted-4-methylenetetrahydropyrans **3**.<sup>5</sup>

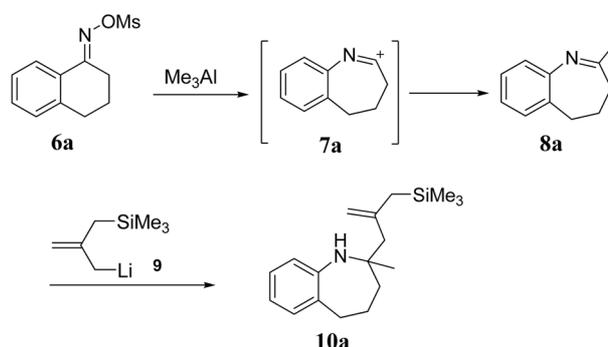


Enantioselective synthesis of the tetrahydropyrans **3** was achieved by using the hydroxy allylsilanes **2** generated from the catalytic asymmetric allylation of **1** with aldehydes.<sup>6</sup> This annulation reaction enabled an efficient synthesis of the biologically active tetrahydropyran natural products.<sup>7</sup> Various 2,6-disubstituted 4-methylenepiperidines were also prepared in one-pot by the sequential reactions of aldimines with bismetallic reagent **1** followed by aldehydes.<sup>8</sup>



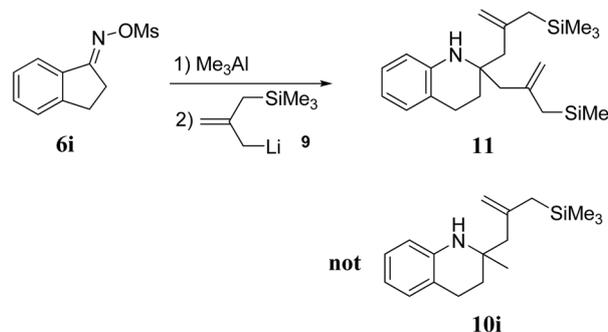
We described herein the synthesis of 1-azabicyclo[n.4.0]-alkanes using bismetallic reagent **1**. Reaction of  $\alpha$ -tetralone oxime mesylate **6a** with 2 equivalents of trimethylaluminum resulted in the formation of cyclic ketimine **8a**. Methylation of intermediate iminocarocation **7a**, which was generated from the organoaluminum-promoted Beckmann rearrangement<sup>9</sup> of oxime mesylate **6a**, with trimethylaluminum afforded cyclic imine **8a**. Allylation of cyclic ketimine **8a** with allyllithium **9** gave cyclic amino allylsilane **10a** in good

yield.<sup>10</sup> 2-(Trimethylsilylmethyl)allyllithium **9** was generated by treating bismetallic reagent **1** with methylolithium.



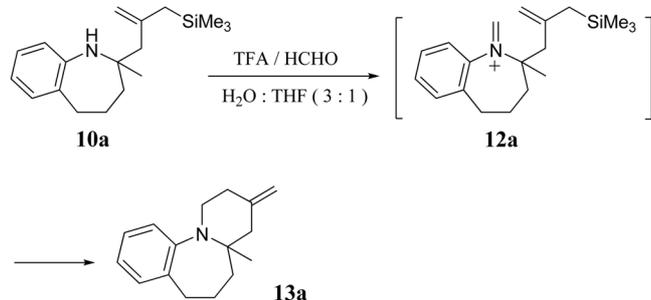
This easy and one-pot reaction has a wide generality. Other synthetic examples of this type are given in Table 1. It permits the introduction of an allylsilane moiety into a substrate with a simultaneous ring expansion.

In the synthesis of ketimines **10a** and **10b**, rigorous regioselectivities were observed. The phenyl group anti to departing mesylate group migrated preferentially.<sup>9</sup> For some reasons, this process did not work for all the oxime mesylates tested. For example, reaction of 1-indanone oxime mesylate **6i** with trimethylaluminum and followed allyllithium **9** under standard reaction condition gave only diallylation product **11** in low yield.<sup>11</sup> Even with larger excess of trimethylaluminum (4 equiv) and after prolonged reaction time, the expected monoallylation product **10i** was not produced. It is not clear why such anomalous behavior was observed for 1-indanone oxime mesylate only.



Mannich cyclization of iminium-vinyl and allylsilanes is to provide an attractive method for the regio-controlled production of piperidines possessing either endo- or exo-

cyclic unsaturation.<sup>12</sup> Cyclic amino allylsilanes **10** as their trifluoroacetate salts were treated at 40–45 °C with 1.2 equiv. of formaldehyde in water:tetrahydrofuran (3:1) to give azabicyclic compounds **13**.<sup>13</sup>



**Table 1.** Synthesis of cyclic amino allylsilanes **10** and 1-azabicyclo[n.4.0]alkanes **13**

Entry	Oxime mesylate <b>6</b>	Amino allylsilane <b>10</b> /Yield (%)	Azabicyclo <b>13</b> /Yield (%)
1			
2			
3			
4			
5			
6			
7			

As shown be seen in Table 1, 1-azabicyclo[n.4.0]alkanes of various ring size ( $n = 4, 5, 6, 7$  and 11) were obtained in good yields.

The present reaction sequence, organoaluminum-promoted Beckmann rearrangement of oxime mesylate, allylation reaction with 2-(trimethylsilylmethyl)allyllithium, and Mannich reaction, provides a versatile and useful synthetic method for 1-azabicyclo[n.4.0]alkanes.

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## References and Notes

- (a) Fleming, I.; Dunogues, J.; Smithers, R. *Org. React.* **1989**, *37*, 57. (b) Langkopf, E.; Schinzer, D. *Chem. Rev.* **1995**, *95*, 1375. (c) Chabaud, L.; James, P.; Landais, Y. *Eur. J. Org. Chem.* **2004**, 3173.
- (a) Kang, K.-T.; U, J. S.; Park, D. K.; Kim, J. G.; Kim, W. J. *Bull. Korean Chem. Soc.* **1995**, *16*, 464. (b) Benoit, D.; Bemand, L. *Synlett* **2006**, 2148.
- (a) Clive, D. L. J.; Paul, C. C.; Wang, Z. *J. Org. Chem.* **1997**, *62*, 7028. (b) Kang, K.-T.; Hwang, S. S.; Kwak, W. Y.; Yoon, U. C. *Bull. Korean Chem. Soc.* **1999**, *20*, 801.
- (a) Majetich, G.; Mishidie, H.; Zhang, Y. *J. Chem. Soc. Perkin 1* **1995**, 453. (b) Kang, K.-T.; U, J. S.; Park, D. K.; Kim, J. G.; Kwon, Y. M. *Synth. Commun.* **1997**, *27*, 1173. (c) Takuwea, A.; Saito, H.; Nishigaichi, Y. *Chem. Commun.* **1999**, 1963.
- Sung, T. M.; Kwak, W. Y.; Kang, K.-T. *Bull. Korean Chem. Soc.* **1998**, *19*, 862.
- (a) Yu, C.-M.; Lee, J.-Y.; So, B.; Hong, J. *Angew. Chem. Int. Ed.* **2002**, *41*, 161. (b) Keck, G. E.; Covell, J. A.; Schiff, T.; Yu, T. *Org. Lett.* **2002**, *4*, 1189.
- (a) Keck, G. E.; Truong, A. P. *Org. Lett.* **2005**, *7*, 2153. (b) Sanchez, C. C.; Keck, G. E. *Org. Lett.* **2005**, *7*, 3053.
- Kang, K.-T.; Kim, E. H.; Kim, W. J.; Song, N. S.; Shin, J. K.; Cho, B. Y. *Synlett* **1988**, 921.
- (a) Maruoka, K.; Miyazaki, T.; Audo, M.; Matsumara, Y.; Sakane, S.; Hattori, K.; Yamamoto, H. *J. Am. Chem. Soc.* **1983**, *105*, 2831. (b) Schinzer, D.; Bo, Y. *Angew. Chem. Int. Ed. Engl.* **1991**, *30*, 687. (c) Schinzer, D.; Langkopt, E. *Synlett* **1994**, 375.
- 10a**: <sup>1</sup>H NMR δ 0.04 (9H, s), 1.06 (3H, s), 1.62–1.72 (4H, m), 1.63 (1H, d,  $J = 13.2$  Hz), 1.74 (1H, d,  $J = 13.2$  Hz), 2.08 (1H, d,  $J = 13.0$  Hz), 2.23 (1H, d, 13.0 Hz), 2.74 (2H, t,  $J = 4.8$  Hz), 3.74 (1H, brs), 4.74 (1H, s), 4.81 (1H, s), 6.67–6.98 (4H, m); <sup>13</sup>C NMR δ 1.5, 22.5, 25.5, 29.5, 35.5, 42.7, 50.6, 54.9, 111.9, 121.0, 121.2, 126.4, 129.9, 134.0, 144.3, 146.2; HRMS  $m/z$  287.2080 (C<sub>18</sub>H<sub>29</sub>NSi requires 287.2071).
- 11**: <sup>1</sup>H NMR δ 0.00 (18H, s), 1.61 (4H, d,  $J = 13.2$  Hz), 1.71 (4H, d,  $J = 13.2$  Hz), 1.83 (2H, t,  $J = 6.8$  Hz), 2.10 (4H, d,  $J = 13.4$  Hz), 2.26 (4H, d,  $J = 13.4$  Hz), 2.78 (2H, t,  $J = 9.8$  Hz), 3.90 (1H, brs), 4.59 (2H, s), 4.71 (2H, s), 6.46–7.06 (4H, m); <sup>13</sup>C NMR δ -1.3, 23.7, 29.4, 30.4, 46.7, 54.0, 111.4, 114.7, 116.7, 120.4, 126.7, 129.2, 143.9, 144.0; HRMS  $m/z$  385.2619 (C<sub>23</sub>H<sub>39</sub>NSi<sub>2</sub> requires 385.2623).
- Grieco, P. A.; Fobare, W. F. *Tetrahedron Lett.* **1986**, *27*, 5067.
- 13a**: <sup>1</sup>H NMR δ 0.97 (3H, s), 1.26–1.42 (2H, m), 1.52–1.72 (1H, m), 1.80–1.99 (2H, m), 2.28–2.45 (2H, m), 2.48–2.74 (2H, m), 2.92–3.50 (3H, m), 4.75 (1H, s), 4.88 (1H, s), 6.88–7.28 (4H, m); <sup>13</sup>C NMR δ 19.2, 19.5, 30.3, 35.5, 37.7, 44.9, 46.8, 56.1, 109.6, 121.4, 122.5, 126.6, 128.2, 137.0, 145.6, 148.8; HRMS  $m/z$  227.1681 (C<sub>16</sub>H<sub>21</sub>N requires 227.1675).