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Scheme 1.

A New Route to N-Substituted Aminomethylphosphonates

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Some of N-substituted aminomethylphosphonic acids and their esters are known to have herbicidal and fire-proofing applications. Synthesis of their acids has been developed in a wide variety of methods. However, synthetic methology of their esters is relatively rare. Only two methods were reported for the preparation of their esters: (1) substitution of chloromethylphosphonates or phosphonomethyltriflates with primary amines (2) rearrangement of P-(halomethyl)-N-phenylphosphonoamidates. These are not generally applicable for diverse N-substituted aminomethylphosphonates.

We recently described that condensation of N-substituted N-methoxymethylamines (1) with trialkylphosphites affords N-substituted aminomethylphosphonates (3.4). The reaction of N-substituted amine with paraformaldehyde and sodium methoxide in methanol yields either N-substituted N-methoxymethylamine (1) or 1,3,5-trisubstituted-hexahydro-1,3,5-triazine (2) depending on the substituent R¹. The heating of 1 under reduced pressure results in a quantitative transformation into the corresponding 2. In this communication we present a new route to the synthesis of 3 and 4 involving the reaction of 2 with tialkylphosphites in the pre-

sence of TiCl₄.9 (Scheme 1)

Typical reaction procedure is as follows. To a stirred solution of 1,3,5-trisubstituted-hexahydro-1,3,5-triazine (2) (6 mmol) in 100 ml of CH₂Cl₂ under nitrogen was slowly added TiCl₄ (21 mmol) at 0 °C. After being stirred for 10 min. trialkylphosphite (18 mmol) was added. The resulting solution was stirred at 0 °C for 1-2 hr. After being observed that all starting material was consumed on TLC the reaction mixture was poured into water. The resulting solution was neutralized with sat. NaHCO₃ solution. The reaction product was extracted with CH₂Cl₂. Organic layer was washed successively with water and brine, dried over anhydrous MgSO₄, filtered, and concentrated under reduced pressure. The reaction product was further purified by column chromatography or short-path distillation. Results are summarized in the Table 1.

This methods is widely applicable to the various N-substituted aminomethylphosphonates including simple alkyl (entry 1,2,3), bulky t-butyl (entry 4), allyl (entry 5), cyclohexyl (entry 6,7), phenyl(entry 8), benzyl (entry 9) as substituents. Furthermore, a commercial herbicide N-phosphonomethylglycine 10 is also accessed by this method 9 as its corresponding trialkylesters (entry 10,11). 11,12

The mechanism of this reaction is not clear. But, α -methyleneamine peaks in ^1H and ^{13}C NMR were observed by mixing 1,3,5-trisubstituted-hexahydro-1,3,5-triazine and TiCl_4 . The observed peaks were close to those of free N-substituted methyleneamine 14 generated by flash vacuum thermolysis of the corresponding N-substituted aminoacetonitrile. Moreover, the color of this mixture became deep wine-red, which indicated that formation of α -methyleneamine complex with TiCl_4 . On the basis of the observations we predict that this reaction is proceeded by

Table 1. N-Substituted Aminomethylphosphonates Prepared from the Reaction of 1,3,5-Trisubstitutedhexahydro-1,3,5-Triazines with Trialkylphosphites in the Presence of Titanium Tetrachloride

	R^2	Yield ^{a,b} (%)	¹ H NMR(CDCl ₃) ^c	$IR (cm^{-1}, P = 0)$
	CoHe	49	3.16(d, 10 Hz)	1256
0 5			3.17(d, 12 Hz)	1249
3. 2.2 2			3.00(d, 12 Hz)	1239
·				1239
. •	-			1255
2 -				1255
* "	-		3.10(d, 10 Hz)	1253
, , , , ,	=		3.54(dd, 12 Hz, 6 Hz)	1212
			3.15(d, 10 Hz)	1257
2			3.17(d, 12 Hz)	1239
	-		3.05(d, 12 Hz)	1238
	R ¹ CH ₃ CH ₂ CH ₃ (CH ₂) ₂ CH ₂ CH ₃ (CH ₂) ₂ CH ₂ (CH ₃)C CH ₂ CHCH ₂ Cyclo-C ₆ H ₁₁ Cyclo-C ₆ H ₁₁ Ph PhCH ₂ C ₂ H ₅ O ₂ CCH ₂	CH ₃ CH ₂ C ₂ H ₅ CH ₃ (CH ₂) ₂ CH ₂ CH ₃ CH ₃ (CH ₂) ₂ CH ₂ C ₂ H ₅ (CH ₃)C C ₂ H ₅ CH ₂ CHCH ₂ C ₂ H ₅ Cyclo-C ₆ H ₁₁ CH ₃ Cyclo-C ₆ H ₁₁ C ₂ H ₅ Ph C ₂ H ₅ Cycl ₅ C ₂ H ₅ Cyclo-C ₆ H ₁₁ C ₂ H ₅ CH ₅ CH ₅ C ₂ H ₅ CH ₃ CH ₃	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

^aYield of isolated pure product, not optimized except entry 10. ^bSatisfactory microanalyses obtained. ^cChemical shift of two methylene protons between nitrogen and phosphorus, referenced by TMS.

nucleophilic addition of phosphites to α -position of N-substituted methyleneamine (5) or its equivalent (6,7) generated from hexahydro-1,3,5-triazine with TiCl₄.

Along with the earlier result 6 N-substituted aminomethyl-phosphonates can be generally synthesized by the reaction of N-methoxymethylamines or hexahydro-1,3,5-triazines with trialkylphosphites in the presence of $TiCl_4$.

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