

## Facile Synthesis of 1,2,3,4-Tetrasubstituted Pyrroles from Baylis-Hillman Adducts

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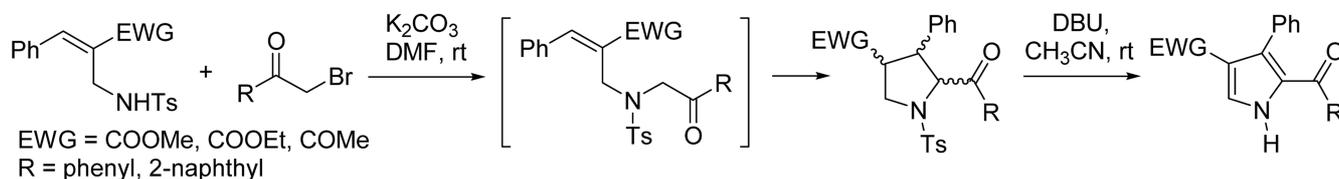
Suitably functionalized pyrroles are the basic skeleton of many biologically important substances and numerous synthetic methods of pyrroles have been investigated extensively.<sup>1,2</sup> However, the synthesis of pyrrole derivatives from Baylis-Hillman adducts was not developed much.<sup>2</sup> Recently, we reported the synthesis of 2,3,4-trisubstituted pyrroles starting from the rearranged *aza*-Baylis-Hillman adducts (Scheme 1).<sup>3</sup>

Meantime we presumed that we could synthesize 1,2,3,4-tetrasubstituted pyrrole derivatives by using the synthetic approach in Scheme 2. As shown in Scheme 2, we imagined that the reaction of Baylis-Hillman acetate **1**, as the representative example, and secondary amine derivatives **2a-d** could give the corresponding S<sub>N</sub>2' product **3a-d**, which could be cyclized to **4a-d** under basic conditions. The following acid-catalyzed dehydration and concomitant double bond isomerization of **4a-d** would provide desired pyrroles **5a-d**.

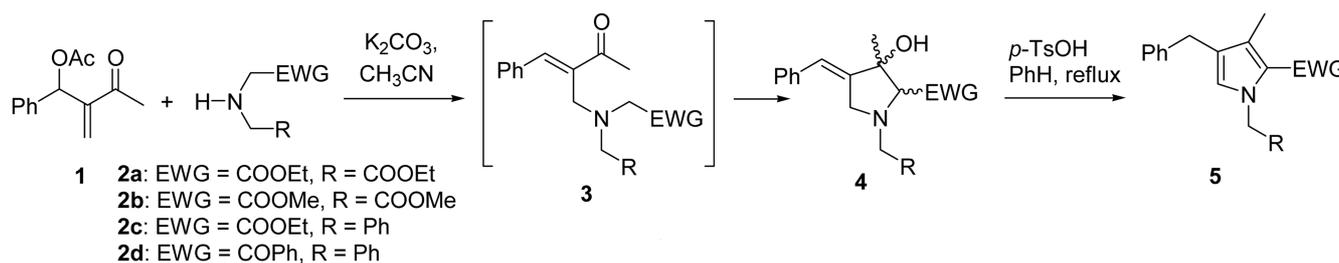
Among the examined conditions the use of K<sub>2</sub>CO<sub>3</sub> in CH<sub>3</sub>CN gave the best results for the preparation of **4a-d**. As expected we could not observe the formation of **3** (except for **3c**, entry 3 in Table 1),<sup>4</sup> instead we obtained **4a-d** directly in 50-74% yields as inseparable *syn/anti* mixtures in a one-pot reaction. Based on the <sup>1</sup>H NMR spectra of **4a-d** the ratio of *syn/anti* was 4:1 to 2:1 (footnotes b-d in Table 1), however,

we did not confirm which isomer is the major one. For the reaction of **1** and **2c** we isolated **3c** in 34% yield (entry 3 in Table 1) together with **4c** in 50% yield. For the synthesis of compound **4d** (entry 4) we used **2d**<sup>5</sup> in slightly excess amount (footnote e in Table 1). The following dehydration step of **4a-d** was carried out under the influence of *p*-TsOH (20-40 mol%) in benzene and we obtained the desired compounds **5a-d** in 41-64% yields. Isomerization of double bond occurred during the dehydration stage simultaneously to afford pyrroles directly. The results are summarized in Table 1.

However, the reaction of **1** and **2e** showed somewhat different reactivity as compared with those of **2a-d** (Scheme 3). When we carried out the reaction of **1** and **2e** in CH<sub>3</sub>CN at room temperature the reaction did not show the formation of any new compounds in appreciable amounts presumably due to the limited solubility of **2e** in CH<sub>3</sub>CN. Thus we elevated the temperature to refluxing, however, rearranged acetate was the major product in this case. After many trials we could obtain **3e** in 74% yield in aqueous CH<sub>3</sub>CN at room temperature. In aqueous CH<sub>3</sub>CN the compound **2e** was dissolved completely and the rearrangement of acetate group of **1** to the primary position was minimized at room temperature. With this compound **3e** in our hand we prepared **4e** under the same conditions of Table 1 (CH<sub>3</sub>CN,



Scheme 1

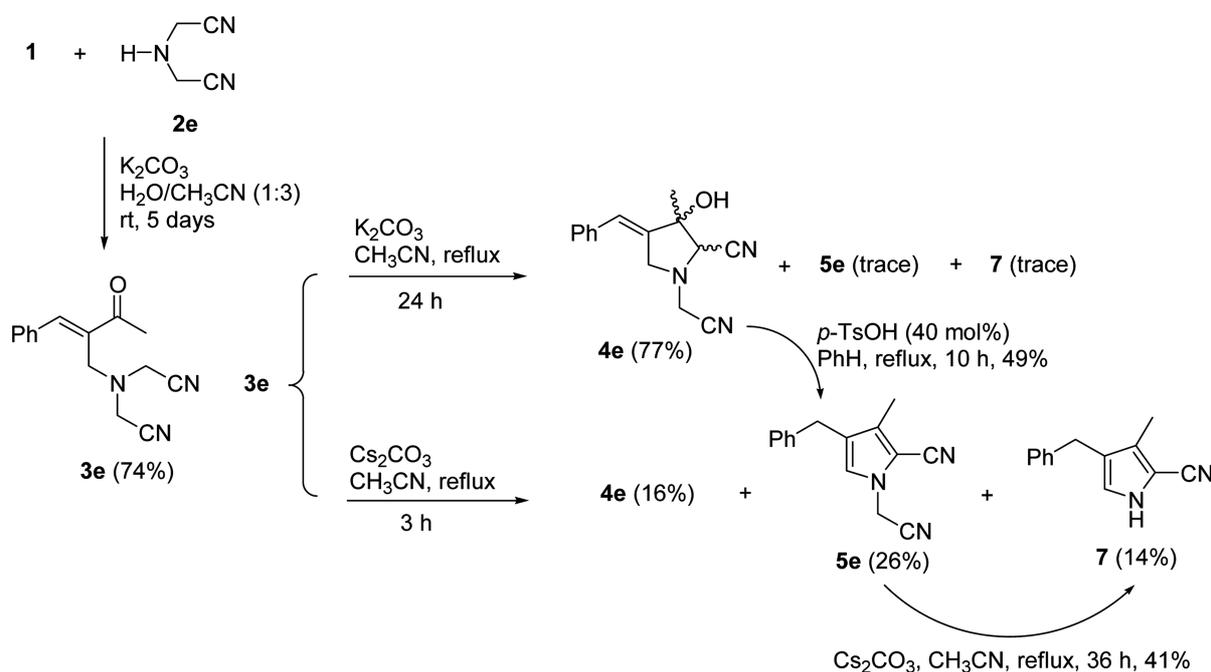


Scheme 2

**Table 1.** Synthesis of 1,2,3,4-tetrasubstituted pyrroles

Entry	1 + 2	Conditions	3 (%) / 4 (%)	Conditions	5 (%) <sup>f</sup>
1	1 + 2a	K <sub>2</sub> CO <sub>3</sub> (1.1 equiv) CH <sub>3</sub> CN, reflux, 27 h	3a (nd) <sup>a</sup> / 4a (69) <sup>b</sup>	<i>p</i> -TsOH (20 mol%) PhH, reflux, 10 h	5a (64)
2	1 + 2b	K <sub>2</sub> CO <sub>3</sub> (1.1 equiv) CH <sub>3</sub> CN, reflux, 26 h	3b (nd) <sup>a</sup> / 4b (71) <sup>c</sup>	<i>p</i> -TsOH (20 mol%) PhH, reflux, 12 h	5b (47)
3	1 + 2c	K <sub>2</sub> CO <sub>3</sub> (2.2 equiv) CH <sub>3</sub> CN, reflux, 7 days	3c (34) / 4c (50) <sup>d</sup>	<i>p</i> -TsOH (40 mol%) PhH, reflux, 2 days	5c (56)
4	1 + 2d <sup>e</sup>	K <sub>2</sub> CO <sub>3</sub> (1.1 equiv) CH <sub>3</sub> CN, rt, 1 h	3d (nd) <sup>a</sup> / 4d (74) <sup>d</sup>	<i>p</i> -TsOH (20 mol%) PhH, reflux, 12 h	5d (41)

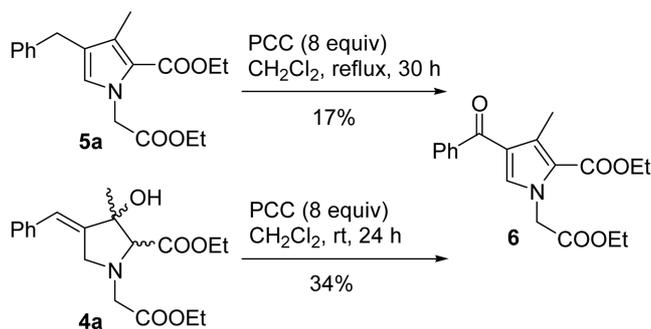
<sup>a</sup>Nd means not detected. <sup>b</sup>The ratio is 2:1 (*syn/anti* mixture). <sup>c</sup>The ratio is 4:1 (*syn/anti* mixture). <sup>d</sup>The ratio is 3:1 (*syn/anti* mixture). <sup>e</sup>Starting material **2d** was prepared by the reaction of benzylamine and phenacyl bromide according to the reference.<sup>5</sup> The compound **2d** was unstable thus we used this compound in a crude state and we used 0.91 equiv of **1**. <sup>f</sup>Isolated yield.

**Scheme 3**

K<sub>2</sub>CO<sub>3</sub>, reflux, 24 h) in 77% yield (*syn/anti*, 3:2). Dehydration of **4e** under the same conditions (*p*-TsOH/benzene/reflux) afforded **5e** in 49% yield. During the synthesis of **4e** we observed the formation of trace amounts of **5e** and **7**. It is interesting to note that the yields of **5e** and **7** were increased with concomitant decrease of **4e** when we used Cs<sub>2</sub>CO<sub>3</sub> (CH<sub>3</sub>CN, reflux, 3 h). The formation of pyrrole derivative **7** can be explained by decyanomethylation of **5e**,<sup>6</sup> and we confirmed the conversion experimentally by transforming **5e** into **7** under the same conditions (41% and recovered **5e** in 10%).

Finally, we examined the possibility for the oxidation of **5a** into 4-benzoylpyrrole derivative **6** as in our previous oxidation involving PCC (pyridinium chlorochromate) in a similar system.<sup>7</sup> However, the yield of oxidized compound **6** was very low to be useful in a synthetic point of view. It is interesting to note that the oxidation with the precursor **4a** instead of **5a** showed somewhat improved yield.

In summary, we disclosed the synthesis of poly-substituted

**Scheme 4**

pyrrole derivatives from the reaction of Baylis-Hillman acetate and some secondary amine compounds.<sup>8</sup>

## Experimental Section

**Typical experimental procedure for the synthesis of compounds 4a and 5a, and the spectroscopic data of 3c,**

**3e, 4a-e, 5a-e, 6, and 7 are as follows.** A stirred mixture of **1** (218 mg, 1.0 mmol), **2a** (189 mg, 1.0 mmol), and  $K_2CO_3$  (152 mg, 1.1 mmol) in  $CH_3CN$  (5 mL) was heated to reflux for 27 h. After the usual aqueous workup procedure and column chromatographic purification process (hexanes/EtOAc, 3:1) we obtained **4a** as colorless oil, 240 mg (69%). A solution of **4a** (174 mg, 0.5 mmol) and *p*-TsOH (19 mg, 0.1 mmol) in benzene (4 mL) was heated to reflux for 10 h. After the usual aqueous workup procedure and column chromatographic purification process (hexanes/EtOAc, 6:1) we obtained **5a** as a white solid, 105 mg (64%).

Compound **3c**: 34%; colorless oil; IR (film) 2924, 1737, 1666, 1231, 1189, 1029  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  1.19 (t,  $J = 7.2$  Hz, 3H), 2.42 (s, 3H), 3.22 (s, 2H), 3.76 (s, 2H), 3.81 (s, 2H), 4.04 (q,  $J = 7.2$  Hz, 2H), 7.19-7.27 (m, 5H), 7.32-7.42 (m, 3H), 7.55-7.58 (m, 3H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  14.14, 26.70, 49.21, 53.37, 57.85, 60.07, 127.10, 128.18, 128.37, 128.83, 129.11, 130.05, 135.11, 138.59, 139.04, 141.62, 171.18, 200.85.

Compound **3e**: 74%; colorless oil; IR (film) 2246, 1664, 1421, 1230, 1132  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  2.51 (s, 3H), 3.55 (s, 4H), 3.64 (s, 2H), 7.42-7.49 (m, 5H), 7.85 (s, 1H).

Compound **4a**: 69% (*syn/anti*, 2:1); colorless oil; IR (film) 3446, 2981, 1738, 1448, 1195, 1097  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz, major isomer)  $\delta$  1.27 (t,  $J = 7.2$  Hz, 3H), 1.31 (t,  $J = 7.2$  Hz, 3H), 1.64 (s, 3H), 2.80 (br s, 1H), 3.51-3.84 (m, 4H), 4.11-4.36 (m, 5H), 6.61 (t,  $J = 2.4$  Hz, 1H), 7.20-7.24 (m, 3H), 7.28-7.36 (m, 2H).

Compound **4b**: 71% (*syn/anti*, 4:1); colorless oil; IR (film) 3452, 2954, 1747, 1693, 1442, 1213, 1178  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz, major isomer)  $\delta$  1.63 (s, 3H), 3.51-3.90 (m, 6H), 3.70 (s, 3H), 3.77 (s, 3H), 6.61 (t,  $J = 2.4$  Hz, 1H), 7.20-7.26 (m, 3H), 7.27-7.36 (m, 2H).

Compound **4c**: 50% (*syn/anti*, 3:1); colorless oil; IR (film) 3454, 2981, 1739, 1448, 1261, 1196  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz, major isomer)  $\delta$  1.32 (t,  $J = 7.5$  Hz, 3H), 1.60 (s, 3H), 2.75 (br s, 1H), 3.34-3.65 (m, 3H), 3.94-4.05 (m, 2H), 4.21-4.31 (m, 2H), 6.56 (t,  $J = 2.4$  Hz, 1H), 7.15-7.21 (m, 3H), 7.24-7.39 (m, 2H).

Compound **4d**: 74% (*syn/anti*, 3:1); colorless oil; IR (film) 3438, 1676, 1448, 1228, 1180, 1092  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz, major isomer)  $\delta$  1.55 (s, 3H), 2.68 (br s, 1H), 3.38-4.23 (m, 4H), 4.38 (s, 1H), 6.53 (t,  $J = 2.4$  Hz, 1H), 7.17-7.34 (m, 10H), 7.43-7.49 (m, 2H), 7.54-7.60 (m, 1H), 7.93-7.97 (m, 2H).

Compound **4e**: 77% (*syn/anti*, 3:2); colorless oil; IR (film) 3429, 2925, 2222, 1448, 1261, 1101  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz, major isomer)  $\delta$  1.66 (s, 3H), 2.60 (br s, 1H), 3.69 (s, 1H), 3.80-3.97 (m, 4H), 6.70 (t,  $J = 2.4$  Hz, 1H), 7.21-7.46 (m, 5H) and  $^1H$  NMR ( $CDCl_3$ , 300 MHz, minor isomer)  $\delta$  1.71 (s, 3H), 2.54 (br s, 1H), 3.78 (s, 1H), 3.81 (s, 1H), 3.87 (s, 1H), 3.91 (d,  $J = 2.4$  Hz, 2H), 6.60 (t,  $J = 2.4$  Hz, 1H), 7.21-7.42 (m, 5H).

Compound **5a**: 64%; white solid, mp 42-44 °C; IR (film) 1755, 1687, 1417, 1298, 1199, 1097  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  1.27 (t,  $J = 7.2$  Hz, 3H), 1.32 (t,  $J = 7.2$  Hz, 3H), 2.24 (s, 3H), 3.76 (s, 2H), 4.21 (q,  $J = 7.2$  Hz, 2H), 4.25 (q,  $J = 7.2$  Hz, 2H), 4.87 (s, 2H), 6.42 (s, 1H), 7.12-7.20 (m, 3H), 7.25-7.30 (m, 2H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  11.60, 14.12, 14.34, 31.24, 51.14, 59.69, 61.30, 119.74, 122.66, 125.84, 127.65, 128.32, 128.53, 128.66, 140.81, 162.08, 169.27; LCMS  $m/z$  329 ( $M^+$ ).

Compound **5b**: 47%; colorless oil; IR (film) 1759, 1693, 1444, 1215, 1124, 1099  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  2.23 (s, 3H), 3.75 (s, 3H), 3.76 (s, 2H), 3.79 (s, 3H), 4.87 (s, 2H), 6.43 (s, 1H), 7.16-7.20 (m, 3H), 7.24-7.30 (m, 2H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  11.54, 31.22, 50.80, 50.97, 52.28, 119.55, 122.77, 125.86, 127.87, 128.33, 128.52, 128.72, 140.68, 162.54, 169.69.

Compound **5c**: 56%; colorless oil; IR (film) 1693, 1452, 1421, 1386, 1297, 1095  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  1.24 (t,  $J = 6.9$  Hz, 3H), 2.24 (s, 3H), 3.76 (s, 2H), 4.19 (q,  $J = 6.9$  Hz, 2H), 5.43 (s, 2H), 6.55 (s, 1H), 7.01-7.04 (m, 2H), 7.14-7.29 (m, 8H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  11.69, 14.28, 31.21, 52.44, 59.47, 119.60, 122.29, 125.76, 126.41, 127.04, 127.33, 128.28, 128.39, 128.43, 128.59, 138.96, 141.03, 161.86; LCMS  $m/z$  333 ( $M^+$ ).

Compound **5d**: 41%; colorless oil; IR (film) 1624, 1495, 1446, 1400, 1215, 1173  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  1.63 (s, 3H), 3.73 (s, 2H), 5.37 (s, 2H), 6.68 (s, 1H), 7.05-7.08 (m, 2H), 7.16-7.30 (m, 8H), 7.34-7.40 (m, 2H), 7.44-7.50 (m, 1H), 7.58-7.61 (m, 2H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  12.04, 31.30, 51.99, 122.72, 125.87, 126.80, 127.28, 128.16, 128.23, 128.34, 128.39 (2C), 128.45, 128.47, 129.00, 129.35, 131.59, 138.71, 140.73, 188.34; LCMS  $m/z$  365 ( $M^+$ ).

Compound **5e**: 49%; colorless oil; IR (film) 2208, 1493, 1425, 1390, 1372  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  2.11 (s, 3H), 3.73 (s, 2H), 4.82 (s, 2H), 6.58 (s, 1H), 7.13-7.16 (m, 2H), 7.19-7.33 (m, 3H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  10.32, 31.25, 35.66, 103.72, 112.38, 113.39, 124.99, 125.64, 126.42, 128.45, 128.63, 132.60, 139.28.

Compound **6**: 34%; colorless oil; IR (film) 2981, 1753, 1693, 1643, 1251, 1203  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  1.29 (t,  $J = 7.5$  Hz, 3H), 1.38 (t,  $J = 7.5$  Hz, 3H), 2.64 (s, 3H), 4.24 (q,  $J = 7.5$  Hz, 2H), 4.32 (q,  $J = 7.5$  Hz, 2H), 4.95 (s, 2H), 7.06 (s, 1H), 7.43-7.47 (m, 2H), 7.52-7.55 (m, 1H), 7.76 (m, 2H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  12.55, 14.12, 14.28, 51.79, 60.47, 61.73, 121.91, 122.65, 128.26, 129.04, 131.69, 132.49, 134.92, 140.18, 168.24 (2C), 191.45; LCMS  $m/z$  343 ( $M^+$ ).

Compound **7**: 41%; pale yellow solid, mp 95-97 °C; IR (film) 3303, 2212, 1396  $cm^{-1}$ ;  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  2.11 (s, 3H), 3.75 (s, 2H), 6.58 (d,  $J = 3.0$  Hz, 1H), 7.14-7.22 (m, 3H), 7.26-7.31 (m, 2H), 8.45 (br s, 1H);  $^{13}C$  NMR ( $CDCl_3$ , 75 MHz)  $\delta$  9.96, 31.29, 100.08, 114.45, 121.97, 123.62, 126.12, 128.43, 128.46, 130.64, 140.16; LCMS  $m/z$  196 ( $M^+$ ).

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## References and Notes

1. For the syntheses and biological activities of pyrrole derivatives, see: (a) Bellina, F.; Rossi, R. *Tetrahedron* **2006**, *62*, 7213-7256. (b) Knight, D. W.; Sharland, C. M. *Synlett* **2004**, 119-121. (c) Singh, V.; Kanojiya, S.; Batra, S. *Tetrahedron* **2006**, *62*, 10100-10110. (d) Knight, D. W.; Sharland, C. M. *Synlett* **2003**, 2258-2260. (e) Magnus, N. A.; Staszak, M. A.; Udodong, U. E.; Wepsiec, J. P. *Org. Proc. Res. Dev.* **2006**, *10*, 899-904. (f) Zen, S.; Harada, K. *Chem. Pharm. Bull.* **1982**, *30*, 366-369. (g) Chen, Q.; Wang, T.; Zhang, Y.; Wang, Q.; Ma, J. *Synth. Commun.* **2002**, *32*, 1051-1058. (h) Nicolaou, I.; Demopoulos, V. J. *J. Med. Chem.* **2003**, *46*, 417-426. (i) Gupton, J. T.; Banner, E. J.; Scharf, A. B.; Norwood, B. K.; Kanters, R. P. F.; Dominey, R. N.; Hempel, J. E.; Kharlamova, A.; Bluhn-Chertudi, I.; Hickenboth, C. R.; Little, B. A.; Sartin, M. D.; Coppock, M. B.; Krumpe, K. E.; Burnham, B. S.; Holt, H.; Du, K. X.; Keertikar, K. M.; Diebes, A.; Ghassemi, S.; Sikorski, J. A. *Tetrahedron* **2006**, *62*, 8243-8255. (j) Cadamuro, S.; Degani, I.; Dughera, S.; Fochi, R.; Gatti, A.; Piscopo, L. *J. Chem. Soc., Perkin Trans. 1* **1993**, 273-283. (k) Cohnen, E.; Dewald, R. *Synthesis* **1987**, 566-568. (l) Misra, N. C.; Panda, K.; Ila, H.; Junjappa, H. *J. Org. Chem.* **2007**, *72*, 1246-1251.
2. For the examples on the synthesis of pyrroles from Baylis-Hillman adducts, see: (a) Declerck, V.; Ribiere, P.; Martinez, J.; Lamaty, F. *J. Org. Chem.* **2004**, *69*, 8372-8381. (b) Shi, M.; Xu, Y.-M. *Eur. J. Org. Chem.* **2002**, 696-701. (c) Roy, A. K.; Pathak, R.; Yadav, G. P.; Maulik, P. R.; Batra, S. *Synthesis* **2006**, 1021-1027.
3. Lee, H. S.; Kim, J. M.; Kim, J. N. *Tetrahedron Lett.* **2007**, *48*, 4119-4122.
4. When we carried out the reaction in DMF in the presence of K<sub>2</sub>CO<sub>3</sub> at room temperature, the corresponding intermediates **3** could be isolated in moderate yields.
5. For the synthesis of compound **2d**, see: (a) Kawamoto, A.; Wills, M. *Tetrahedron: Asymmetry* **2000**, *11*, 3257-3261. (b) Guarna, A.; Bucelli, I.; Machetti, F.; Menchi, G.; Occhiato, E. G.; Scarpi, D.; Trabocchi, A. *Tetrahedron* **2002**, *58*, 9865-9870. (c) Deng, B.-L.; Demillequand, M.; Laurent, M.; Touillaux, R.; Belmans, M.; Kemps, L.; Ceresiat, M.; Marchand-Brynaert, J. *Tetrahedron* **2000**, *56*, 3209-3217.
6. For the decyanomethylation, see: (a) Katritzky, A. R.; Latif, M.; Urogdi, L. *J. Chem. Soc., Perkin Trans. 1* **1990**, 667-672. (b) Overman, L. E.; Shin, J. *J. Org. Chem.* **1991**, *56*, 5005-5007. (c) Yang, T.-K.; Hung, S.-M.; Lee, D.-S.; Hong, A.-W.; Cheng, C.-C. *Tetrahedron Lett.* **1989**, *30*, 4973-4976. (d) Padwa, A.; Chen, Y.-Y.; Dent, W.; Nimmegern, H. *J. Org. Chem.* **1985**, *50*, 4006-4014.
7. For the related PCC oxidations, see: (a) Kim, S. J.; Lee, H. S.; Kim, J. N. *Tetrahedron Lett.* **2007**, *48*, 1069-1072. (b) Dauben, W. G.; Michno, D. M. *J. Org. Chem.* **1977**, *42*, 682-685.
8. For our recent publications on the synthesis of nitrogen-containing five-membered heterocyclic compounds, see: (a) Lee, K. Y.; Lee, Y. J.; Kim, J. N. *Bull. Korean Chem. Soc.* **2007**, *28*, 143-146. (b) Kim, S. C.; Lee, K. Y.; Gowrisankar, S.; Kim, J. N. *Bull. Korean Chem. Soc.* **2006**, *27*, 1133-1139. (c) Lee, H. S.; Kim, S. J.; Kim, J. N. *Bull. Korean Chem. Soc.* **2006**, *27*, 1063-1066.